

Transition metal perovskites

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Outline:

Prospective in thermoelectrics- stability at high temperatures, possibility to enhance the thermoelectric power by means of the spin degree of freedom , depression of thermal conductivity - “disorder or fluctuations (spin,orbital)”

Introduction

- Electrical conductivity, Charge transfer vs Mot Hubbard insulator
- Thermoelectric power, electron transport, thermoelectricity

Thermoelectric phenomenology

Electrical and thermal transport , magnetism from 4 K up to 300 K(λ),800 K(χ ,**M**),1200 K (**S**, ρ)

MATERIALS

Mn^{3+}/Mn^{4+} perovskites: ferromagnetic double-exchange \Leftrightarrow degenerate carriers vs. antiferromagnetic super-exchange \Leftrightarrow orbitally polarized, insulating charge ordered electronic states.

Co^{3+}/Co^{4+} perovskites : the minute energy difference between the low-spin ground-state (filled t_{2g}) vs magnetic one (active e_g states) \Leftrightarrow the origin of “exotic” charge carriers and thermal properties

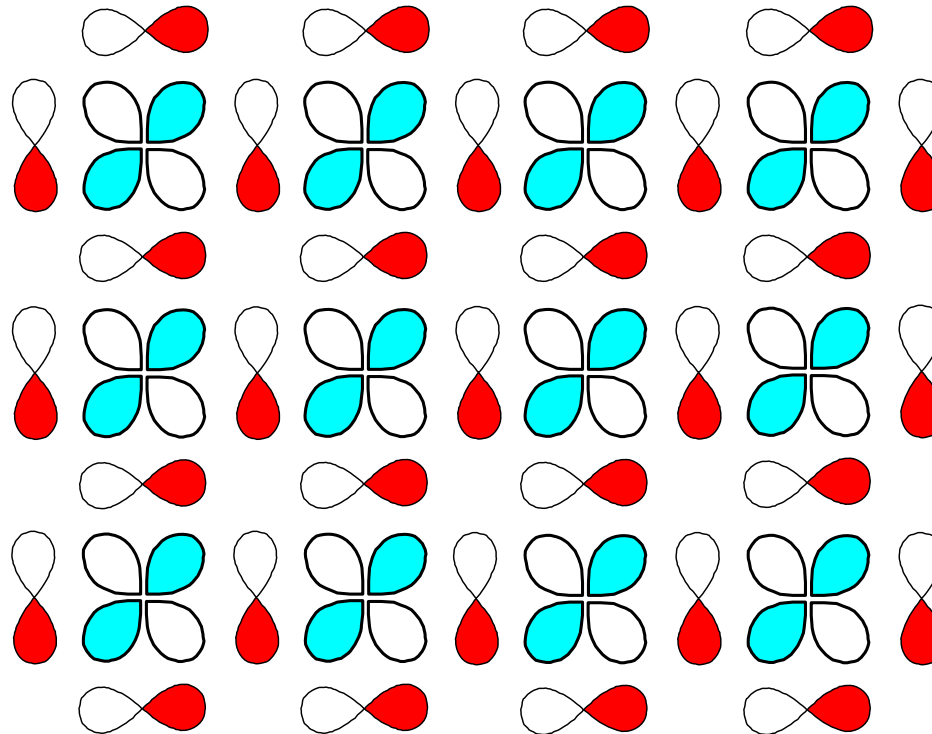
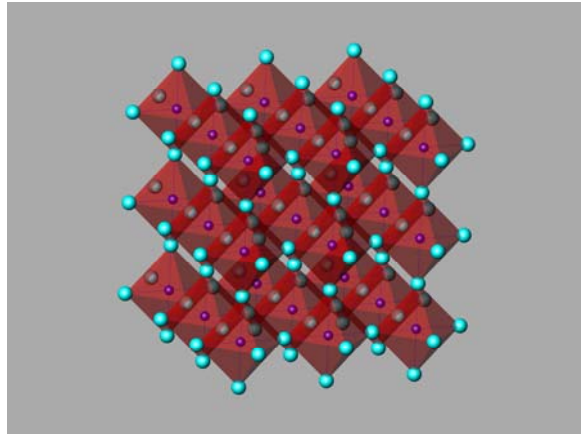
Cr^{3+}/Cr^{4+} (t_{2g}) and Fe^{3+}/Fe^{4+} (e_g) perovskites with identical concentration of both species documented; chromite represents likely a unique example of material with a pronounced role of orbital entropy in the thermopower.

Ferromagnetic $SrRuO_3$ has a high positive thermoelectric power and low thermal conductivity \Leftrightarrow a close link between the thermal and electron transport and magnetism; published results of the thermoelectric power of isoelectronic $SrFeO_3$ and $SrRuO_3$ are confronted. The role of charge compensation effect due to Na^{1+} for Sr^{2+} substitution is probed on solid solutions $Sr_{1-x}Na_xRuO_3$ ($x = 0.0 - 0.25$) is documented.

As a novelty one of the most conducting perovskite $SrMoO_3$ in connection with nitridation ($SrMoO_2N$) is also mentioned.

Key features of band structure

Orbital Overlap in the t_{2g} Band

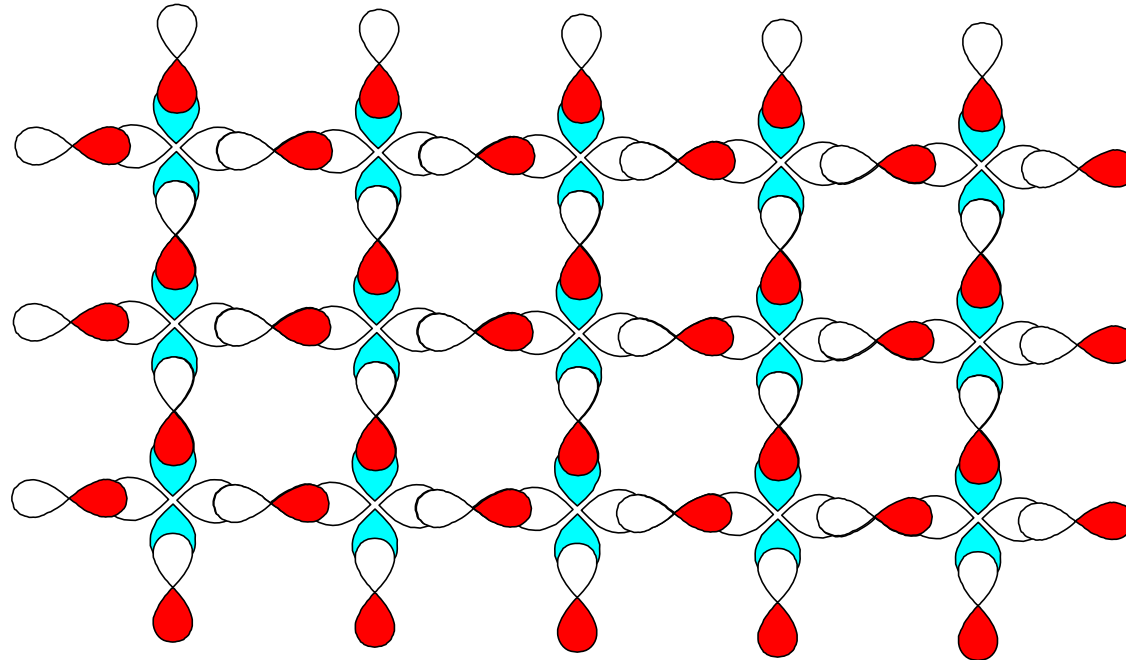
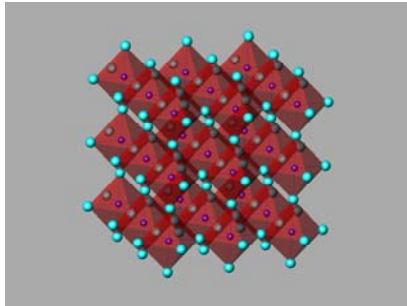


Γ point ($k_x=k_y=k_z=0$)

π bonding in MO_6 framework

Key features of band structure of perovskites

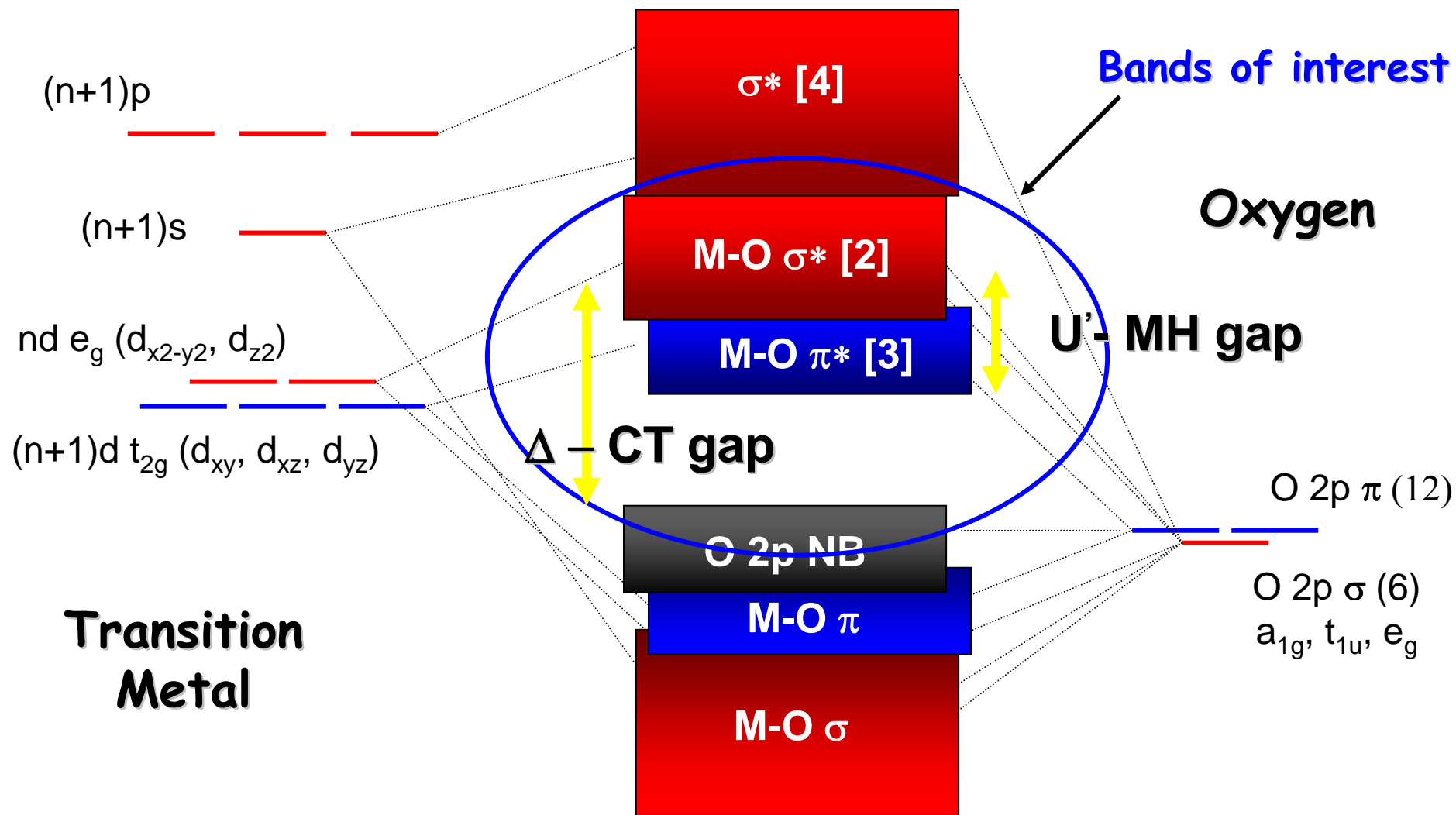
Orbital Overlap in the e_g Band



Γ point ($k_x=k_y=k_z=0$)

σ bonding in MO_6 framework

Key features of band structure of perovskites



Simple Band Structure

Key features of band structure of perovskites

- Considerations \Leftrightarrow energy is gained (lowered, system is stabilized) via hybridization between occupied and empty orbitals with same symmetry \Leftrightarrow AFM or FM interaction inferred
- **$e_g - e_g$ hybridization is stronger than $t_{2g} - t_{2g}$ hybridization because of greater overlap.**
- Simple projection in Superexchange and Double-exchange

Spin polarized Energy Diagram 6

Figure of merit – Thermoelectric parameters

$$ZT = \frac{\alpha^2 \sigma T}{\kappa}$$

$$\kappa = \kappa_e + \kappa_{ph}$$

Minimize κ_{ph} : usual strategies apply also to oxides

Wiedemann-Franz
law valid :

$$\kappa_e = \frac{\pi^2 k^2}{3e^2} \times \sigma T$$

J. Hejtmanek et. al,

L_0 Phys. Rev B, 66, (2002) 014426

Large α -values required

semiconductor doped
oxides (high T)

or

highly correlated
metallic oxides

Thermal Conductivity of Solids

•Solids transmit thermal energy by three modes

- Elastic vibrations of the lattice moving through the crystal in the form of waves
- Free electrons moving through the lattice carry energy similar to the case in gasses
- Magnetic excitations can also carry heat by a similar way as phonons

$$\lambda_{\text{total}} = \lambda_{\text{phon}} + \lambda_e + \lambda_{\text{mag}}$$

Respective thermal conductivity:

$$\lambda = \frac{1}{3} C v \ell$$

Specific heat Velocity Mean free path

Thermopower in metals

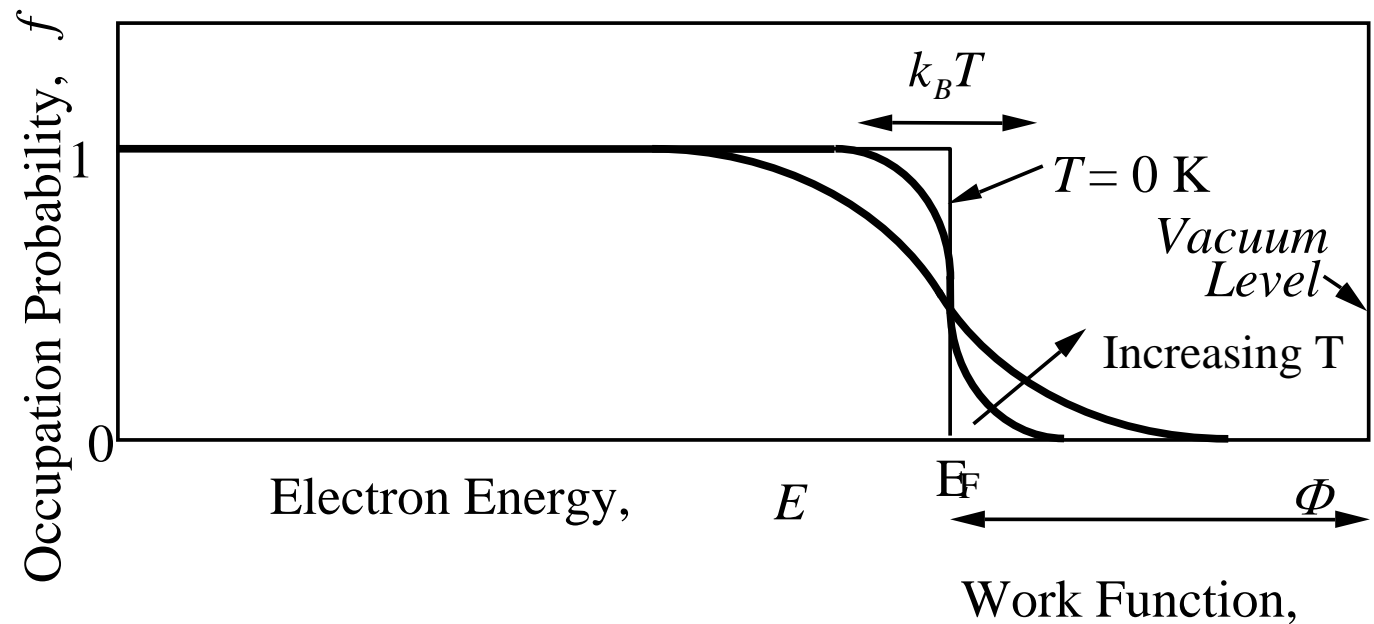
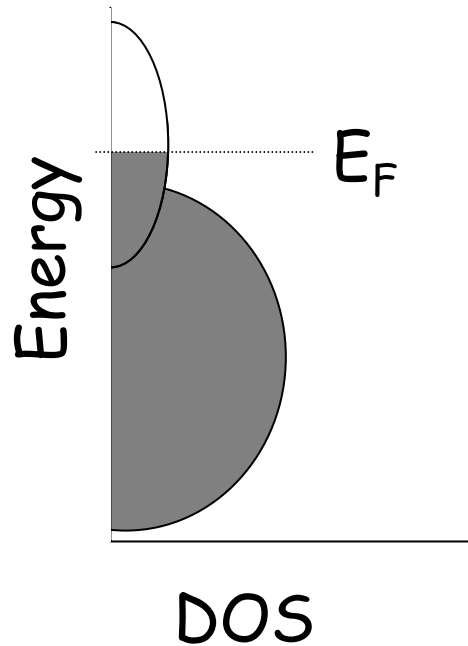
(Band Structure View Point)

