

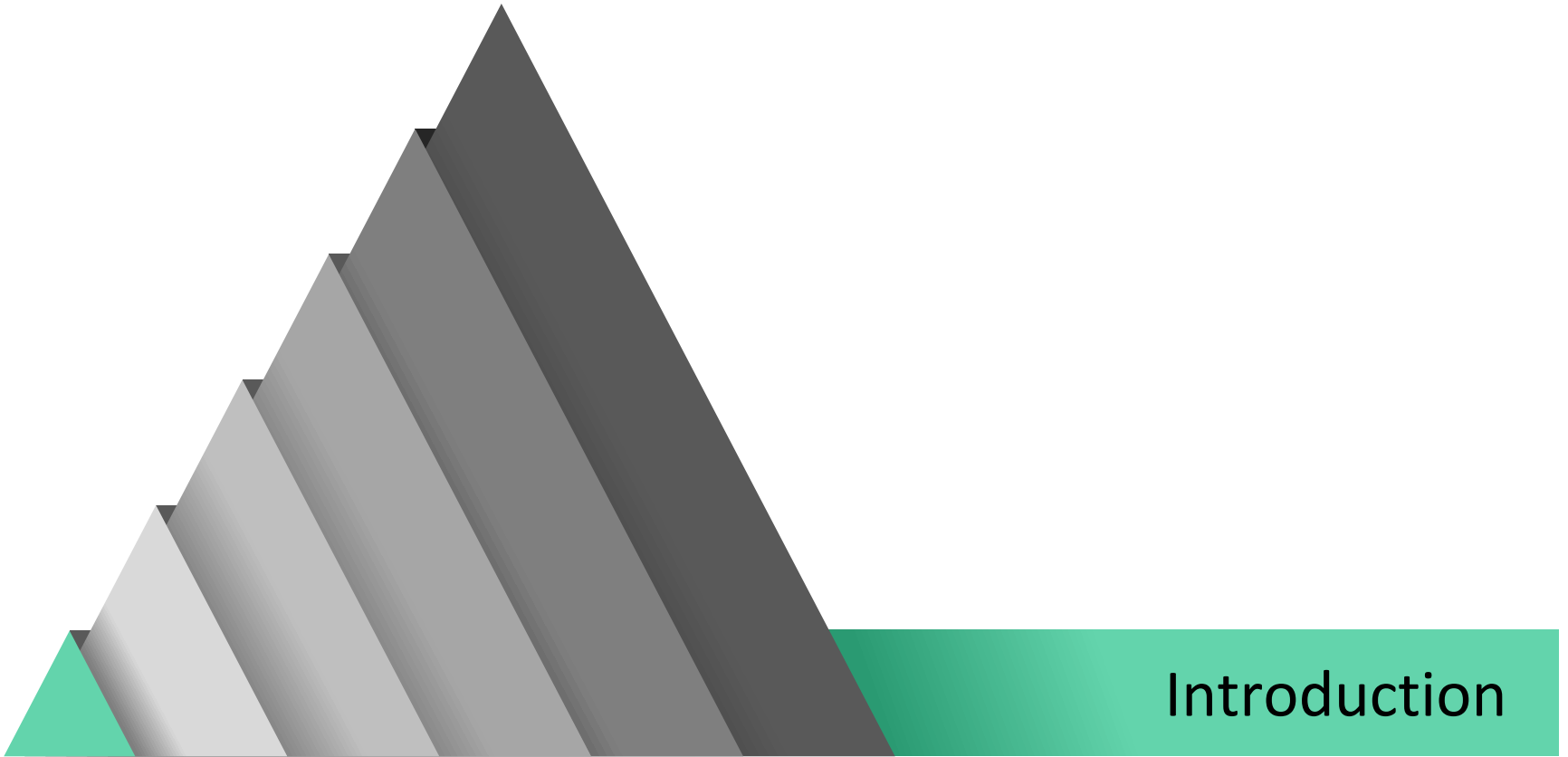


# Thermoelectric properties of oxysulfide $\text{Bi}_{1-x}\text{Pb}_x\text{CuOS}$ compounds

*Jean-Baptiste LABÉGORRE, Laboratoire CRISMAT*  
*PhD supervisor: Emmanuel GUILMEAU*