Key role of Fermi-Dirac equilibrium distribution

for the probability of electron occupation of energy level E at temperature T

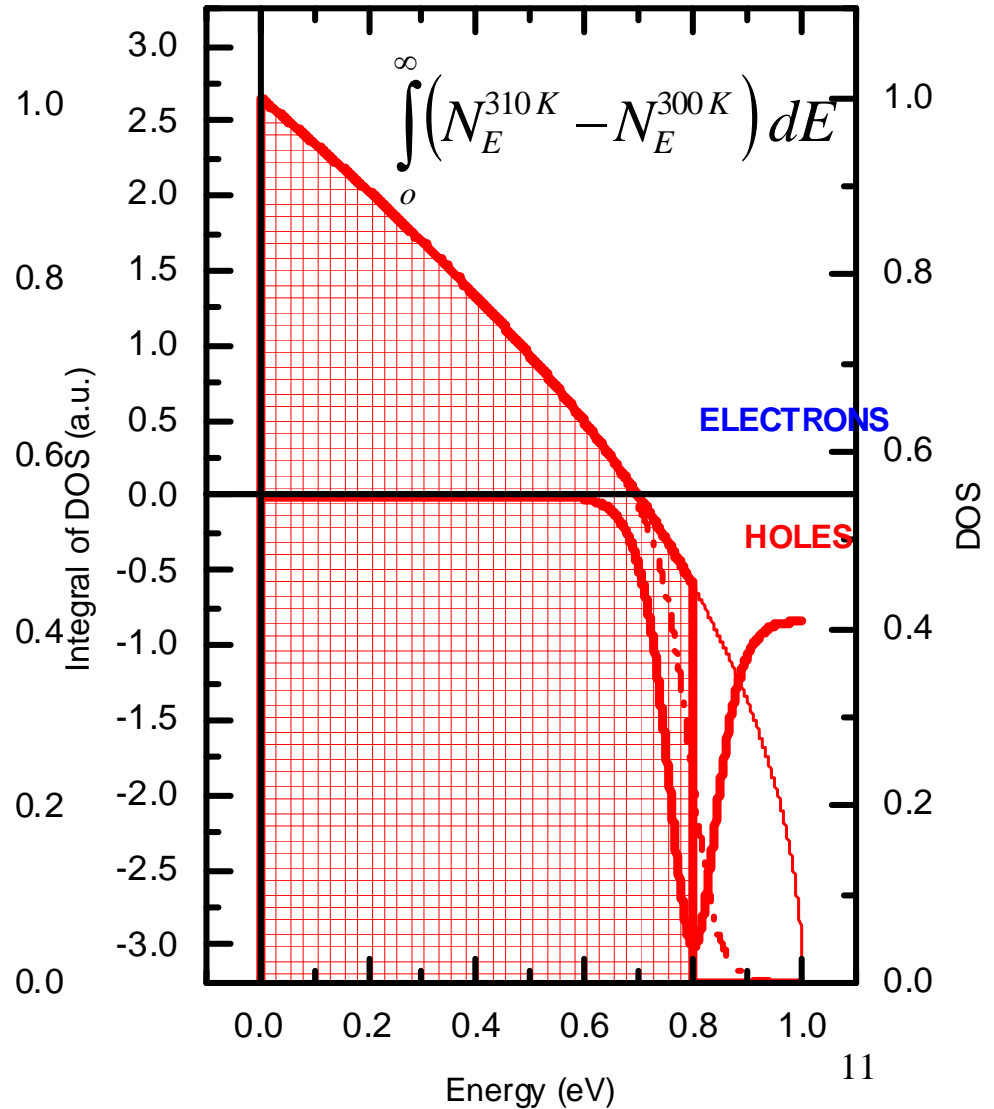
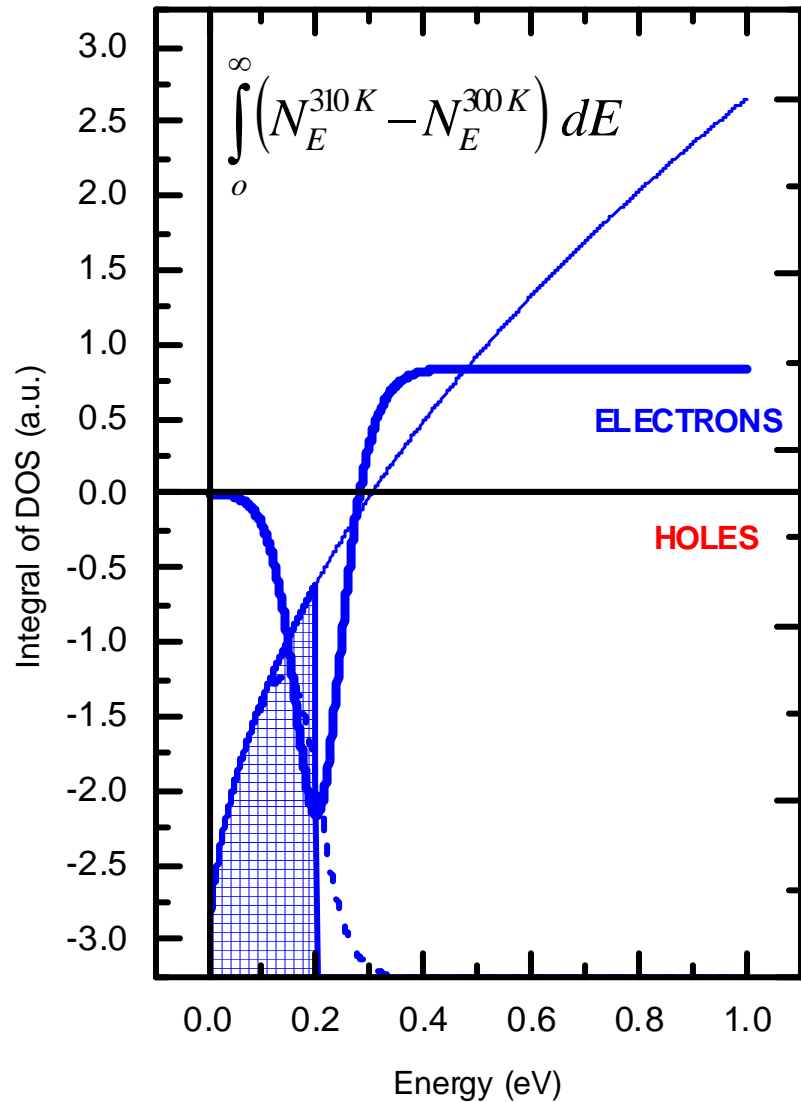
$$f(E) = \frac{1}{1 + \exp\left(\frac{E - E_F}{k_B T}\right)}$$

Effect of Temperature



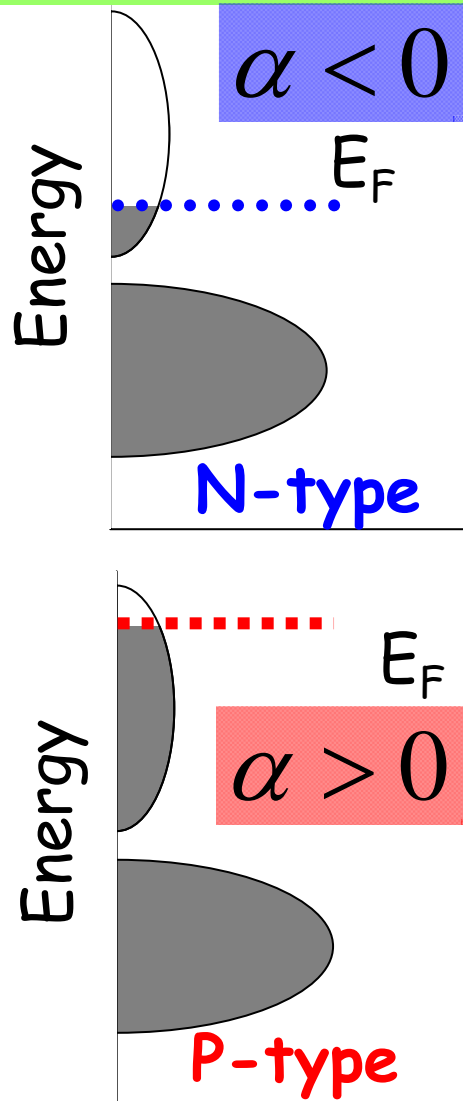
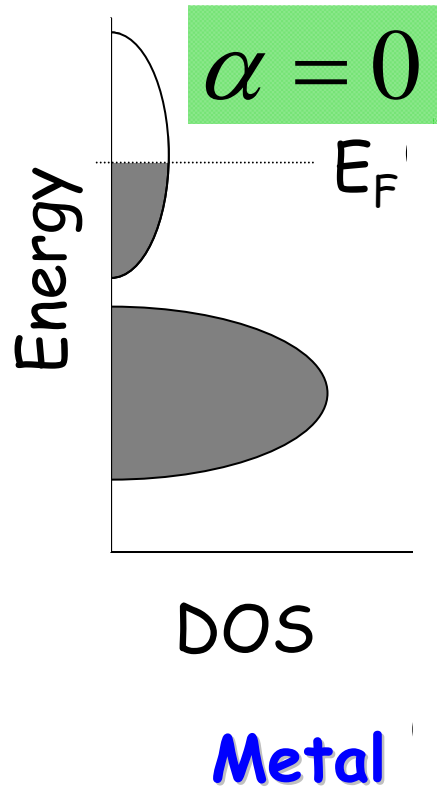
Thermopower in metals

(Band Structure View Point)



Thermopower in metals (Band Structure View Point)

$$D_e(E) = \frac{m}{\hbar^2 \pi^2} \sqrt{\frac{2mE}{\hbar^2}} \text{ in 3D}$$



Thermopower:

- From energy dependent conductivity.

- Mott formula:

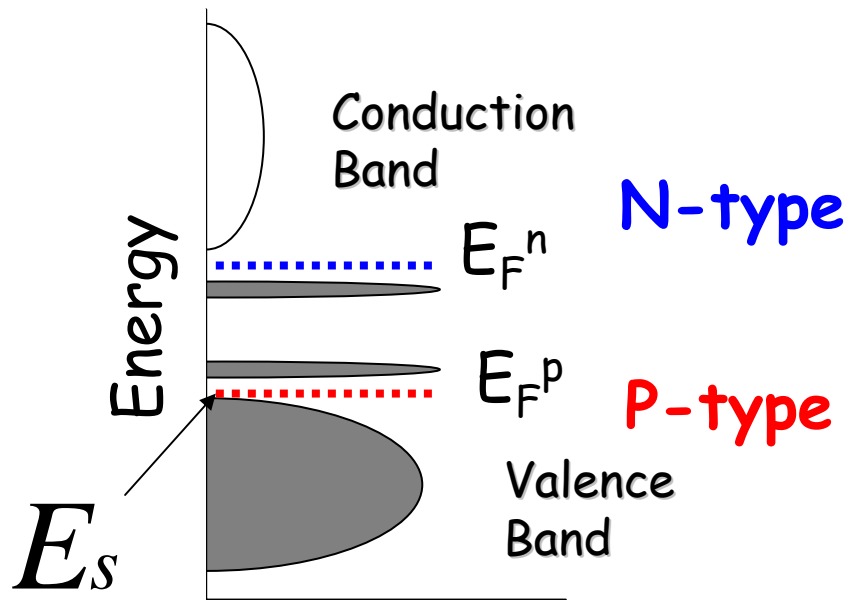
$$s = \left(\frac{\pi^2 k_B^2 T}{3e\sigma} \right) \frac{\partial \sigma}{\partial E} \Big|_{E=E_F}$$

- Note log derivative (not an extensive quantity – multiplicative factors in density of states (specific heat, entropy) or in σ do not change S).

Density of States

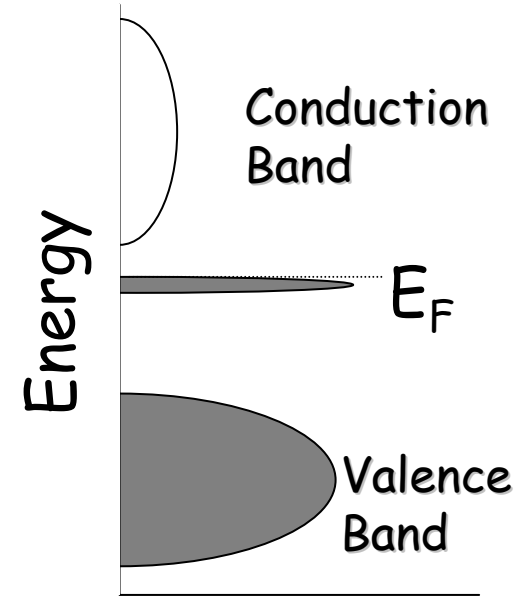
-- Number of electron states available between energy E and $E+dE$

Thermopower in non-metals



DOS

**Band gap
Semiconductor**



DOS

**Hopping
semiconductor
HEIKS (~1960)**

$$S_d = \frac{k_B}{e} \left(\frac{E_s}{k_B} T + B \right)$$

$$S_d = \frac{k_B}{e} \ln \left(\beta_s \beta_o \left\{ \frac{1 - \frac{n}{N}}{\frac{n}{N}} \right\} \right)$$