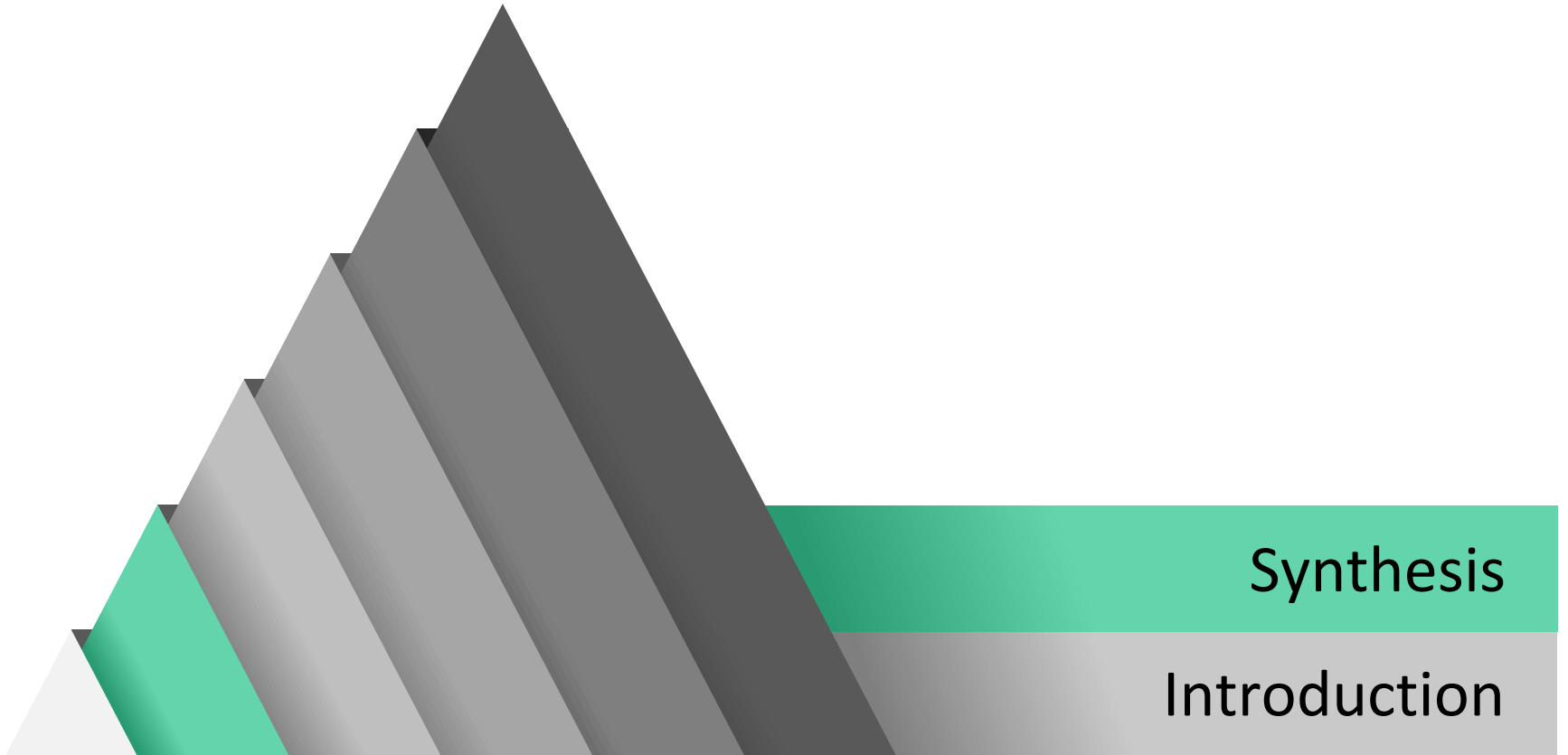
# Plan

---



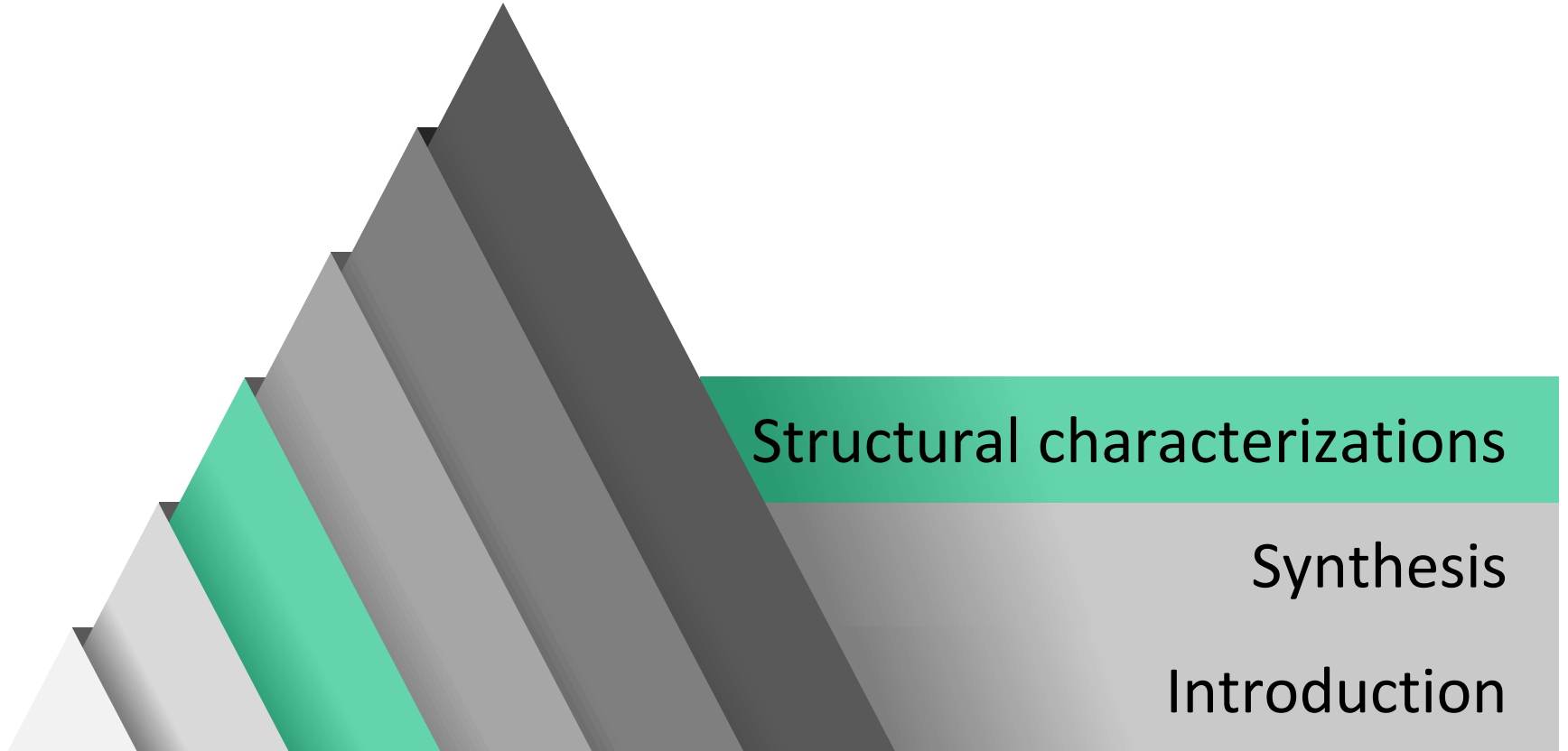
# Plan

---



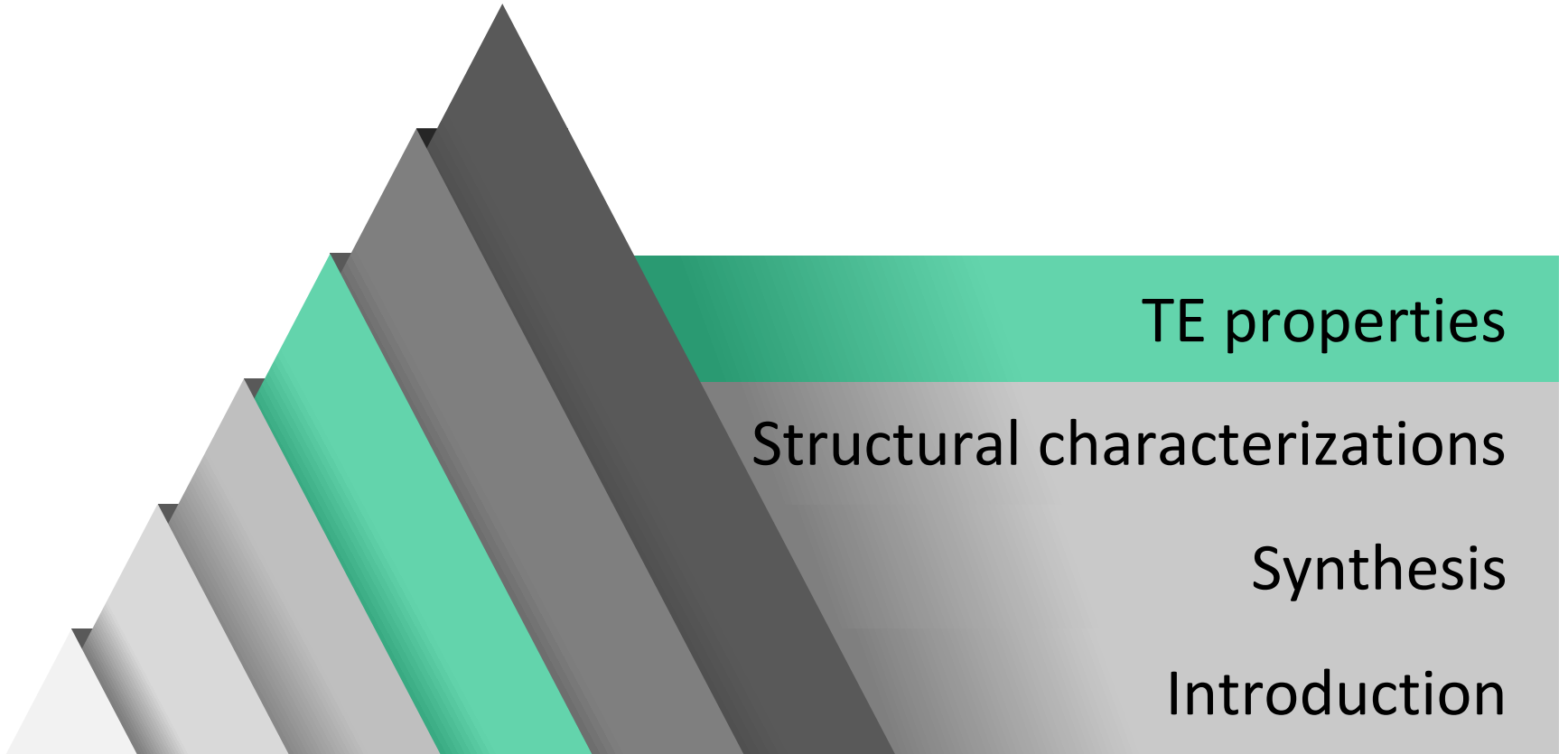
# Plan

---



# Plan

---



# Plan

---



Theoretical calculations

TE properties

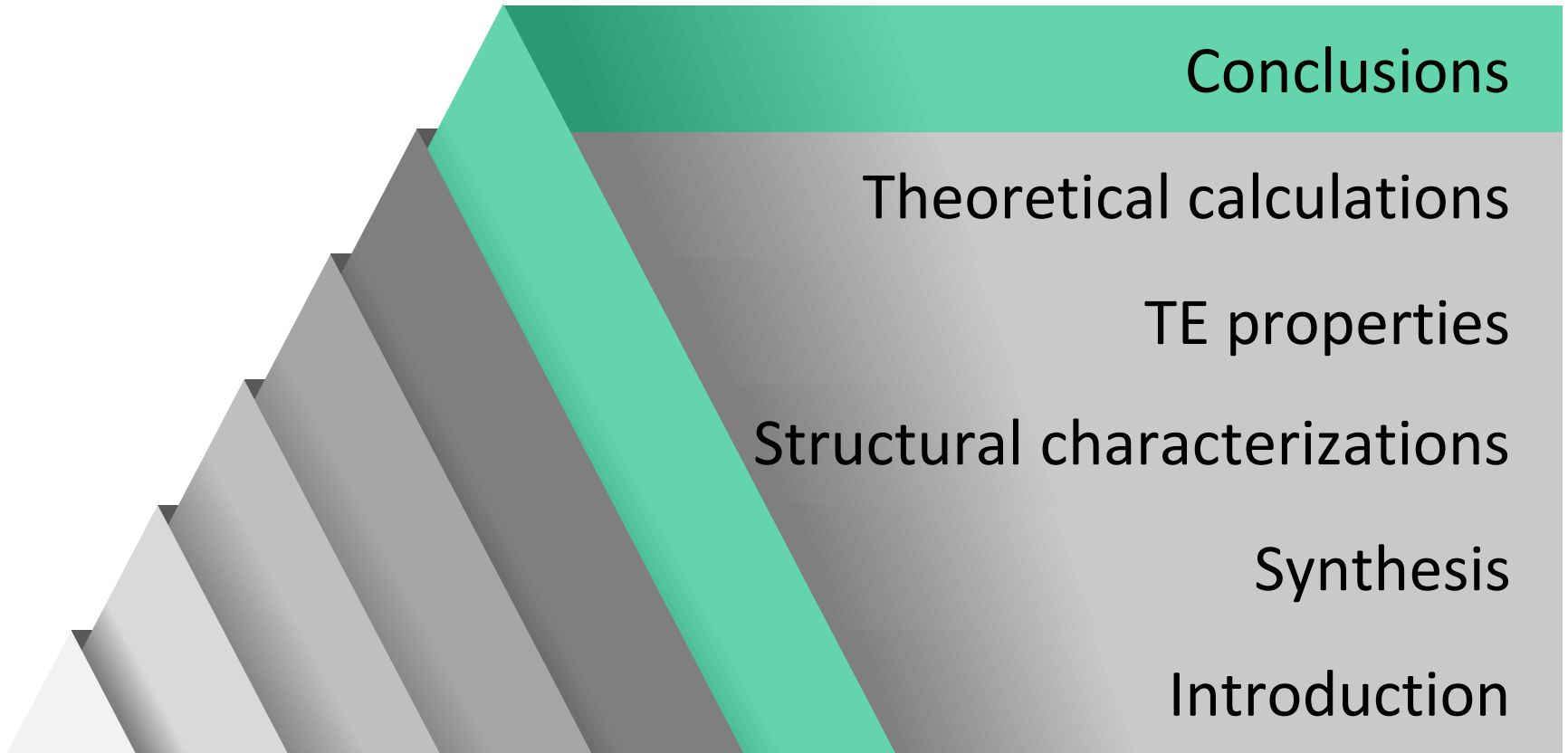
Structural characterizations

Synthesis

Introduction

# Plan

---



# Introduction



discovery of  
 $\text{BiCuOCh}$  (Ch =  