• B is configurational entropy term

Thermopower for hopping charge carriers

(localised picture)
hopping of charge carriers

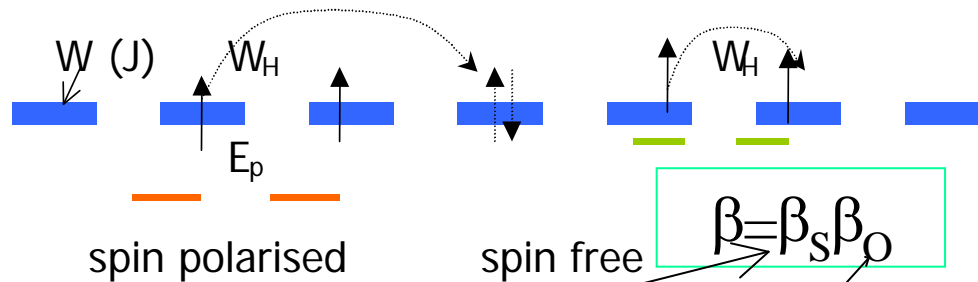
Formulae

entropy

$$S = -\frac{1}{|e|} \frac{\partial \Sigma}{\partial n}$$

hopping polaronic effect

hopping



Spin degeneracy

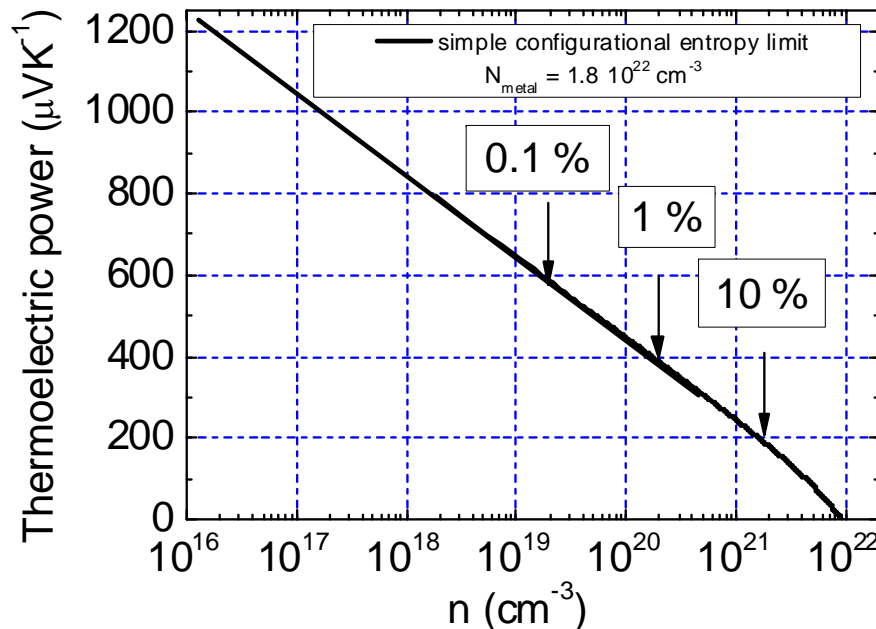
Orbital degeneracy e_g



x = concentration of Co^{4+}

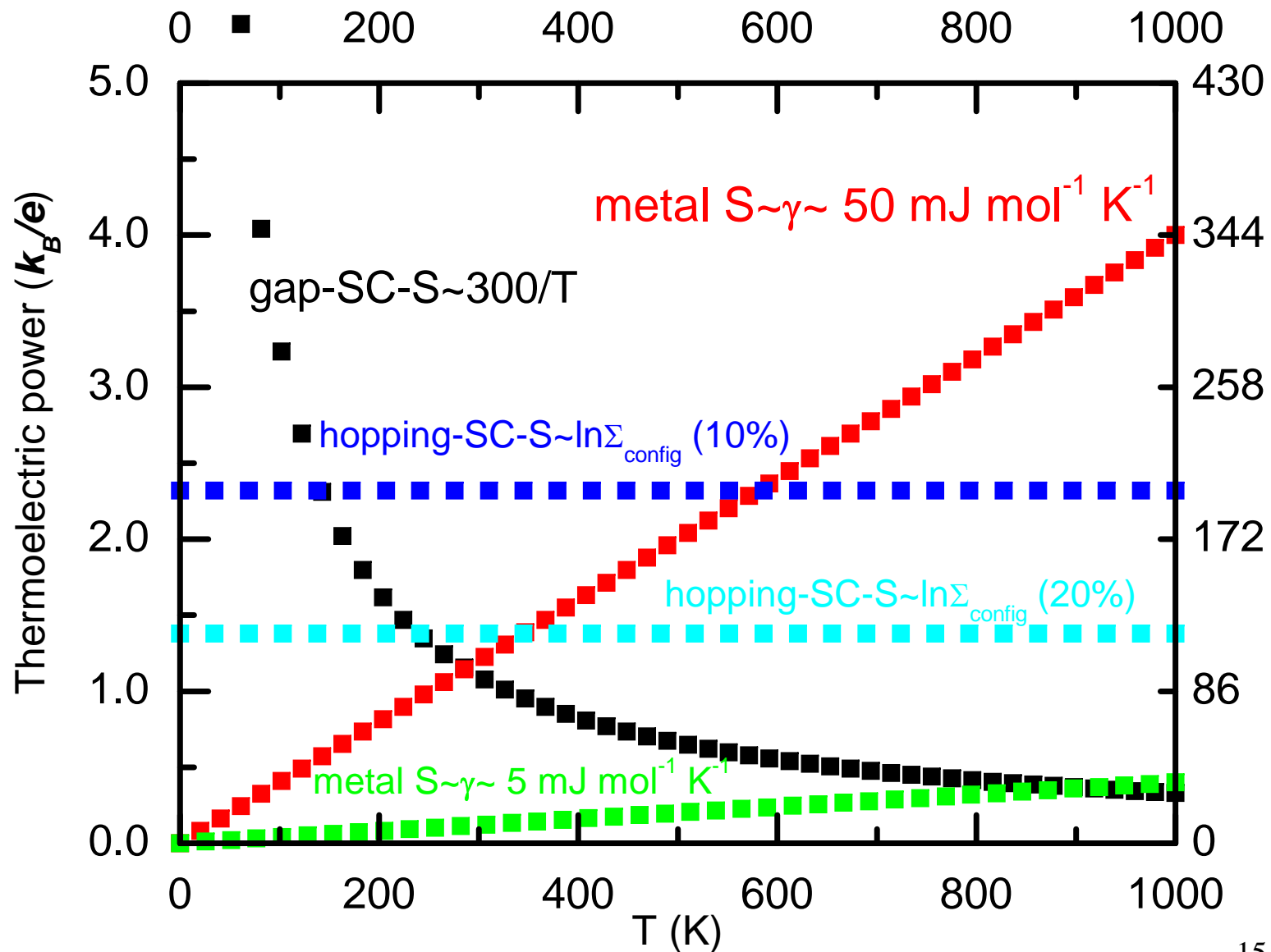
g_3 = degeneracy of Co^{3+}

g_4 = degeneracy of Co^{4+}

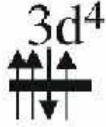


$$S = -\frac{k_B}{e} \ln \left[\frac{g_3}{g_4} \frac{x}{1-x} \right]$$

Thermopower-temperature dependence

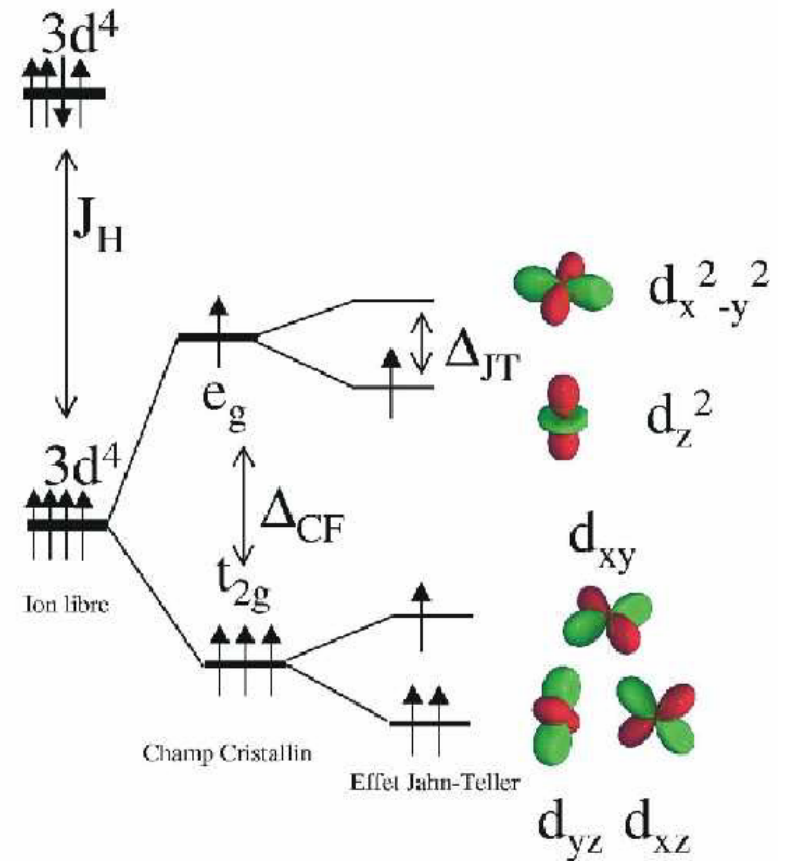


Manganites

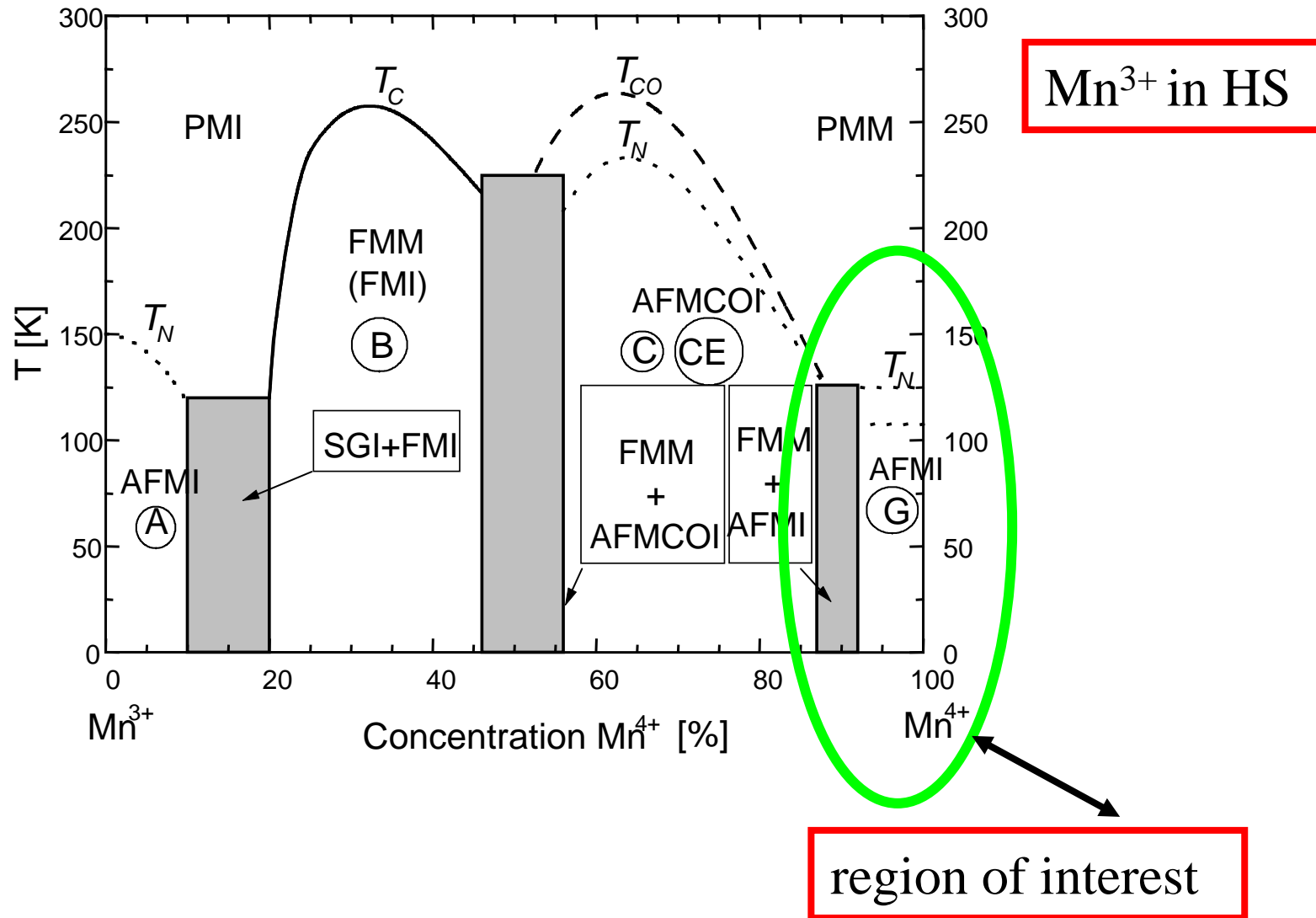
$\text{Mn}^{3+}(\text{t}_{2g}^3\text{e}_g^1, S=2)$, JT active 

$\text{Mn}^{4+}(\text{t}_{2g}^3\text{e}_g^0, S=1.5)$

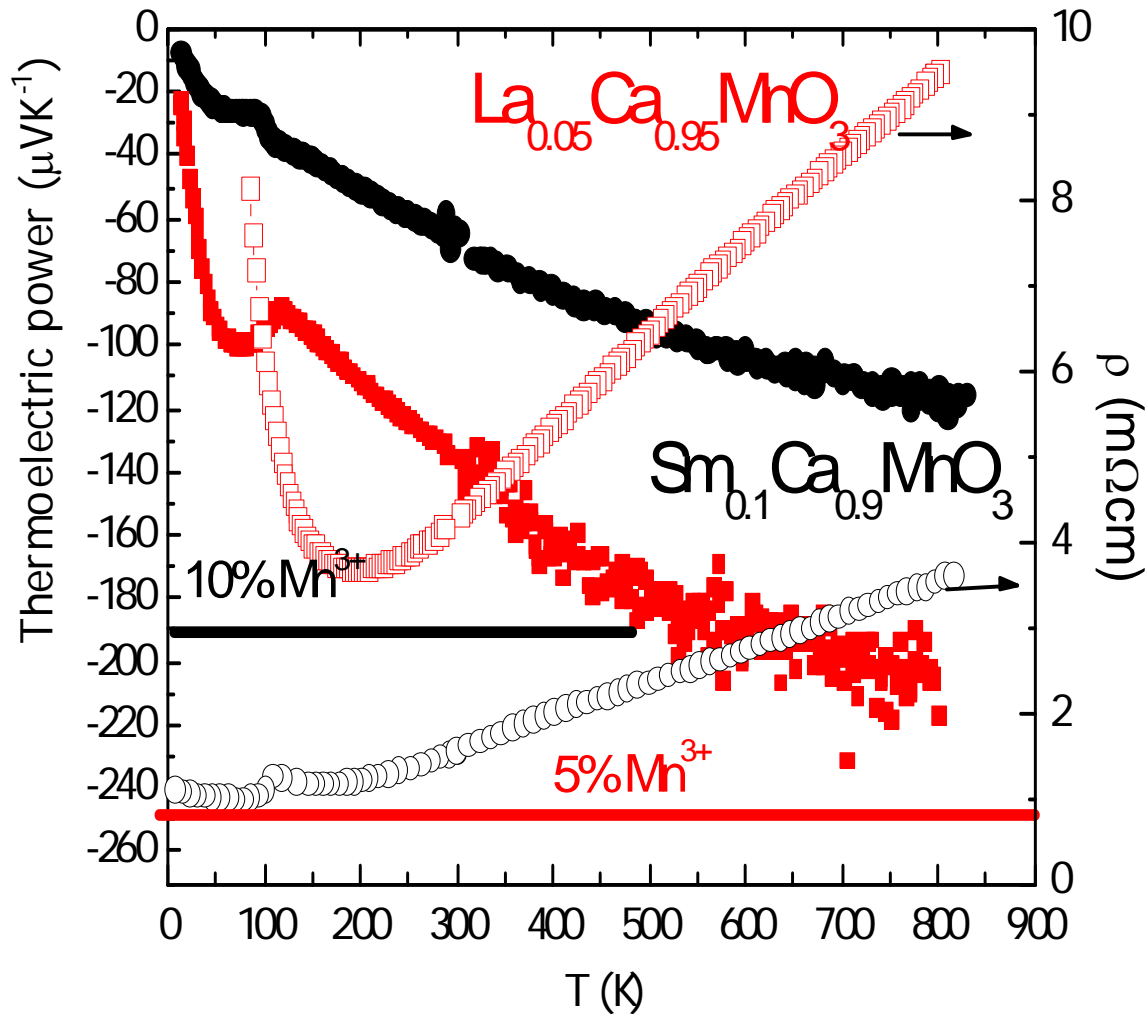
$\text{Mn}^{2+}(\text{t}_{2g}^3\text{e}_g^2, S=2.5)$



Phase diagram of Mn³⁺/Mn⁴⁺ perovskites



The thermoelectric properties of $\text{Ca}_{1-x}\text{Re}_x\text{MnO}_3$ ceramics – low doping



The best thermoelectrics in $\text{Ca}_{1-x}\text{Re}_x\text{MnO}_3$ system

– the Heiks limit for 90% Mn^{4+}

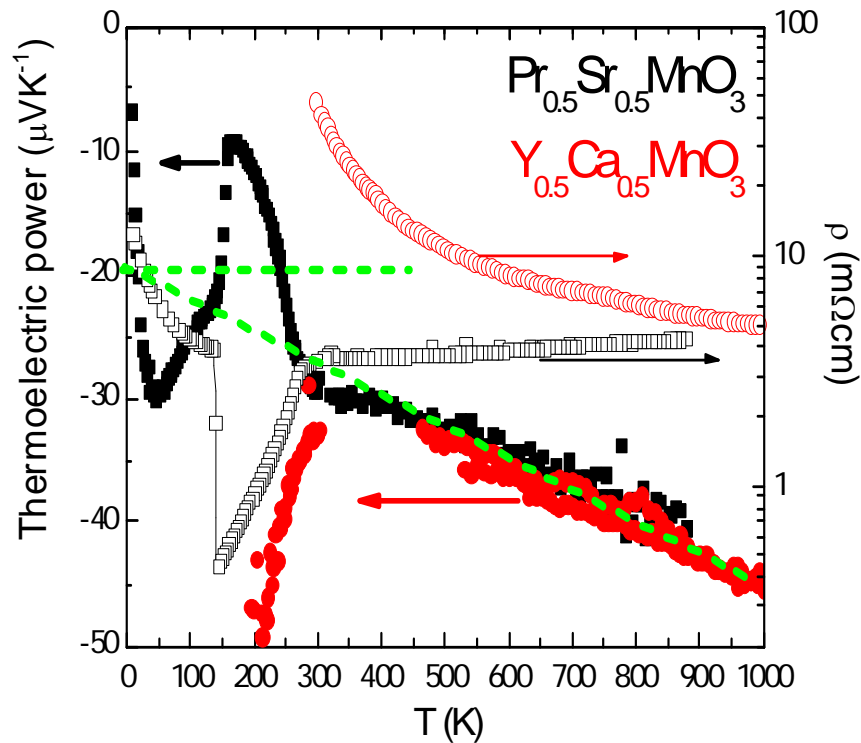
$$S_{90\%} = -190 \mu\text{VK}^{-1}$$

– the Heiks limit for 95% Mn^{4+}

$$S_{95\%} = -250 \mu\text{VK}^{-1}$$

The thermoelectric properties of $\text{Sr}(\text{Ca})_{1-x}\text{Pr}_x\text{MnO}_3$ ceramics – high doping

$$S_{\text{mag}} = \frac{k_B}{e} \ln \left[\frac{2S^{n+1} + 1}{2S^n + 1} \right]$$

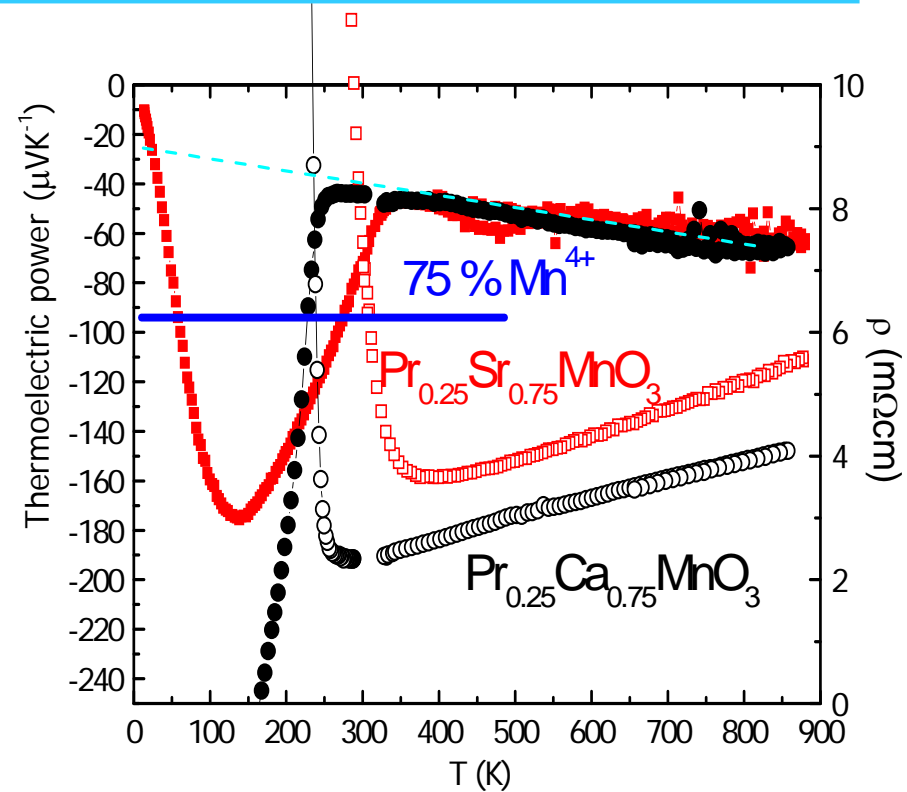


The role of tolerance factor for 50% Mn^{3+}
the Heikes limit for 50% Mn^{3+}

$$S_{50\%} = 0 \mu\text{VK}^{-1}$$

!additional degree of freedom AND diffusive component!

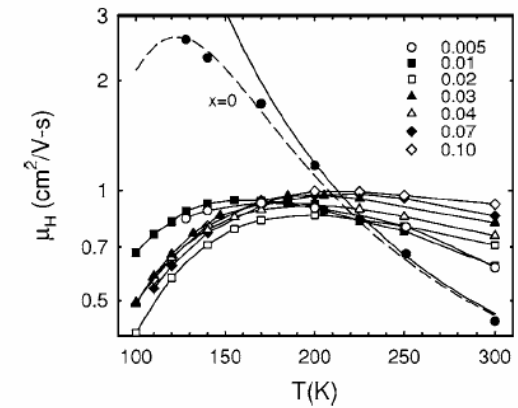
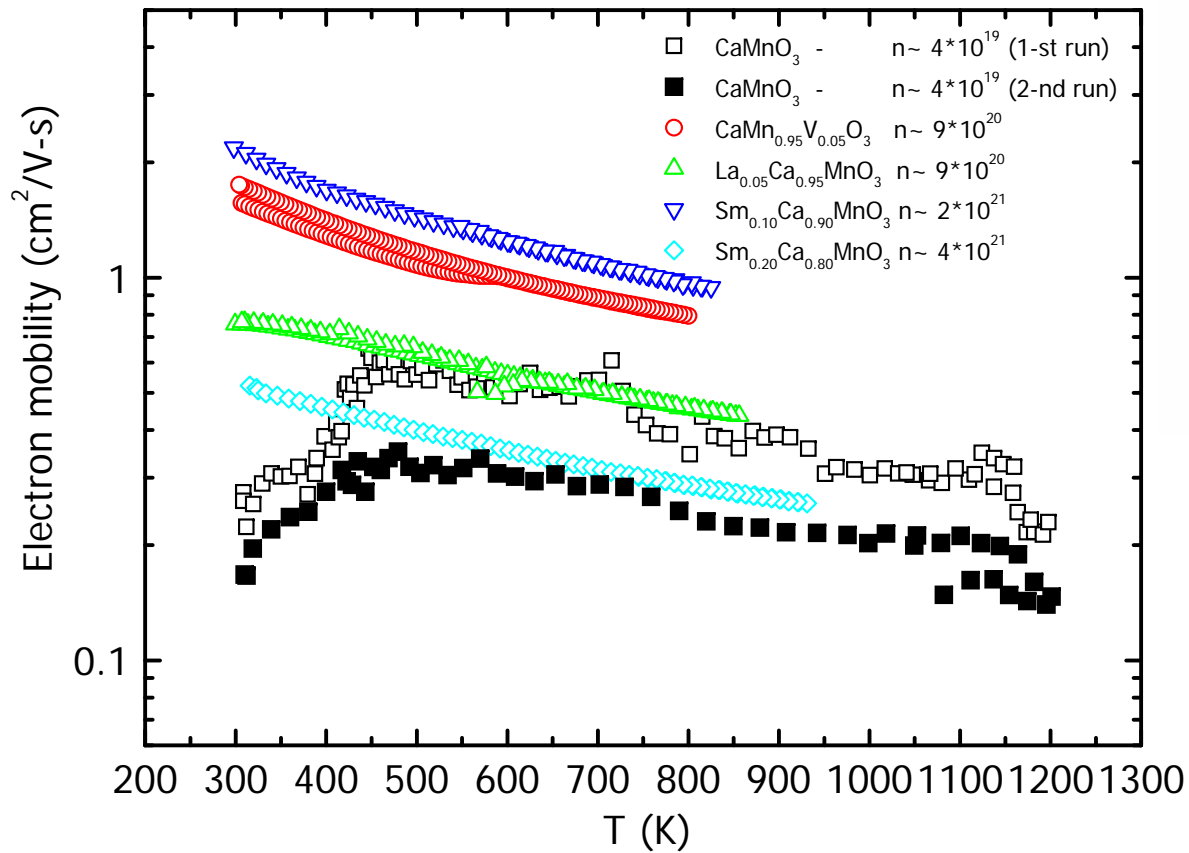
For $\text{Mn}^{3+}/\text{Mn}^{4+}$ HS = $-19 \mu\text{VK}^{-1}$