S, Se)<sup>1</sup>

1<sup>st</sup> TE  
study of  
 $\text{BiCuOSe}^2$

> 60 papers  
relative to  
 $\text{BiCuOSe}$

## Structure:

tetragonal,  $\text{ZrSiCuAs}$  type ( $P4/nmm$ )

- 2 layers alternately stacked along  $c$
- covalent bonding inside the layers <sup>(4)</sup>
- ionic bonding between the blocks

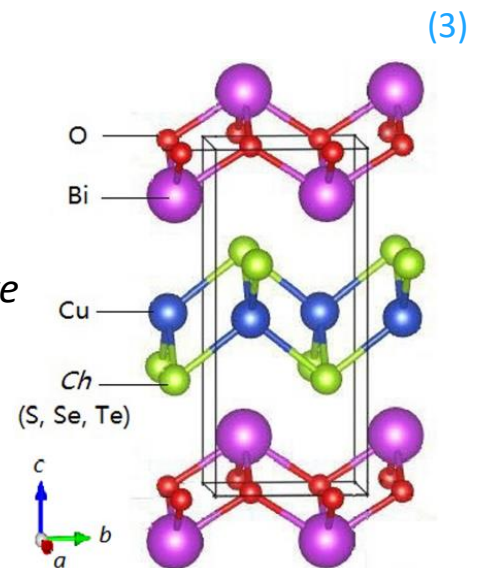
↳ anisotropy of electrical and thermal properties

↳ low thermal conductivity

fluorite-like  
 $(\text{Bi}_2\text{O}_2)^{2+}$

anti-fluorite-like  
 $(\text{Cu}_2\text{Ch}_2)^{2-}$

fluorite-like  
 $(\text{Bi}_2\text{O}_2)^2$



(1) Kusainova, A. M. *et al. J. Solid State Chem.* **1994**, *112*, 189–191.

(3) Liu, G. *et al. J. Appl. Phys.* **2016**, *119*, 185109.

(2) Zhao, L. D. *et al. Appl. Phys. Lett.* **2010**, *97*, 092118.

(4) Shein, I. R. *et al. Solid State Commun.* **2010**, *150* (13-14), 640–643. 3



# Introduction

BiCuOCh (Ch = S, Se, Te) → copper vacancies → p-type<sup>(1)</sup>

(3,4)

	BiCuOSe	BiCuOS
gap (eV)	0.8	1.1
$\rho_{RT}$ (m $\Omega$ cm)	~ 200	~ 8 x 10 <sup>5</sup>
$\kappa_{RT}$ (W m <sup>-1</sup> K <sup>-1</sup> )	1.05	1.1
ZT <sub>max</sub> (673 K)	<b>0.31</b>	<b>0.07</b>

## TE properties of BiCuOSe:

- stable in medium temperature range
- numerous successful substitutions : Ba<sup>2+</sup>, Sr<sup>2+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup>, Pb<sup>2+</sup>...
- when substituted and textured : ZT<sub>max</sub> = 1.4 @ 923 K<sup>(2)</sup>

→ Lack of studies on BiCuOS due to high resistivity

(1) Ueda, K. *et al. Thin Solid Films* **2002**, 411, 115–118.

(2) Sui, J. *et al. Energy Environ. Sci.* **2013**, 6, 2916–2920.

(3) Bérardan, D. *et al. Materials.* **2015**, 8, 1043–1058.

(4) Zhu, H. *et al. J. Eur. Ceram. Soc.* **2017**, 37, 1541–1546.

# Synthesis



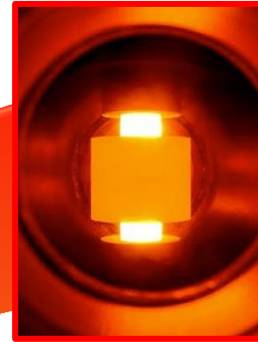
$\text{Bi}_2\text{O}_3$ , Bi, Cu,  
S, (PbO)



**Mechanosynthesis**

- synthesized in 3 h vs sealed tubes (> 10 h) or hydrothermal synthesis (> 55 h) <sup>(1,2)</sup>