The thermoelectrics in $\text{A}_{1-x}\text{Pr}_x\text{MnO}_3$
system – the Heikes limit for 25% Mn^{3+}

$$\text{HEIKS } S_{75\%} = -95 \mu\text{VK}^{-1}$$

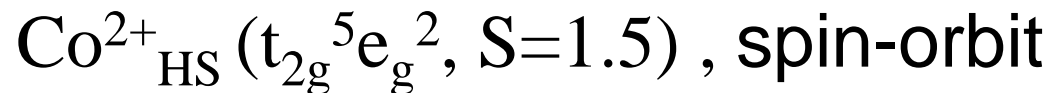
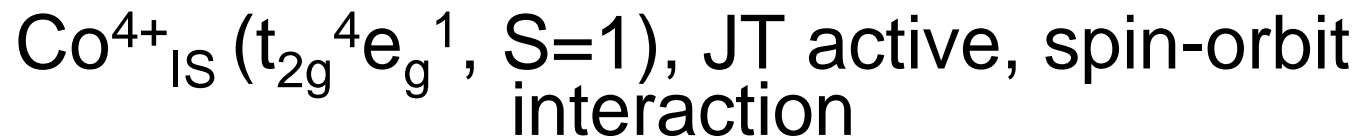
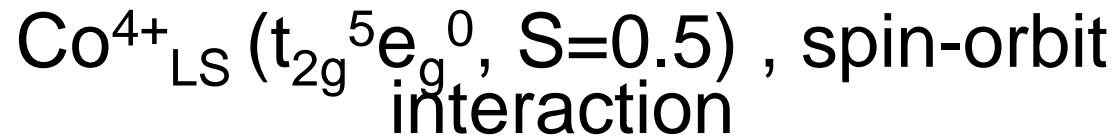
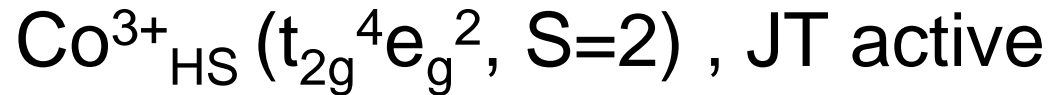
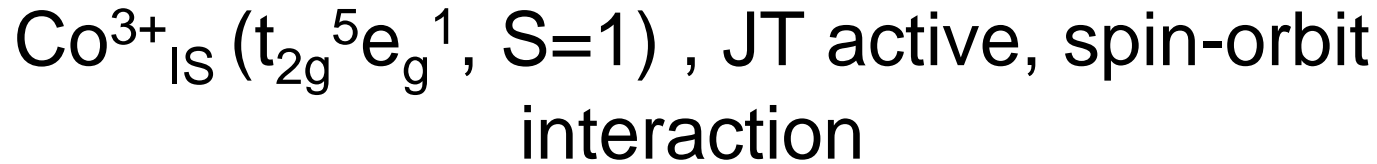
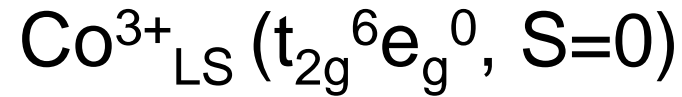
The temperature dependence of the charge carrier mobility for Mn⁴⁺ rich manganites



J. L. Cohn, C. Chiorescu, and J. J. Neumeier, Phys. Rev B, 72, (2005) 024422

-μ similar to that observed in Sr_{1-x}La_xTiO₃

Cobaltites

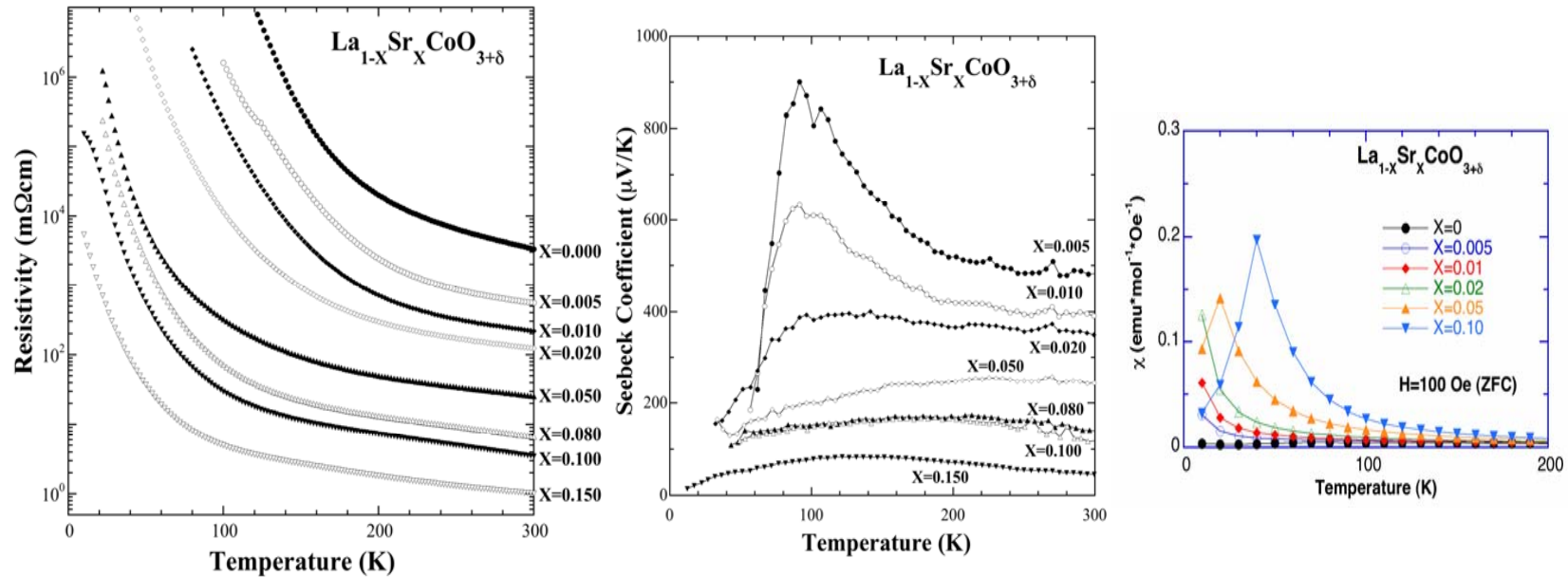


Possible spin states and total degeneracy of ground-states of Co^{2+} , Co^{3+} a Co^{4+} species
(S=Spin only number neglecting orbital moment)

Ionic state	HS ($J_H > \Delta_{CF}$)		LS ($J_H < \Delta_{CF}$)		IS ($J_H \sim \Delta_{CF}$)	
	No distortion	Distortion ($\Delta_{JT} \gg 0$)	No distortion	Distortion ($\Delta_{JT} \gg 0$)	No distortion	Distortion ($\Delta_{JT} \gg 0$)
Co^{2+}					×	×
	$G_{\text{spin}} = 4$	$G_{\text{spin}} = 4$	$G_{\text{spin}} = 2$	$G_{\text{spin}} = 2$		
	$G_{\text{orb}} = 3$	$G_{\text{orb}} = 1$	$G_{\text{orb}} = 2$	$G_{\text{orb}} = 1$		
	Gtot = 12	Gtot = 4	Gtot = 4	Gtot = 2		
Co^{3+}				×		
	$G_{\text{spin}} = 5$	$G_{\text{spin}} = 5$	$G_{\text{spin}} = 1$		$G_{\text{spin}} = 3$	$G_{\text{spin}} = 3$
	$G_{\text{orb}} = 3$	$G_{\text{orb}} = 1$	$G_{\text{orb}} = 1$		$G_{\text{orb}} = 6$	$G_{\text{orb}} = 1$
	Gtot = 15	Gtot = 5	Gtot = 1		Gtot = 18	Gtot = 3
Co^{4+}		×				
	$G_{\text{spin}} = 6$		$G_{\text{spin}} = 2$	$G_{\text{spin}} = 2$	$G_{\text{spin}} = 4$	$G_{\text{spin}} = 4$
	$G_{\text{orb}} = 1$		$G_{\text{orb}} = 3$	$G_{\text{orb}} = 1$	$G_{\text{orb}} = 6$	$G_{\text{orb}} = 1$
	Gtot = 6		Gtot = 6	Gtot = 2	Gtot = 24	Gtot = 4

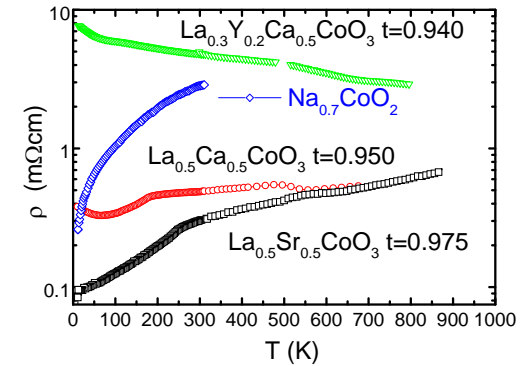
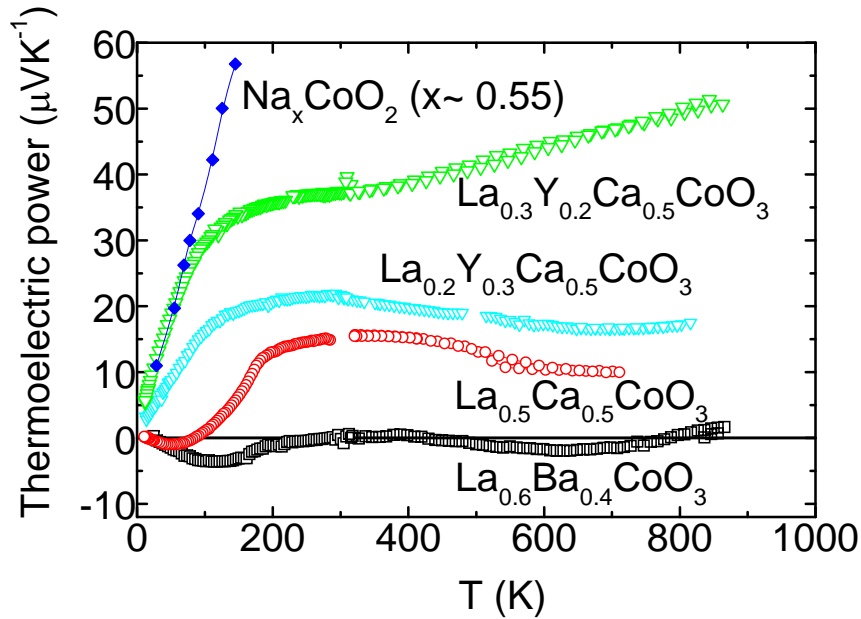
3D oxide perovskites –transport and magnetism

Sr,Ba-doped LaCoO_3 LS—LS/HS —IS σ^*

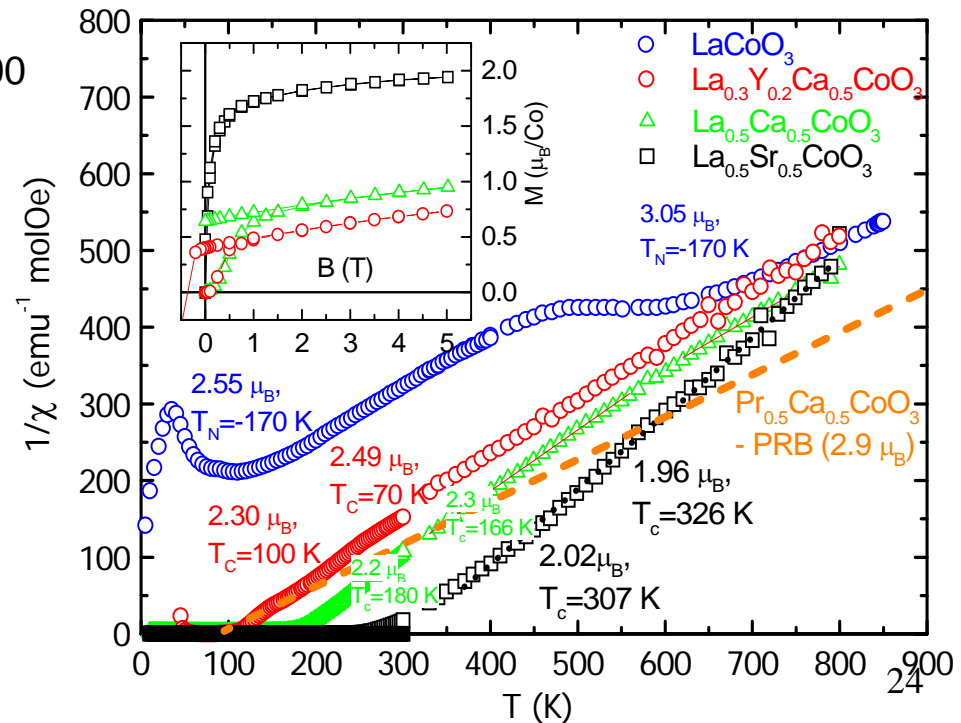


- M-I transition is linked with magnetic one, metallic samples are FM with enhanced metallicity below T_c ($t_{2g}^5\sigma^*$)
- For low x the thermoelectric power is temperature weakly dependent, the absolute value at room temperature corresponds to that deduced from a simple configurational entropy approximation ($x=0.005$ $S_{\text{Heiks}} = 455 \mu\text{VK}^{-1} \Leftrightarrow S_{\text{exp}} \sim 500 \mu\text{VK}$, $x=0.05$ $S_{\text{exp}} \sim 250 \mu\text{VK}^{-1} \Leftrightarrow S_{\text{Heiks}} \sim 257 \mu\text{VK}^{-1}$)

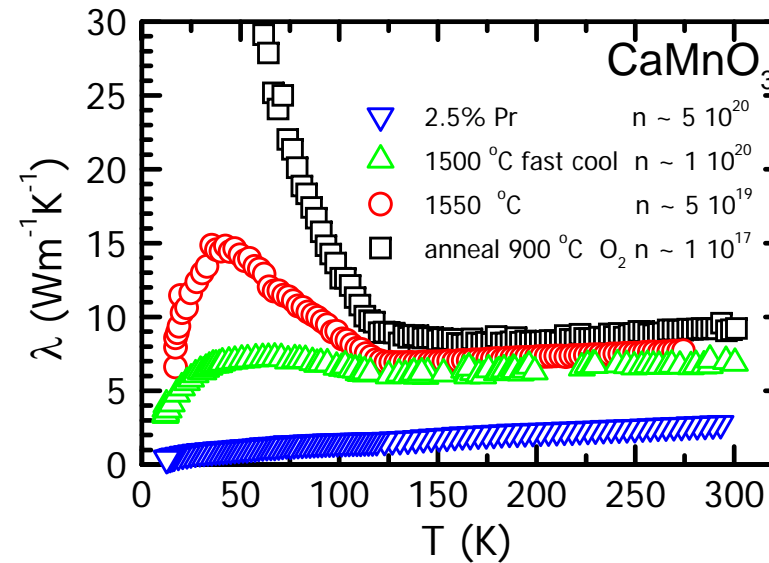
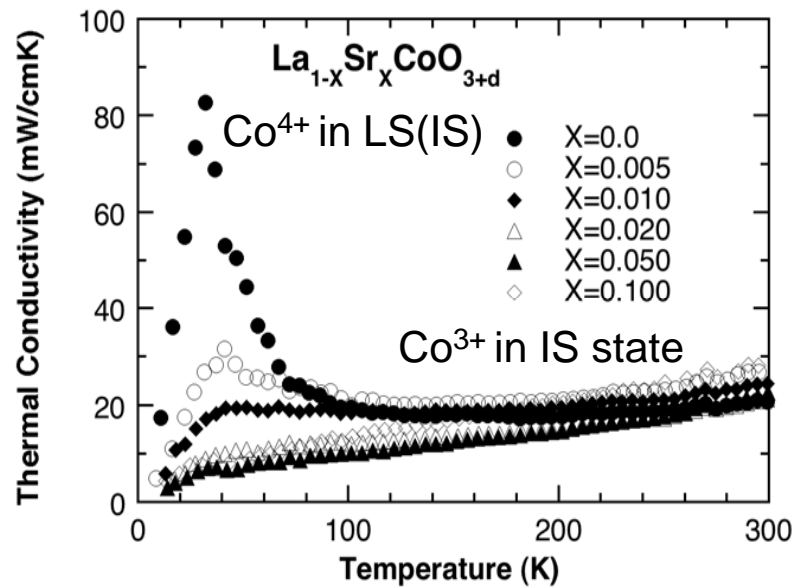
Co³⁺:Co⁴⁺ =1:1; role of tolerance factor t



- The ferromagnetic moment and T_c decreases with decreasing t
- the low temperature metallic resistivity changes to temperature activated behaviour
- the generalised Heikes formula explains the hopping conductivity (La-Y-Ca) supposing the IS spin state of both Co³⁺ and Co⁴⁺
- Ferromagnetic metallic state (La-Sr) is characterized by negative thermoelectric power corresponding to the $t_{2g}^5 \sigma^{*0.5}$ electronic configuration (IS Co³⁺ and LS Co⁴⁺)



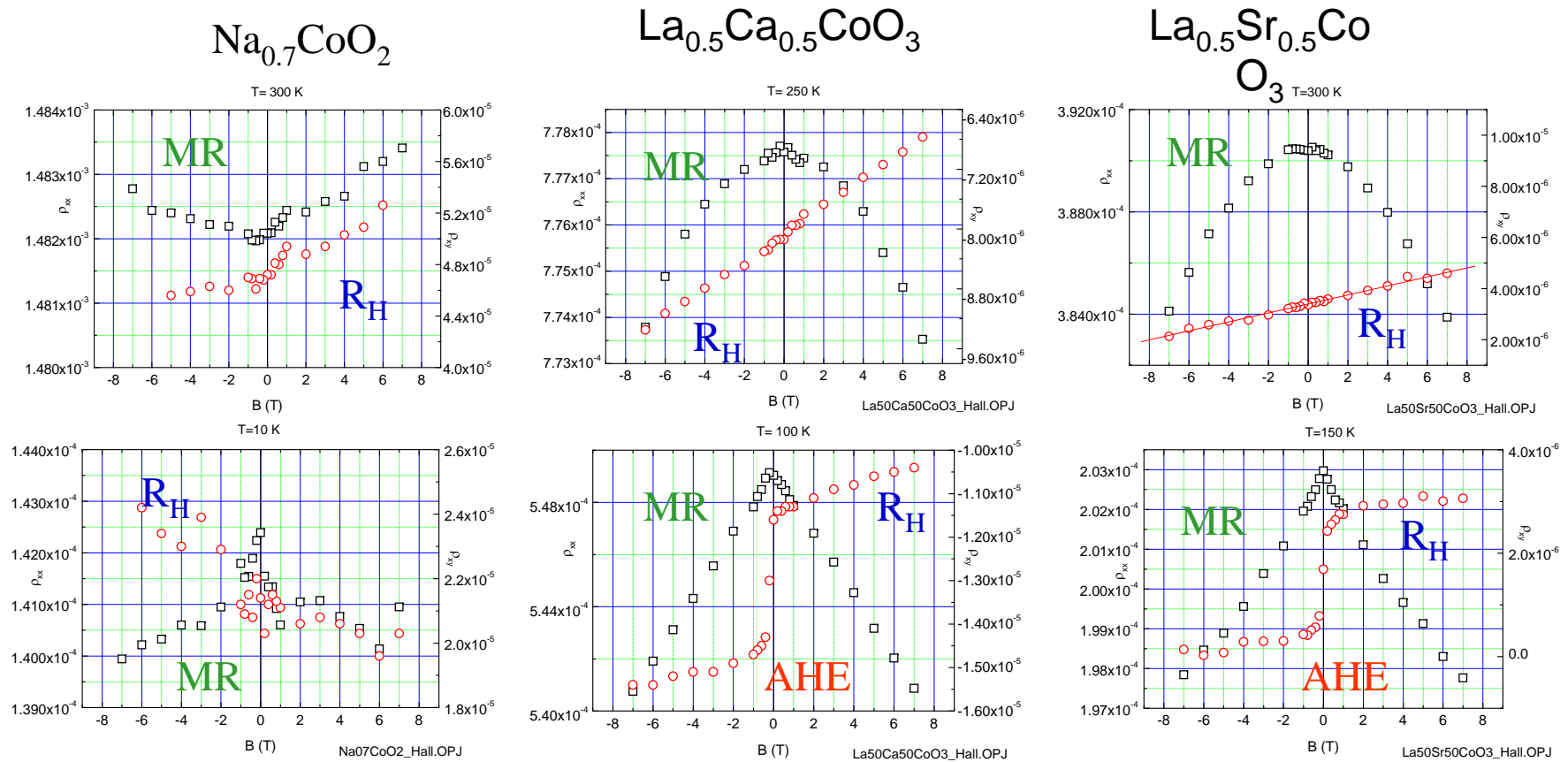
Sr-doped LaCoO_3 LS-HS, CaMnO_3 , spin fluctuations



- Thermal conductivity is high for LS Co³⁺, Co⁴⁺
- the depressed thermal conductivity in $\text{Ln}_{1-x}\text{A}_x\text{CoO}_3$ is due to fast electron fluctuations between LS / HS Co³⁺ species (LS ⇌ HS+LS)
- the gradual stabilization of IS Co³⁺ species of $t_{2g}^5 \sigma^{*0.5}$ character
- spin density fluctuations are supposed to decrease the thermal conductivity in pure CaMnO_3 above T_N

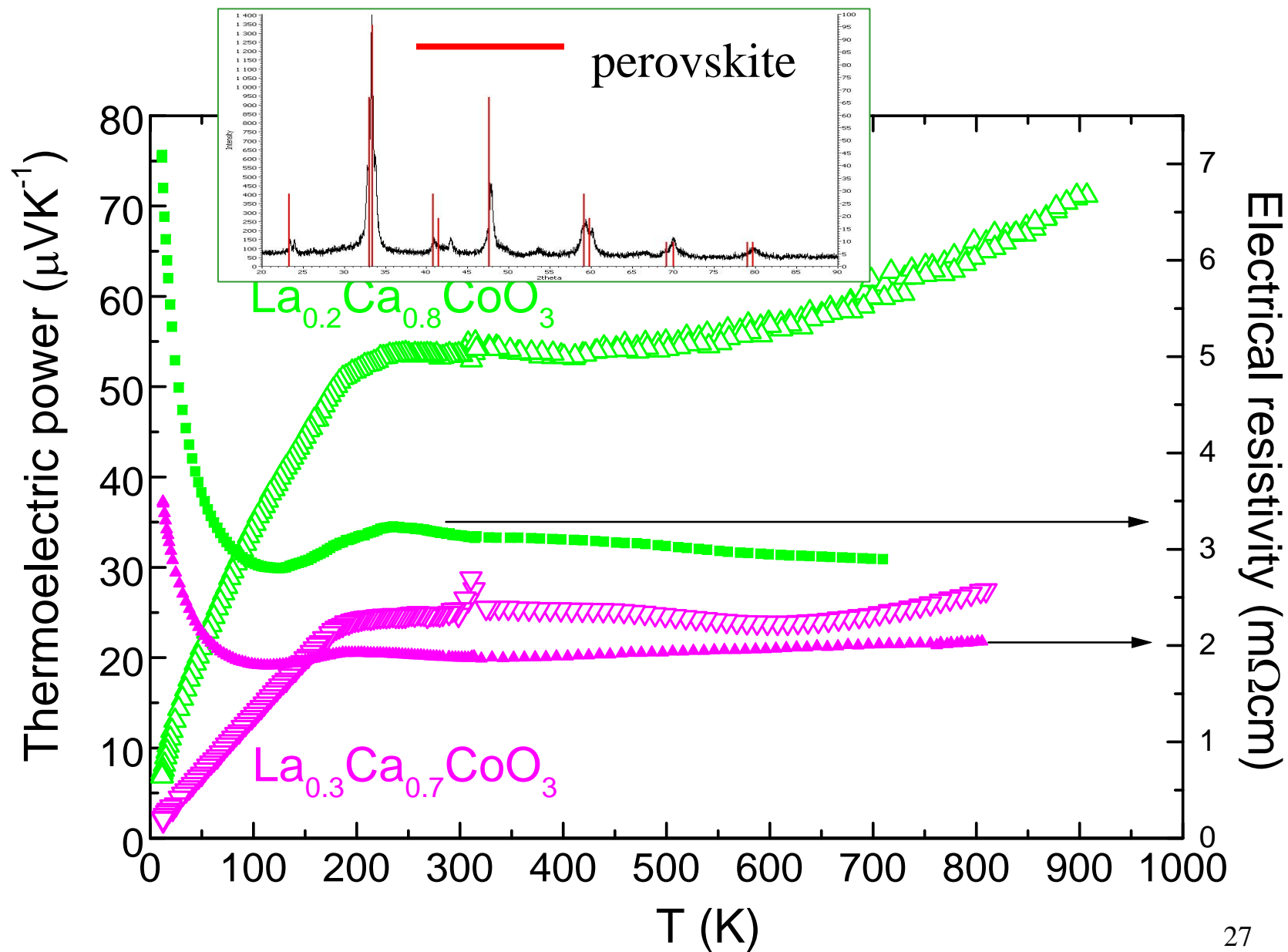
Cobaltites – magnetotransport & carrier concentration (n/Co)

Anomalous Hall Effect

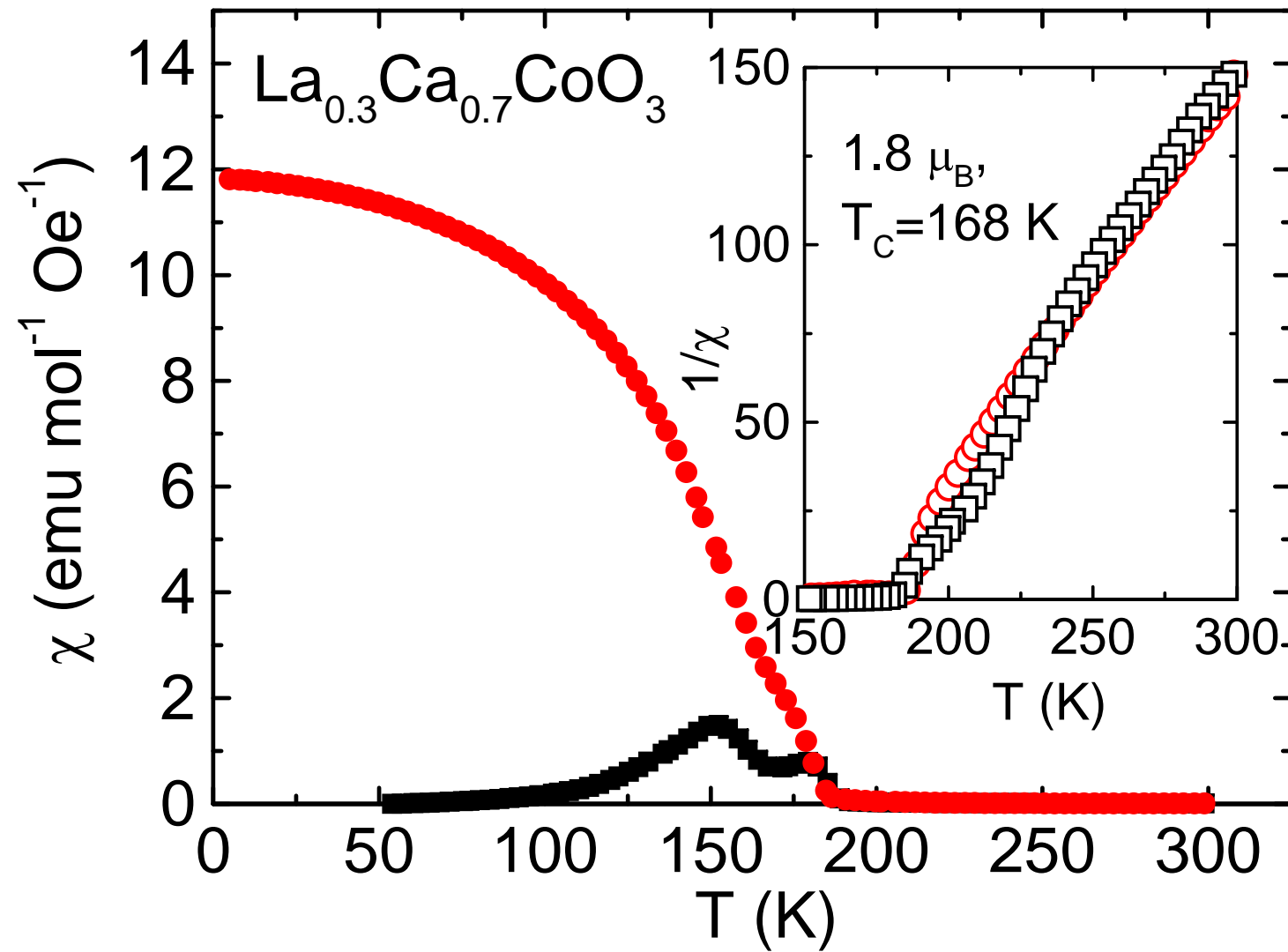


T	$\text{Na}_{0.7}\text{CoO}_2$	$\text{La}_{0.5}\text{Ca}_{0.5}\text{CoO}_3$	$\text{La}_{0.5}\text{Sr}_{0.5}\text{CoO}_3$
5	-0.04	0.3	0.7
20	-0.06	0.3	0.7
100	0.02	0.3	0.8
150	~0	0.17	0.3
200	0.02	0.1	0.6
225		0.13	0.6
250		0.25	
300	0.03	0.4	0.25
350	0.03	0.3	0.2

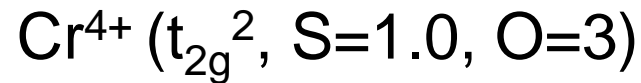
THERMOELECTRIC PERFORMANCE 3D oxide perovskites $\text{La}_{1-x}\text{Ca}_x\text{CoO}_3$



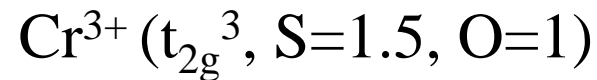
Magnetic properties of $\text{La}_{0.3}\text{Ca}_{0.7}\text{CoO}_3$ – magnetic susceptibility



Chromites



$G_{\text{spin}}=3, G_{\text{orb}}=3, G_{\text{tot}}= 9$



$G_{\text{spin}}=4, G_{\text{orb}}=1, G_{\text{tot}}= 4$

$$\frac{\text{Cr}^{3+}}{\text{Cr}^{4+}} = 1$$

$$S_{\text{mag}} = -\frac{k_B}{e} \ln \left(\frac{G_{\text{spin}}^{\text{Cr}^{3+}}}{G_{\text{spin}}^{\text{Cr}^{4+}}} \right) = -29 \mu\text{VK}^{-1}$$

$$S_{\text{mag+orb}} = -\frac{k_B}{e} \ln \left(\frac{G_{\text{tot}}^{\text{Cr}^{3+}}}{G_{\text{tot}}^{\text{Cr}^{4+}}} \right) = +69 \mu\text{VK}^{-1}$$

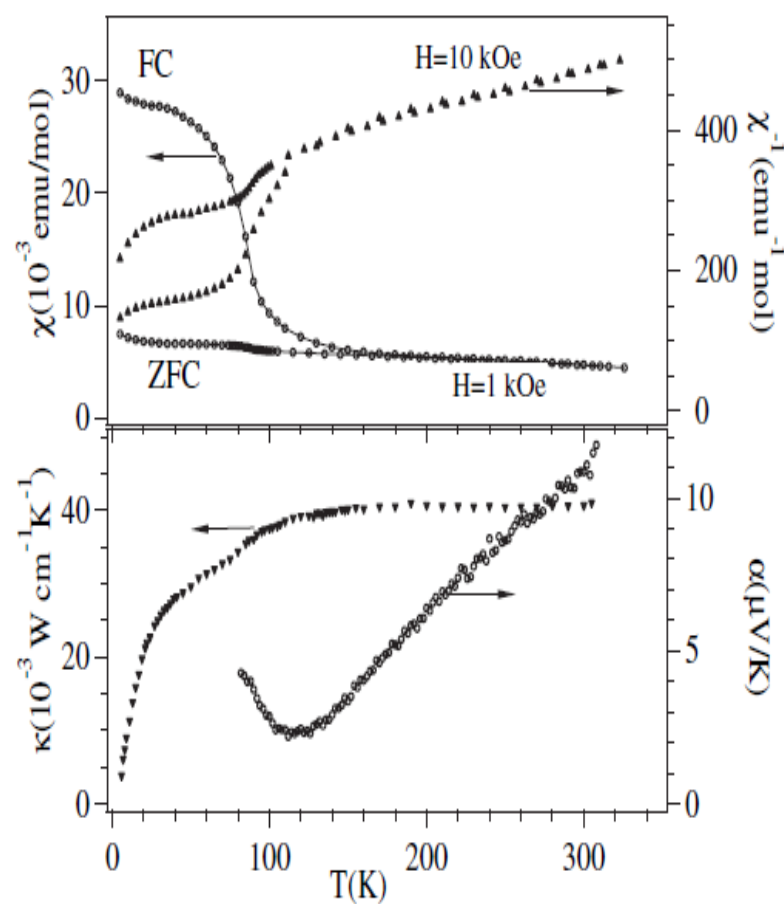
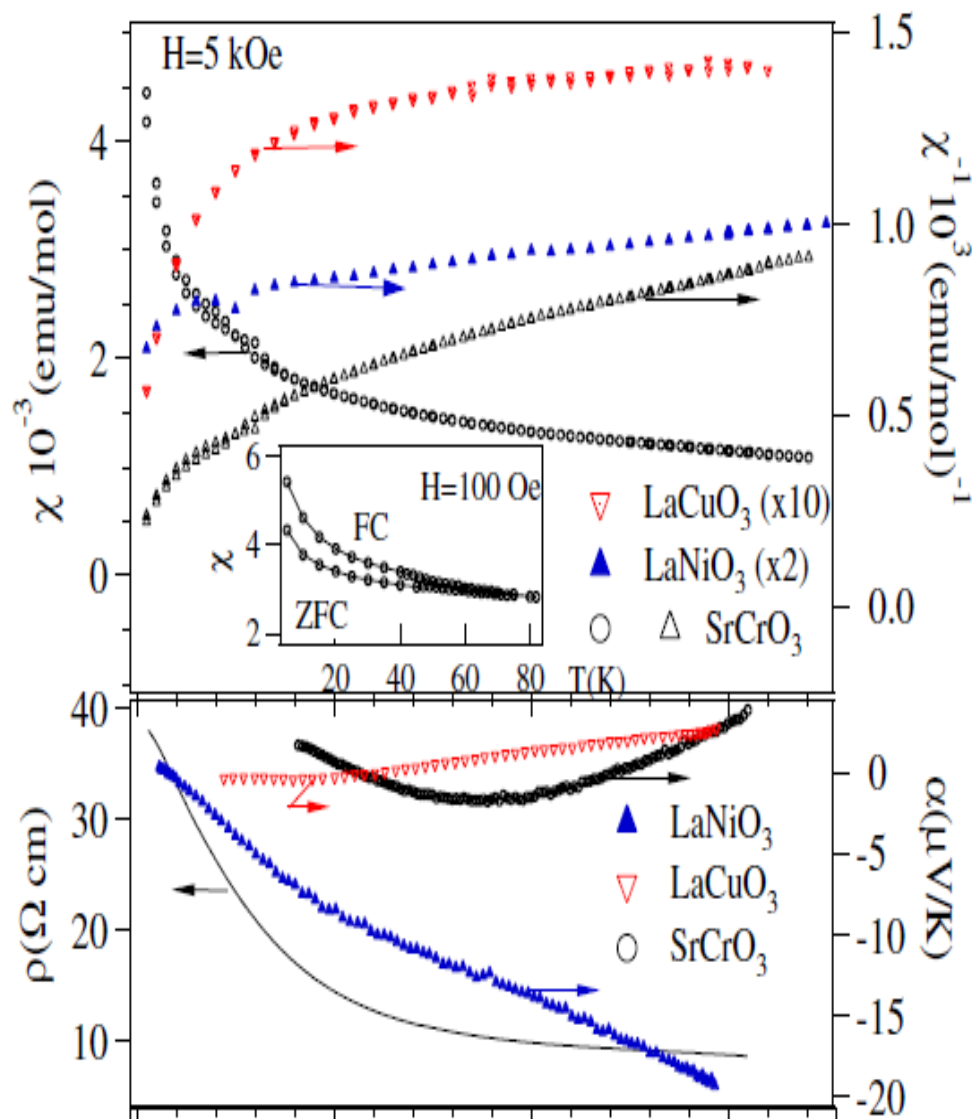
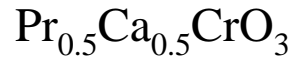
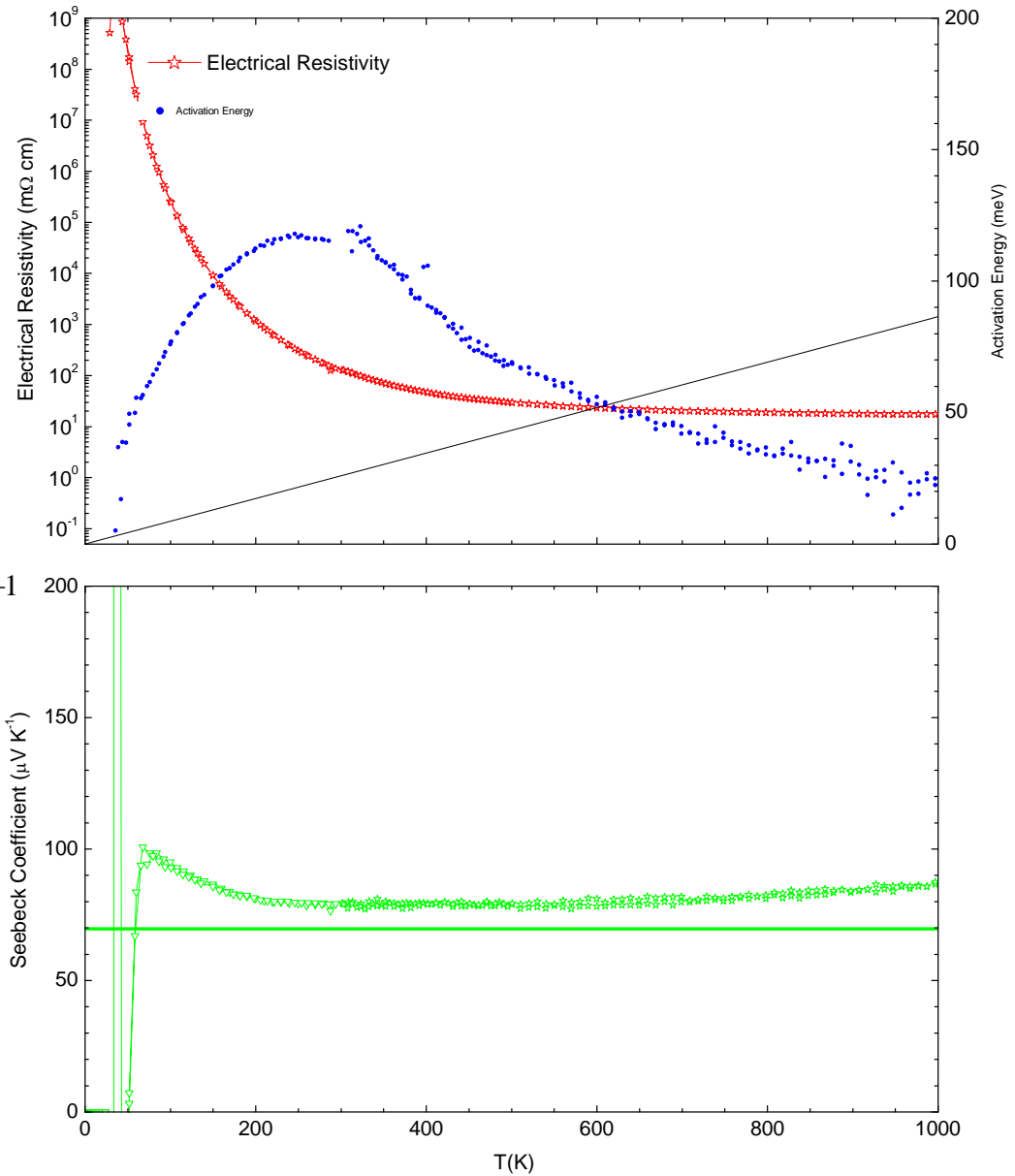


FIG. 3. The same as Fig. 2 (without resistivity) for CaCrO_3 . Inset: magnetization to 5.5 T at 5 K.