- scalable for mass production

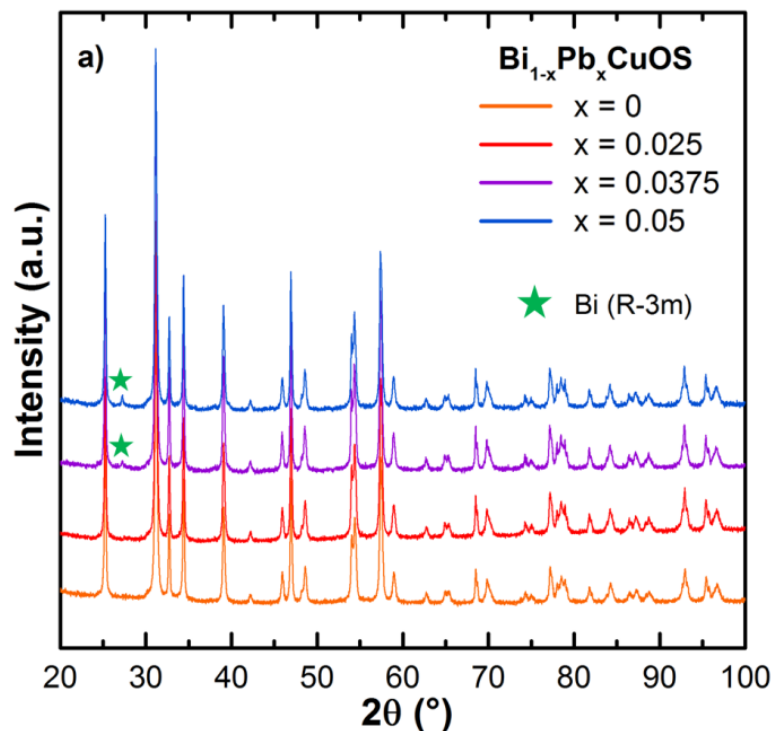


**Spark Plasma Sintering**

- short process duration, high density

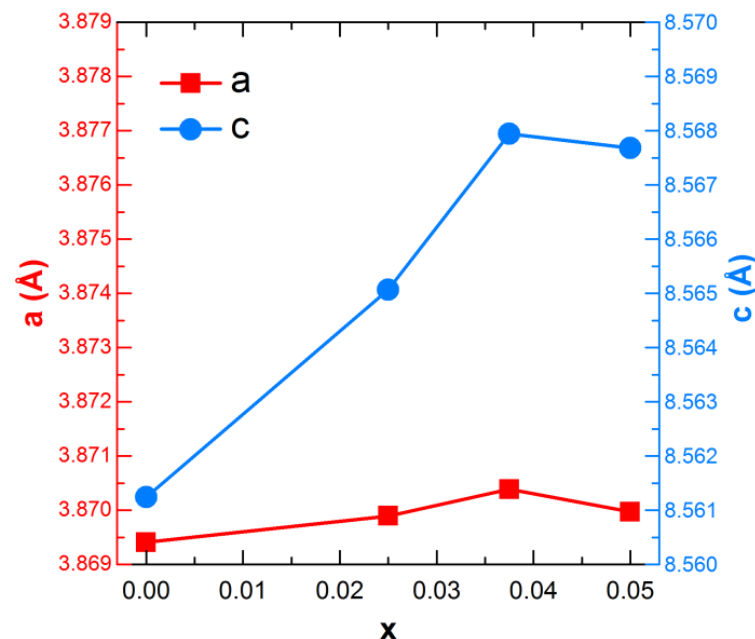
4'45'' / 475 °C 25 min / 4'45''  
pressure : 64 MPa