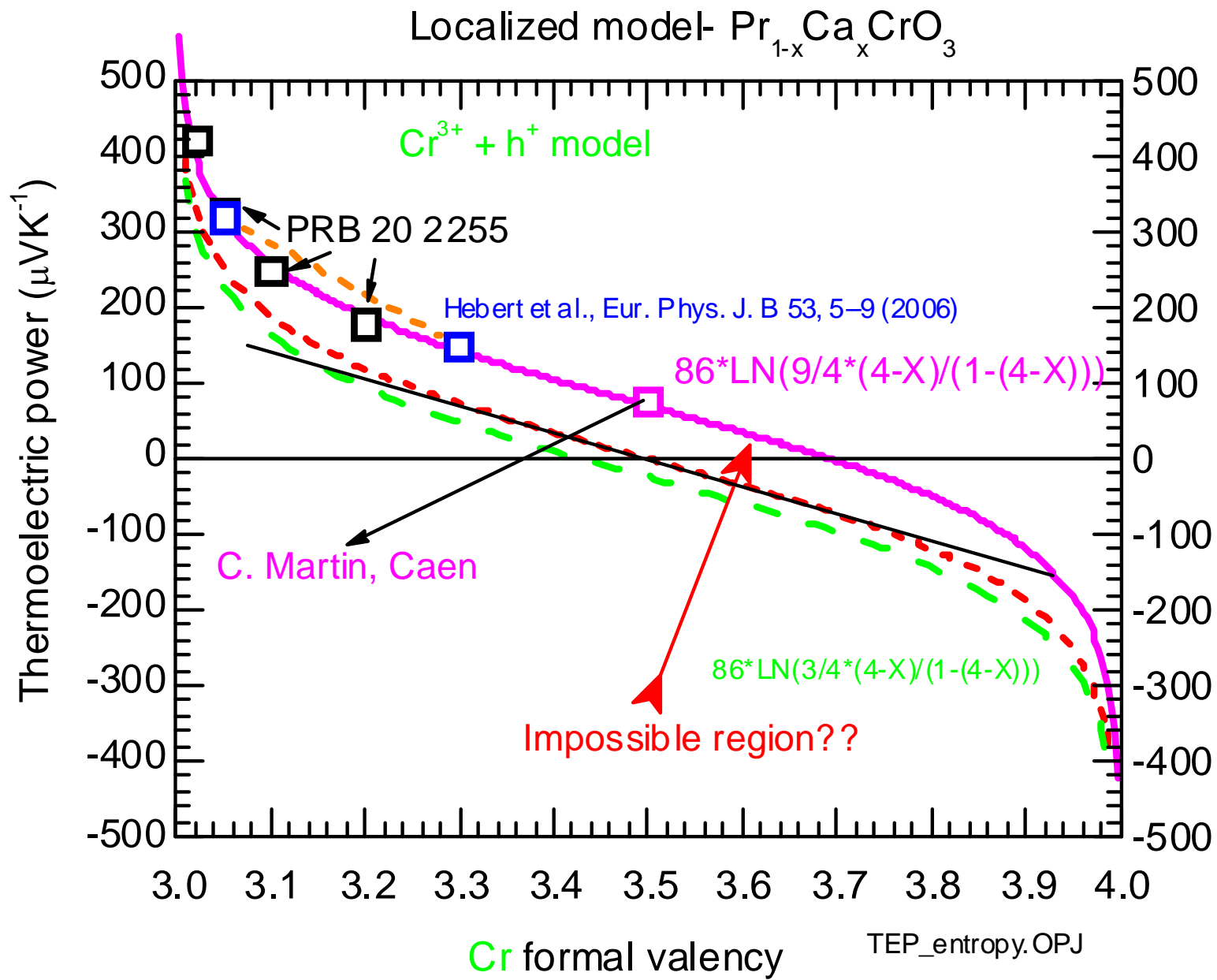


$$S_{\text{mag+orb}} = -\frac{k_B}{e} \ln\left(\frac{G_{\text{tot}}^{\text{Cr}^{3+}}}{G_{\text{tot}}^{\text{Cr}^{3+}}}\right) = +69 \mu\text{VK}^{-1}$$



➤ complex configurational entropy approximation applies

➤ (magnetic, orbital contribution) $S_{\text{Heiks}} = +69 \mu\text{VK}^{-1}$



Ferrites

LS Fe⁴⁺ (t_{2g}⁴, S=1.0, O=3)

$$G_{\text{spin}}=3, G_{\text{orb}}=3, G_{\text{tot}}= 9$$

$$\frac{\text{Fe}^{3+}}{\text{Fe}^{4+}} = 1$$

$$S^{LS}_{\text{mag}} = -\frac{k_B}{e} \ln \left(\frac{G_{\text{spin}}^{\text{Fe}^{3+}}}{G_{\text{spin}}^{\text{Fe}^{4+}}} \right) = -59 \mu\text{VK}^{-1}$$

HS Fe⁴⁺ (t_{2g}⁴, S=2.0, O=2)

$$G_{\text{spin}}=5, G_{\text{orb}}=2, G_{\text{tot}}= 10$$

$$S^{HS}_{\text{mag}} = -\frac{k_B}{e} \ln \left(\frac{G_{\text{spin}}^{\text{Fe}^{3+}}}{G_{\text{spin}}^{\text{Fe}^{4+}}} \right) = -16 \mu\text{VK}^{-1}$$

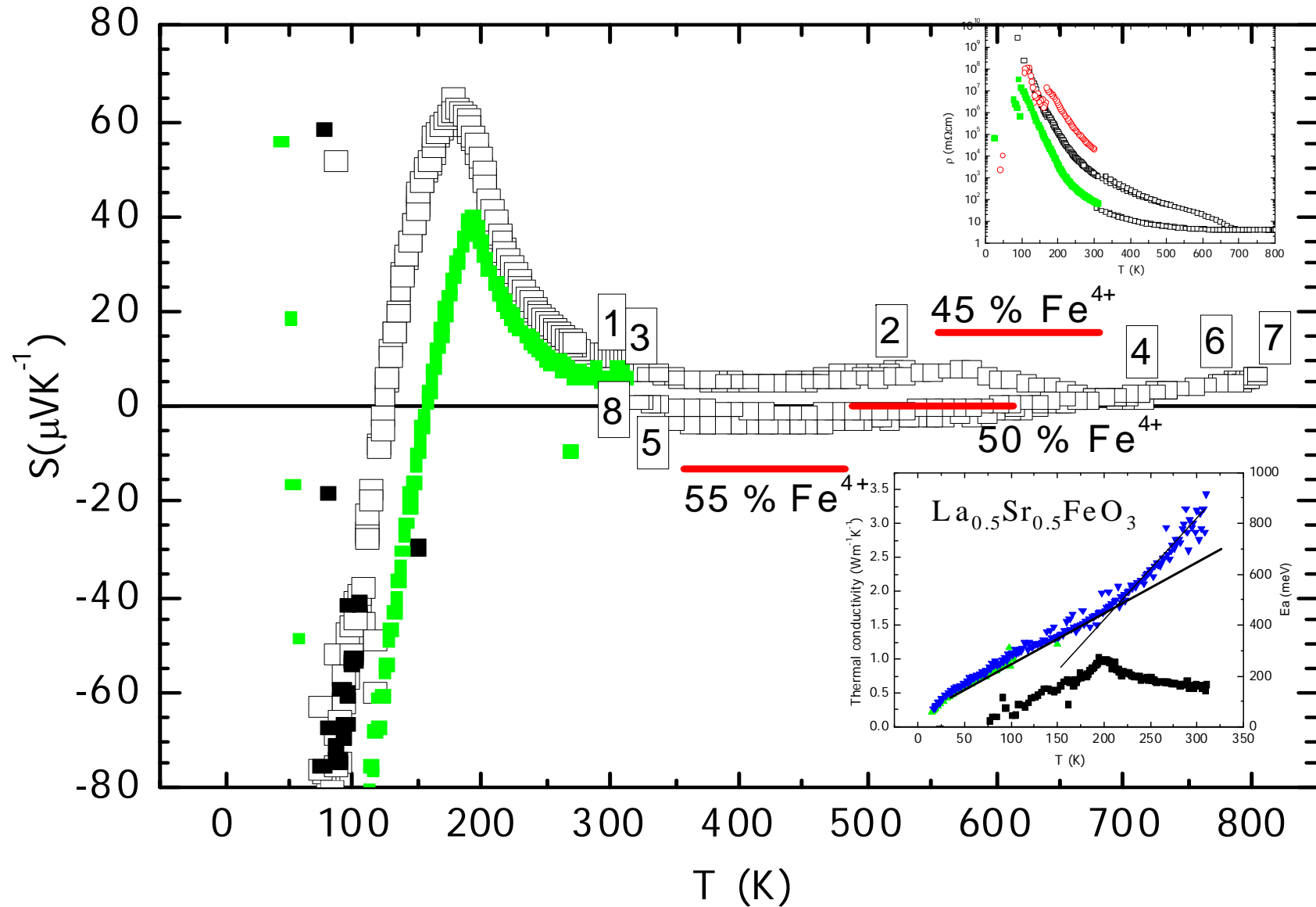
$$S^{LS}_{\text{mag+orb}} = -\frac{k_B}{e} \ln \left(\frac{G_{\text{tot}}^{\text{Fe}^{3+}}}{G_{\text{tot}}^{\text{Fe}^{4+}}} \right) = +35 \mu\text{VK}^{-1}$$

HS Fe³⁺ (t_{2g}⁵, S=2.5, O=1)

$$G_{\text{spin}}=6, G_{\text{orb}}=1, G_{\text{tot}}= 6$$

$$S^{HS}_{\text{mag+orb}} = -\frac{k_B}{e} \ln \left(\frac{G_{\text{tot}}^{\text{Fe}^{3+}}}{G_{\text{tot}}^{\text{Fe}^{4+}}} \right) = +44 \mu\text{VK}^{-1}$$

La_{0.5}Sr_{0.5}FeO₃ – oxidation during measurement



➤ a simple configurational entropy approximation applies

(no magnetic, no orbital contribution) $S_{\text{Heiks}} = 0 \mu\text{VK}^{-1}$

SrFeO_{3-δ}

E. Hemery, thesis, Victoria University of Wellington, 2007

HS Fe⁴⁺ 2S+1=5, O=2, SO=10
 LS Fe⁴⁺ 2S+1=3, O=3, SO=9
 HS Fe³⁺ 2S+1=6, O=1, SO=6
 S_{mag} = -18 μV/K

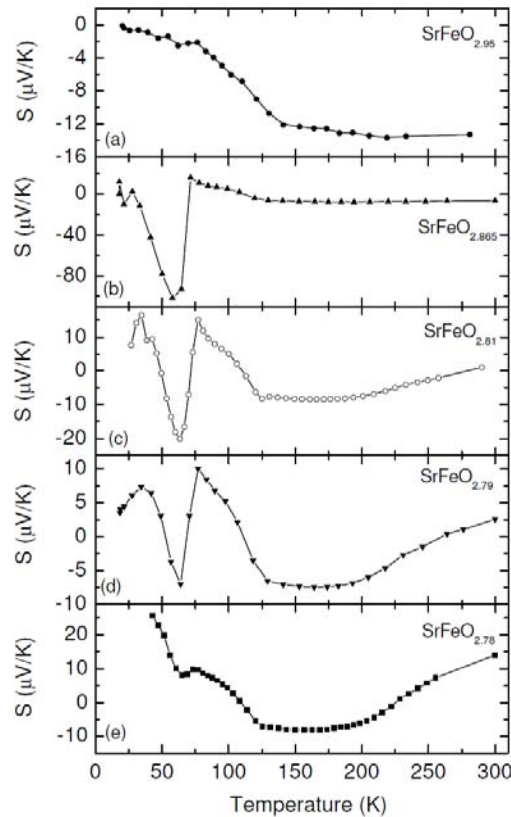


Figure 7.8: TEP versus temperature for (a) SrFeO_{2.95}, (b) SrFeO_{2.865}, (c) SrFeO_{2.81}, (d) SrFeO_{2.79} and (e) SrFeO_{2.78}.

E. Hemery, thesis, Victoria University of Wellington, 2007

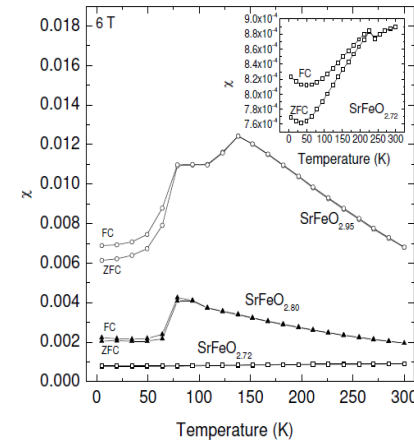


Figure 7.3: Zero Field Cooled (ZFC) and Field Cooled (FC) magnetisation at 6 T for SrFeO_{2.95}, SrFeO_{2.80} and SrFeO_{2.72}. Inset: zoom in of the SrFeO_{2.72} measurement.

$$S_{mag}^{HS} = -\frac{k_B}{e} \ln \left(\frac{G_{spin}^{Fe^{3+}}}{G_{spin}^{Fe^{4+}}} \right) = -16 \mu V K^{-1}$$

➤ configurational entropy approximation applies above 150 K

➤ (magnetic, no orbital contribution) $S_{\text{Heik}} = -16 \mu V K^{-1}$



E. Hemery, thesis, Victoria University of Wellington, 2007

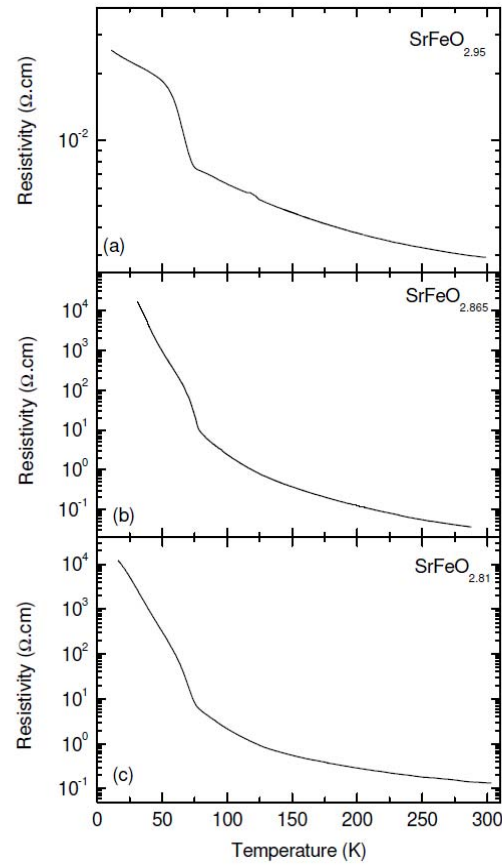


Figure 7.6: Temperature dependence of the resistivity for (a) $\text{SrFeO}_{2.95}$, (b) $\text{SrFeO}_{2.865}$ and (c) $\text{SrFeO}_{2.81}$.

Ruthenates

Ru^{5+} (t_{2g}^3 , $S=1.5$, $O=1$) No mixing entropy, only magnetic or orbital

$$G_{\text{spin}}=4, G_{\text{orb}}=1, G_{\text{tot}}= 4$$

$$S^{LS}_{\text{mag}} = -\frac{k_B}{e} \ln \left(\frac{G_{\text{spin}}^{\text{Ru}^{5+}}}{G_{\text{spin}}^{\text{Ru}^{4+}}} \right) = +25 \mu\text{VK}^{-1}$$

Ru^{4+} (t_{2g}^4 , $S=1.0$, $O=3$)

$$S^{LS}_{\text{mag}} = -\frac{k_B}{e} \ln \left(\frac{G_{\text{spin}}^{\text{Ru}^{4+}}}{G_{\text{spin}}^{\text{Ru}^{3+}}} \right) = +35 \mu\text{VK}^{-1}$$

$$G_{\text{spin}}=3, G_{\text{orb}}=3, G_{\text{tot}}= 9$$

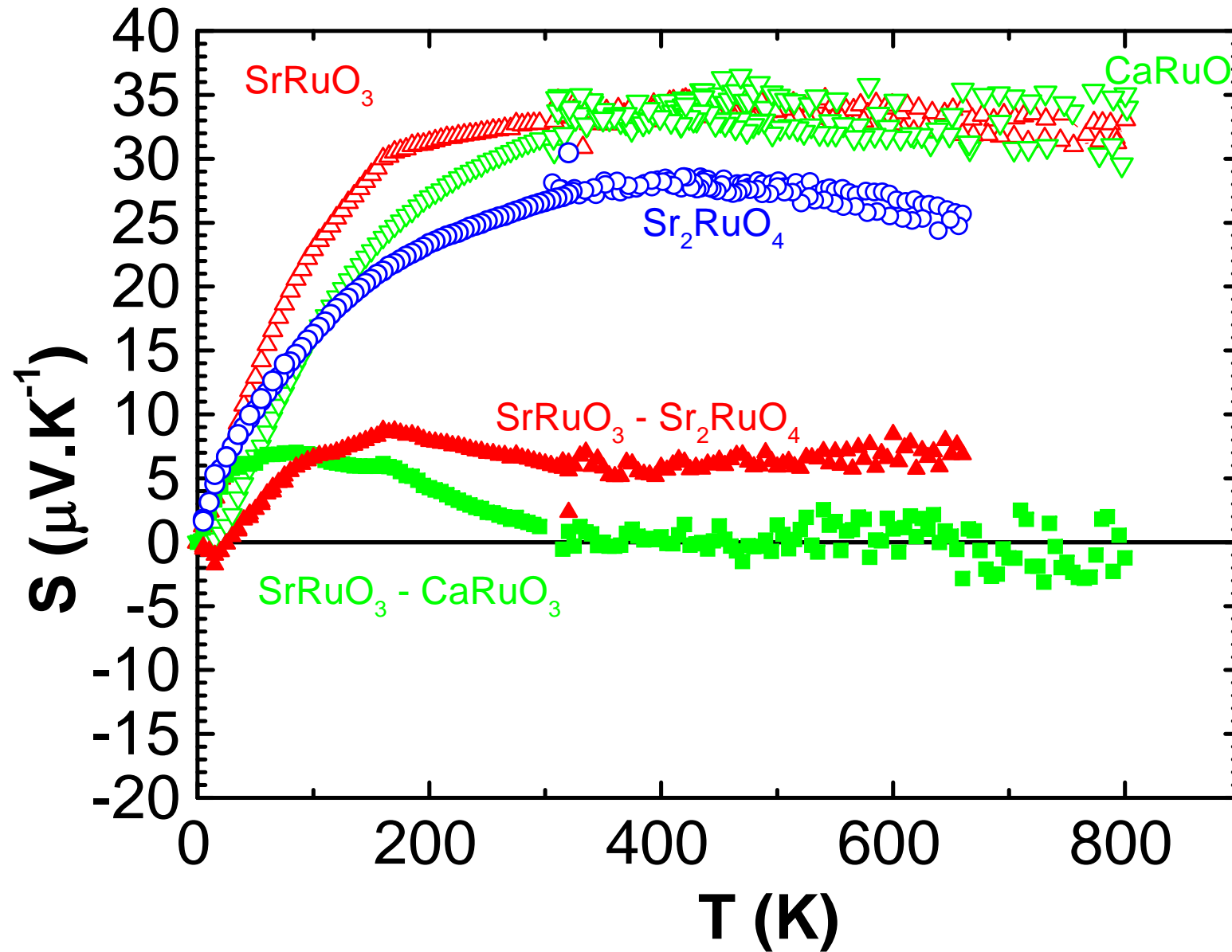
$$S^{LS}_{\text{mag+orb}} = -\frac{k_B}{e} \ln \left(\frac{G_{\text{tot}}^{\text{Ru}^{5+}}}{G_{\text{tot}}^{\text{Ru}^{4+}}} \right) = -69 \mu\text{VK}^{-1}$$

Ru^{3+} (t_{2g}^5 , $S=0.5$, $O=3$)

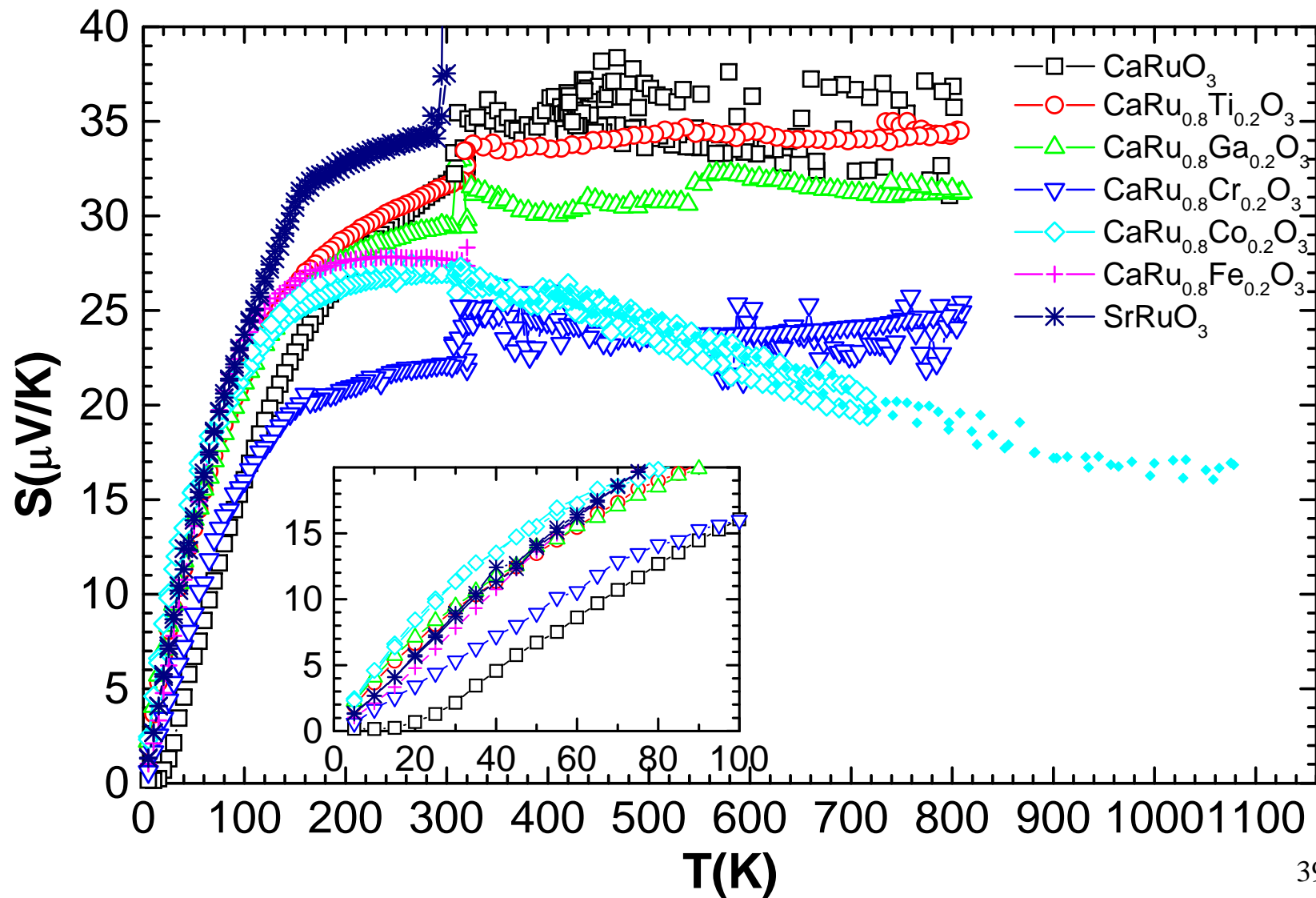
$$S^{LS}_{\text{mag+orb}} = -\frac{k_B}{e} \ln \left(\frac{G_{\text{spin}}^{\text{Ru}^{4+}}}{G_{\text{spin}}^{\text{Ru}^{3+}}} \right) = +35 \mu\text{VK}^{-1}$$

$$G_{\text{spin}}=2, G_{\text{orb}}=3, G_{\text{tot}}= 6$$

Ru-perovskites – CW behaviour (FM, AFM, PM)

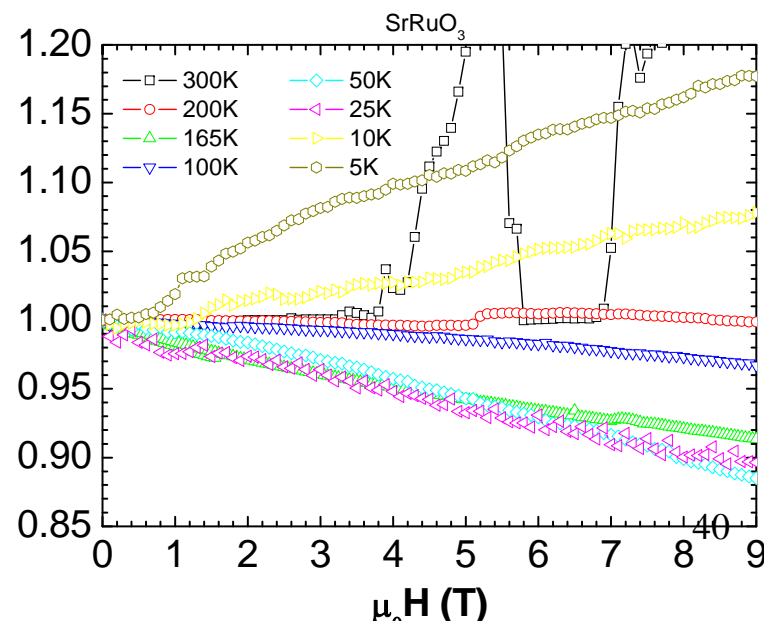
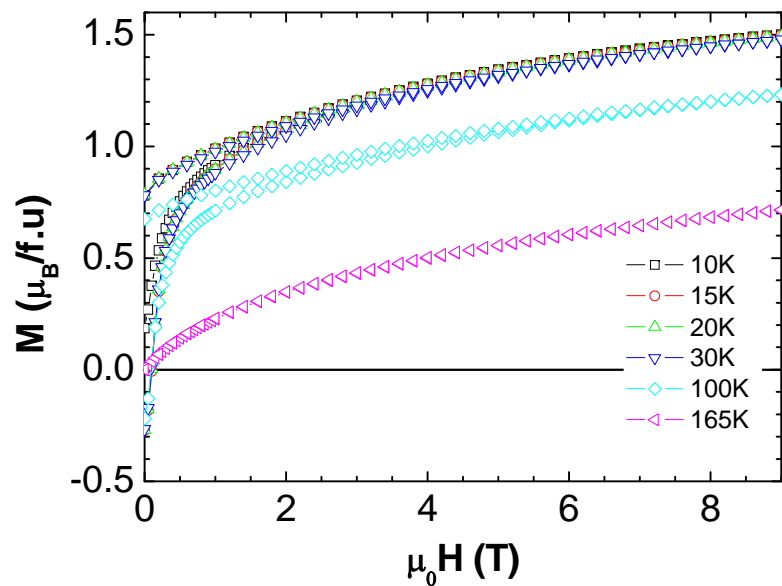
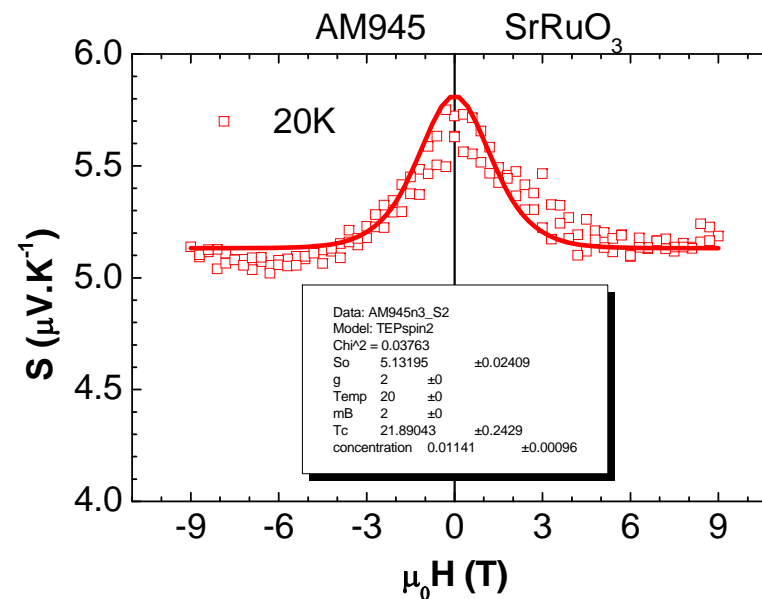
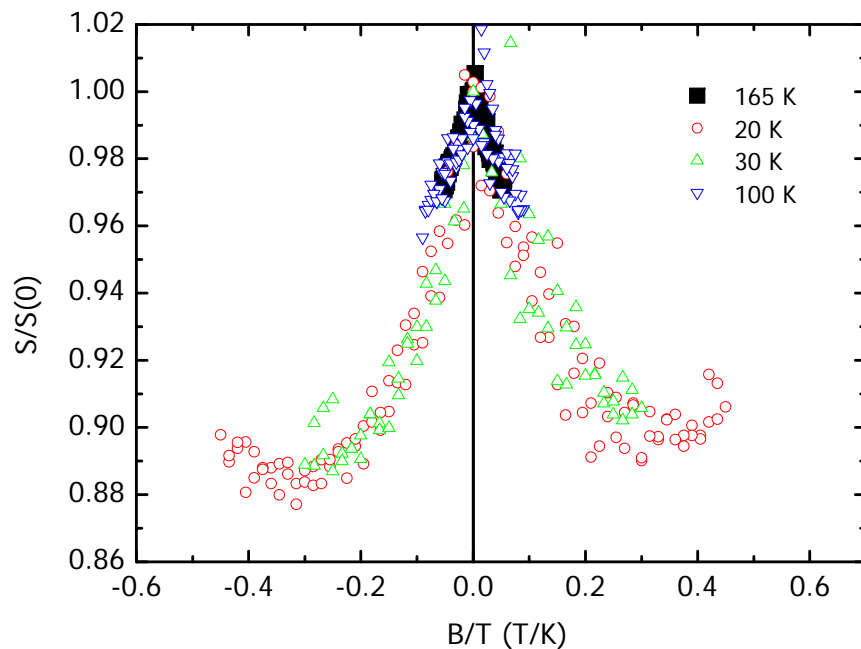


Ru-perovskites –CW behaviour (FM, AFM, PM)



Ru-perovskites – magnetism

SrRuO₃



Properties of metallic ruthenates

(Directly taken or calculated on a base of Reference below)

	CaRuO ₃	Ca _{0.83} Sr _{0.17} RuO ₃	Ca _{0.5} Sr _{0.5} RuO ₃	Ca _{0.25} Sr _{0.75} RuO ₃	SrRuO ₃	BaRuO ₃	Sr ₂ RuO ₄	RuO ₂
μ_{eff} (μ_B)	2.1 2.2	2.4	2.8	2.8	2.8, 2.8	-	4.95	
Θ_C (K)	-57 -68	14	57	82	+170 170	-	-2100	
$\chi_o^{\text{experiment}} * 10^{-4}$ (emu mol ⁻¹ Oe ⁻¹)	6.9 7	29	13	21	8.7 9	<3		1.39
γ (Jmol ⁻¹ K ⁻²)	-74 75	95	60	95	29 30	7.7		5.77
m^*/m_0 ($\sim \gamma/\gamma_{\text{bare}}$)	6 m_0				3.4 m_0			
$\chi_o^{\text{calcul}} = 3\mu_B^2 \gamma/\pi^2 k_B^2$ (emu mol ⁻¹ Oe ⁻¹)*10 ⁻⁴	10	12.8	8.1	12.8	3.9			0.78
$R_W = \pi^2 k_B^2 / 3\mu_B^2 \chi_o^{\text{exp}} / \gamma$	0.7	2.26	1.6	1.6	2.3	<2.9		1.7
$n^{200\text{K}}(\text{cm}^{-3}) * 10^{22}$	+1.5 0.9		+0.9	+0.9	+1.2 +1.8	+0.7	>10 (comp)	5.6
$n^{5\text{K}}(\text{cm}^{-3}) * 10^{22}$	-3		+0.3	+0.3	-1.8			5.0 (77K)
$\rho^{300\text{K}}(\mu\Omega\text{cm})$	~200 200 3000		~600		~200 200 1000	~100	~100 20000	35 330
$S^{800\text{K}}(\mu\text{VK}^{-1})$	32				32		24 (700 K)	10
$S^{300\text{K}}(\mu\text{VK}^{-1})$	32				34		28	0
$S^{20\text{K}}(\mu\text{VK}^{-1})$	0.6				5.0		5.7	2.7
$S/T^{10\text{K}}(\mu\text{VK}^{-2})$	0.0 !!				0.27		0.28	0.12

Reference: J. Appl.Phys, 81(8), 4978, PRB, 51, 16432, PRB,63,R 161102, PRB,56, 321, Phys.Rev.,37,303(1931), PhysRevB, Vol1,1494(1970), our data

SrRuO₃ thin layers (properties at 4 K), APL 82,No.3 ,427,

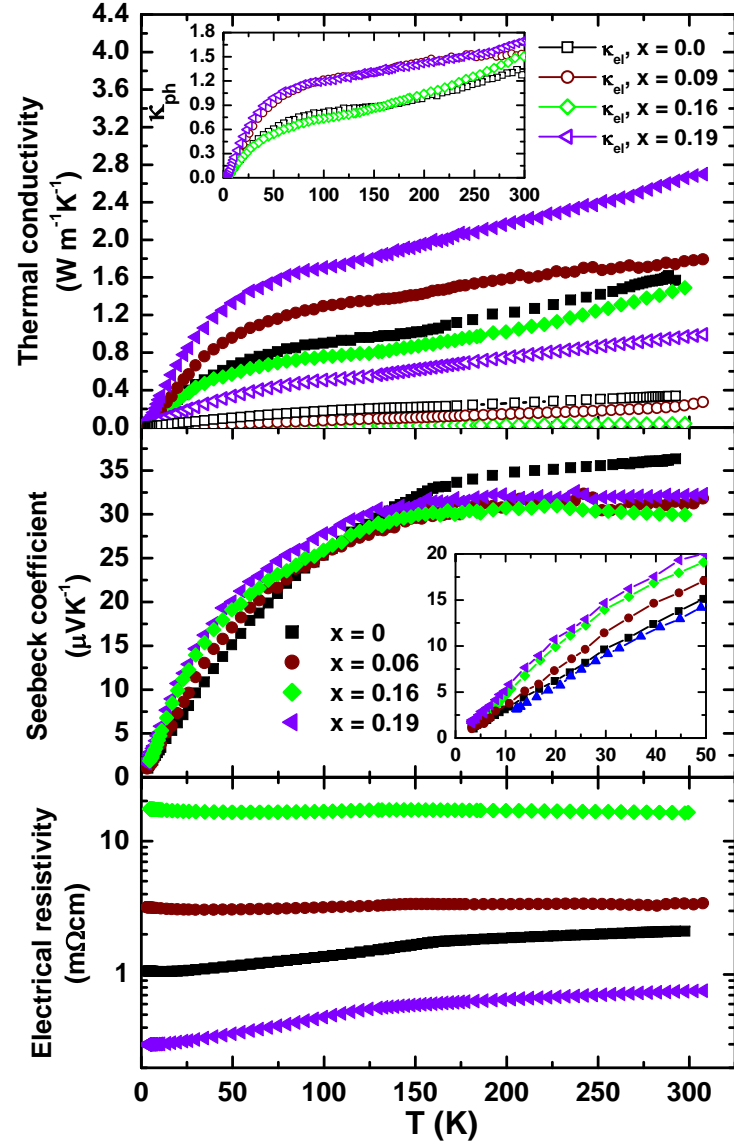
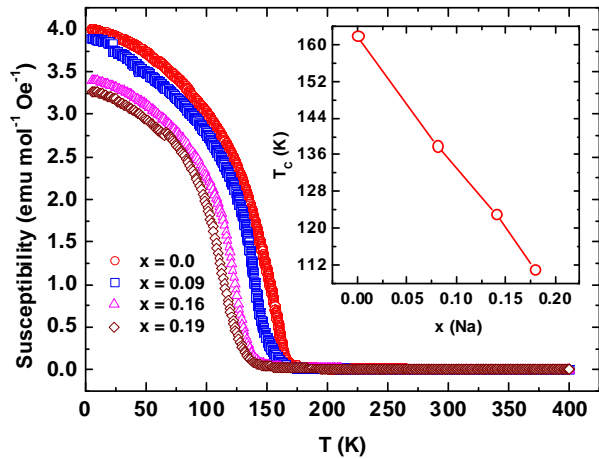
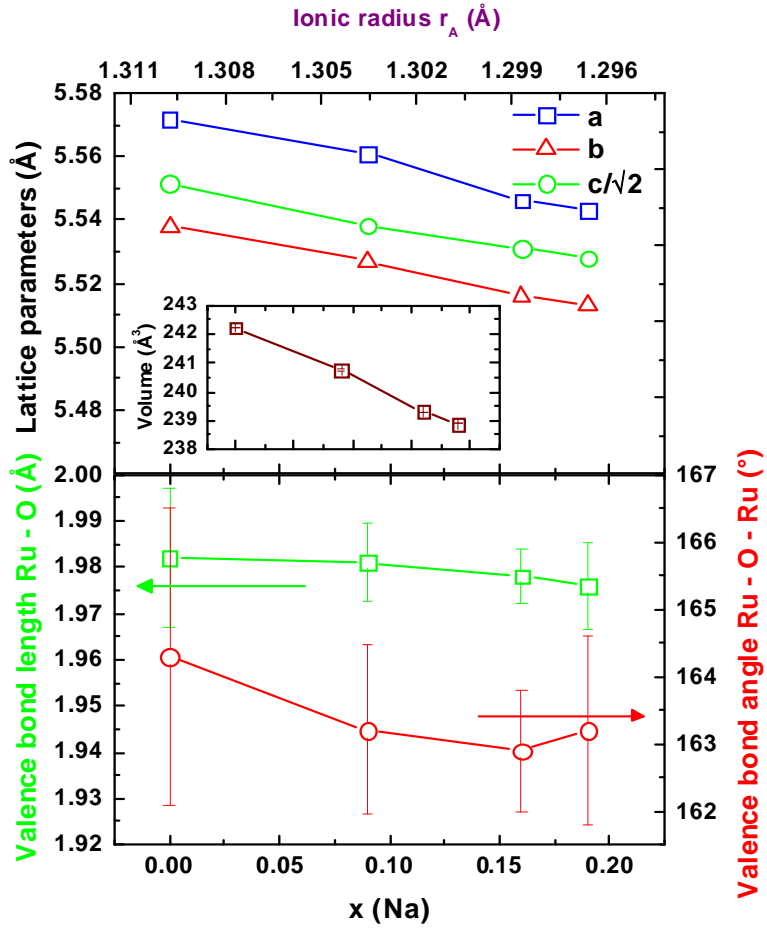
Majority carriers , $l = 45 \text{ \AA}$

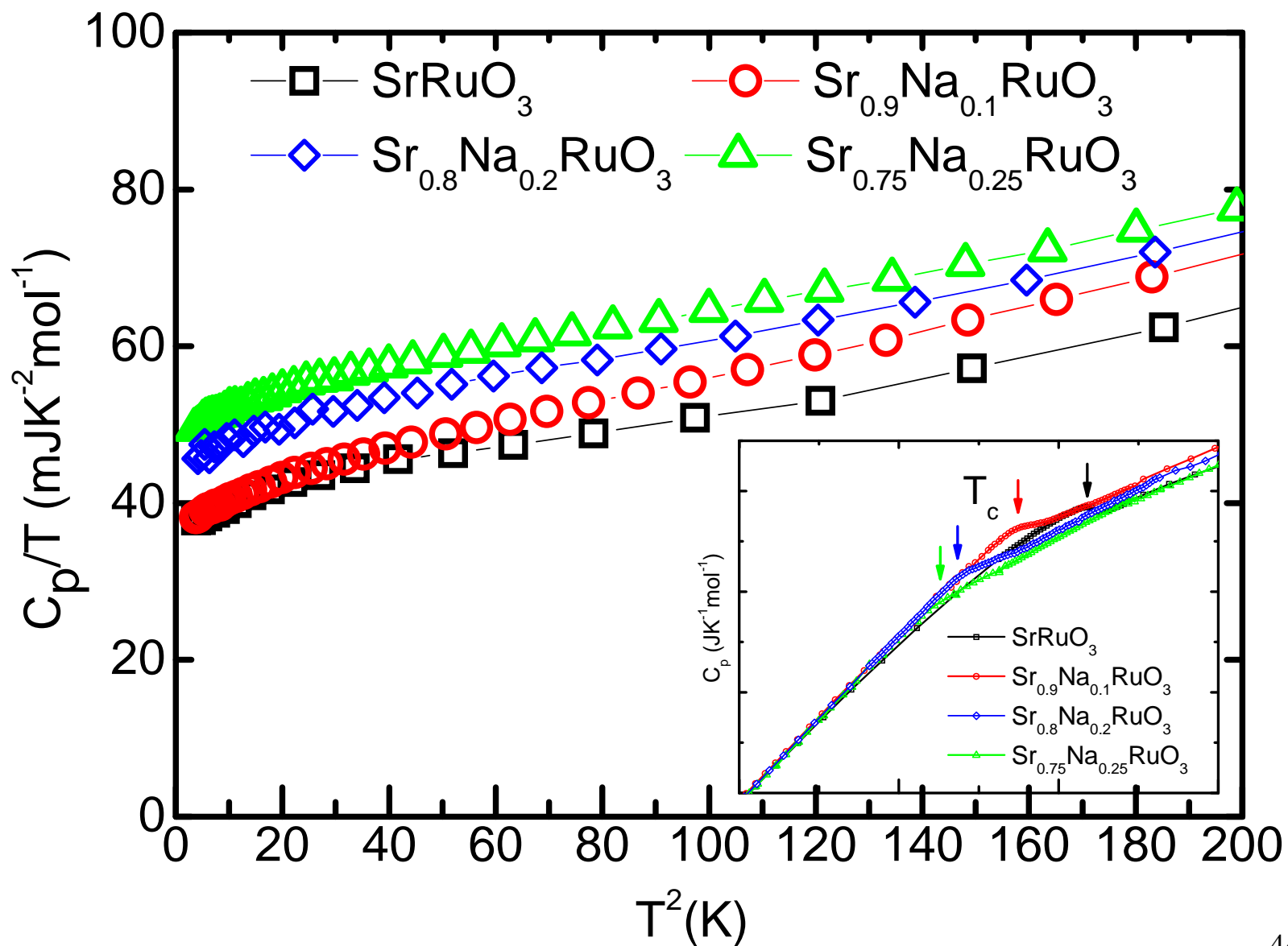
Minority carriers , $l = 80 \text{ \AA}$

Spin polarization ~ 50 % due to different Fermi velocity of \uparrow (spin up) \downarrow (spin down) carriers, NOT DIFFERENT CONCENTRATION

Fermi velocity – majority $\langle |v| \rangle = 0.65 * 10^5 \text{ ms}^{-1}$

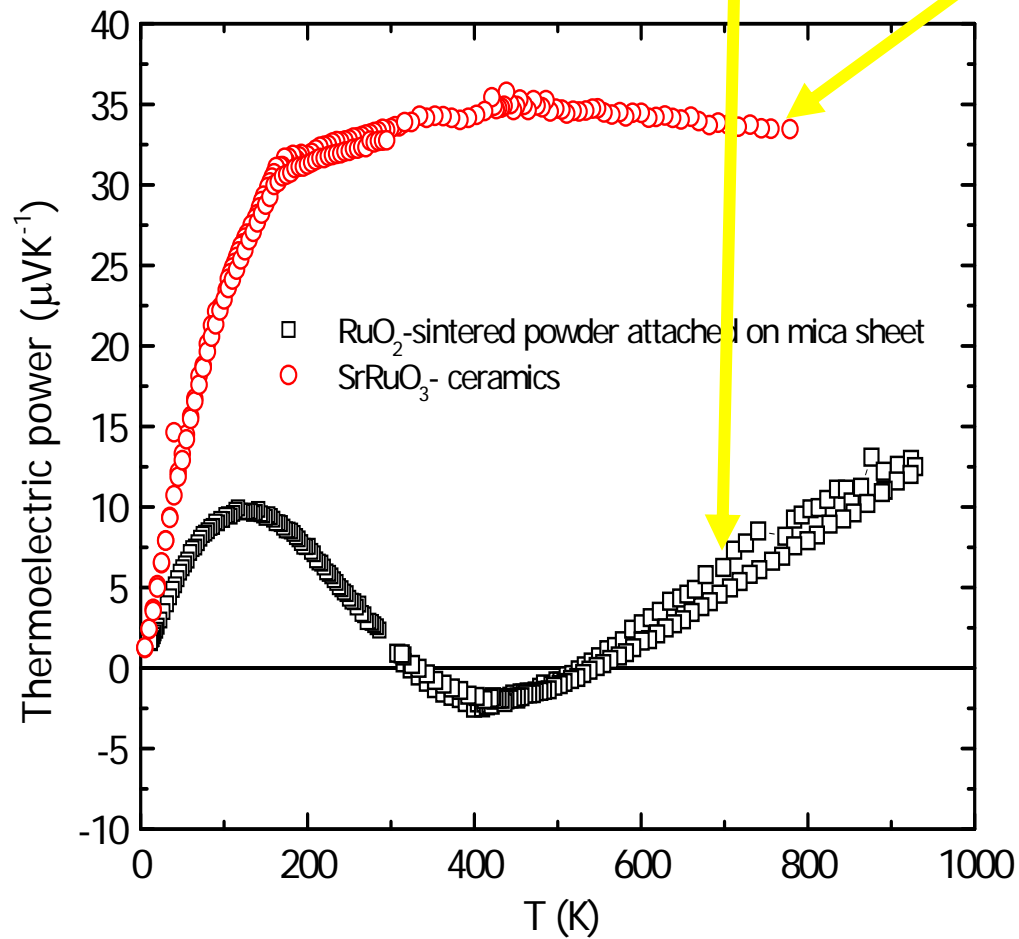
Fermi velocity – minority $\langle |v| \rangle = 1.1 * 10^5 \text{ ms}^{-1}$





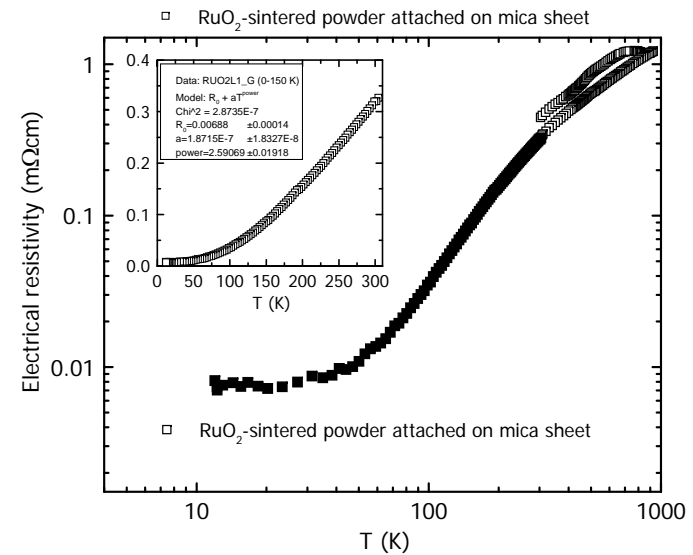
Common oxide metals

RuO₂- Pauli metal vs. SrRuO₃ – CW metal,

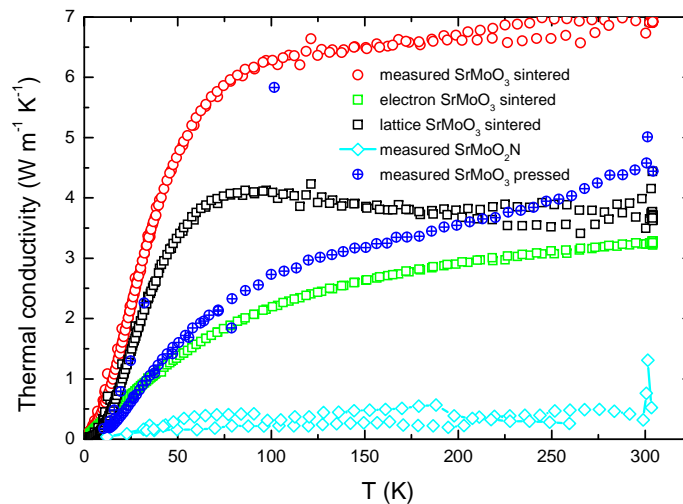
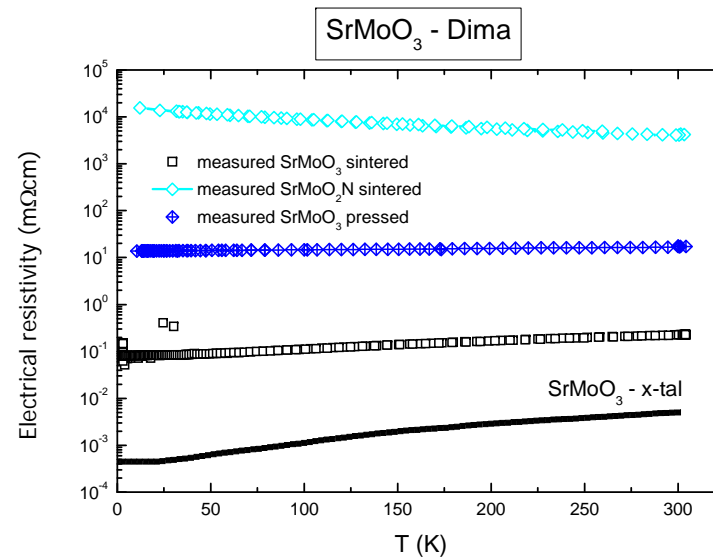
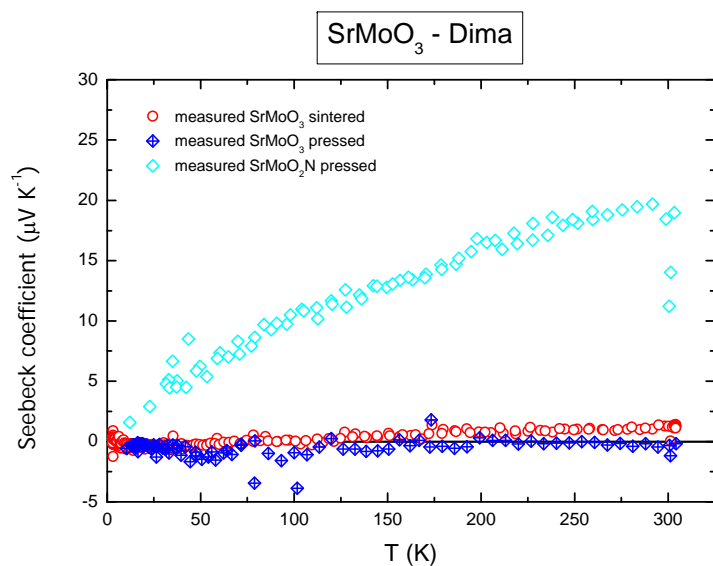


$$S_{mag} = -\frac{k_B}{e} \ln\left(\frac{2S^n + 1}{2S^{n+1} + 1}\right) = +35 \mu\text{VK}^{-1}$$

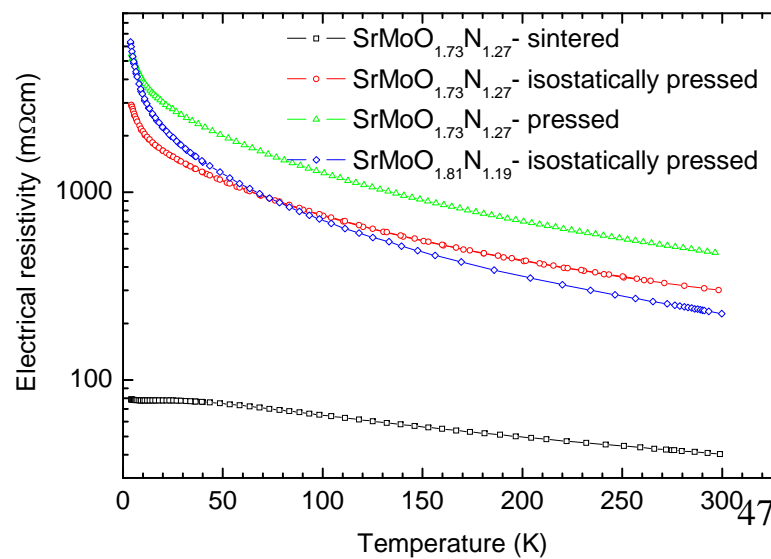
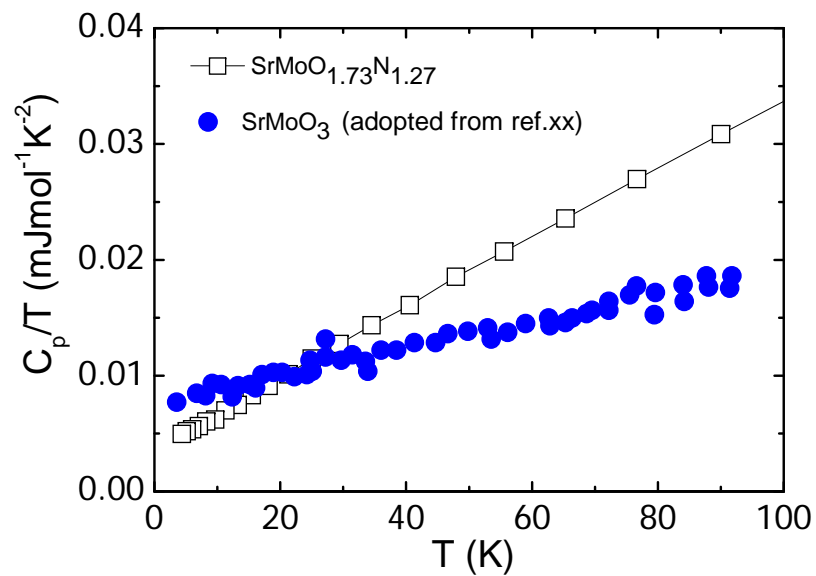
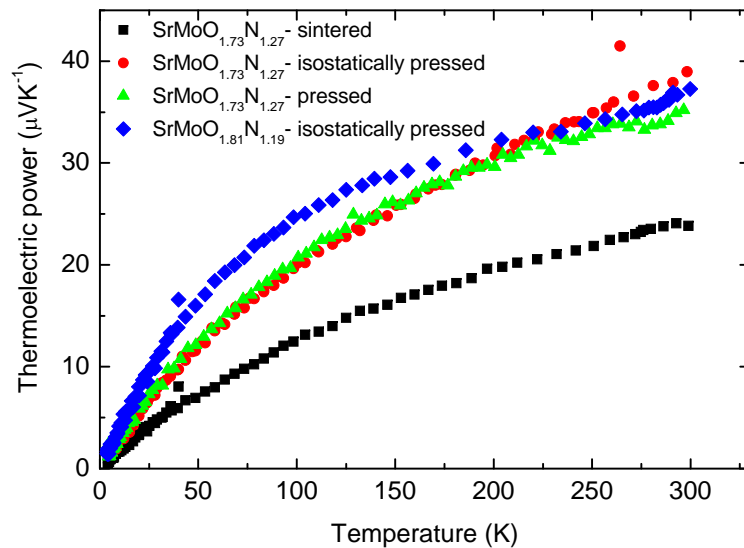
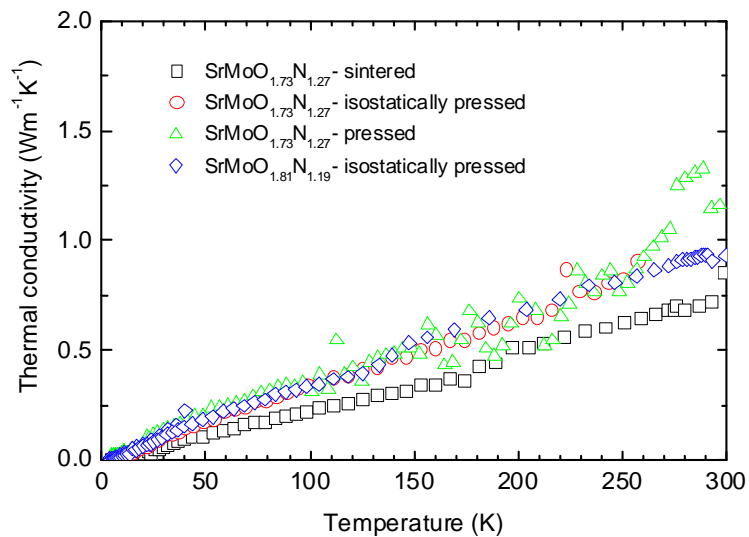
➤ a configurational entropy approximation applies
(magnetic, no orbital contribution) $S_{\text{Heiks}} = +35 \mu\text{VK}^{-1}$



Molybdenates – sintering, synthesis, properties



Molybdenates – oxynitride



Transition metal perovskites and thermoelectricity

- Oxide perovskites represent an interesting class of chemically stable materials with a potential to be used as high-temperature thermoelectrics
- Both diffusive “metallic” (linear or quasilinear in T) or “hopping” (temperature independent) thermopower behaviour is observed in highly electrically conducting perovskites
- Magnetic interactions of conducting electron-holes are likely at the origin of dominance of thermopower configurational entropy character over the diffusive one
- Cr³⁺/Cr⁴⁺ perovskites represent a unique example where the orbital degree of freedom to the configurational entropy applies
- Curie-Weiss magnetic behaviour seems to be the essential prerequisite for the magnetic contribution to the thermopower
- Magnetic and/or spin-state fluctuations are efficient in lowering thermal conductivity as evidenced for Co perovskites

Acknowledgments

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