# Structural characterizations of $\text{Bi}_{1-x}\text{Pb}_x\text{CuOS}$ ( $0 \leq x \leq 0.05$ )



## powder XRD after sintering :

- $x \leq 0.025$  : single phase
- $x \geq 0.0375$  : presence of ~ 1 % Bi metal

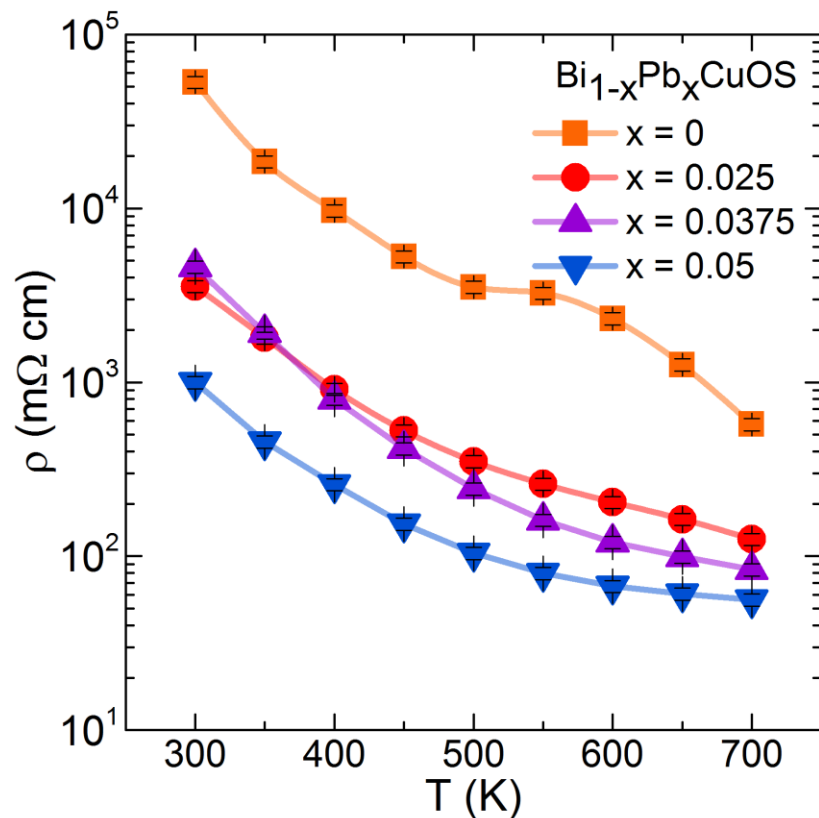


## cell parameters (from Rietveld refinements) :

when  $x \nearrow$ :

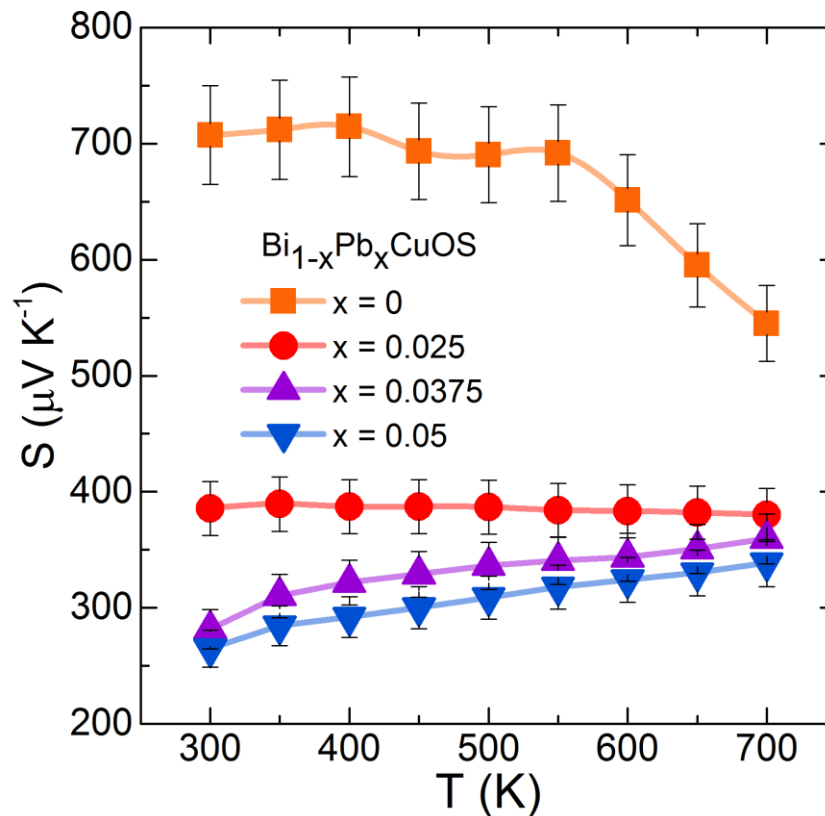
- $a$  and  $c \nearrow \rightarrow$  substitution of  $\text{Pb}^{2+}$  (129 pm) on  $\text{Bi}^{3+}$  (117 pm) site
- $c$  increases more than  $a \rightarrow$  weakening of Coulombic attraction between the layers  $[(\text{Bi}_{1-x}\text{Pb}_x)_2\text{O}_2]^{(2-2x)+}$  and  $[\text{Cu}_2\text{S}_2]^{(2-2x)-}$  (1)

# TE properties of $\text{Bi}_{1-x}\text{Pb}_x\text{CuOS}$ ( $0 \leq x \leq 0.05$ )



- when  $x \nearrow$  :  $\rho \searrow$

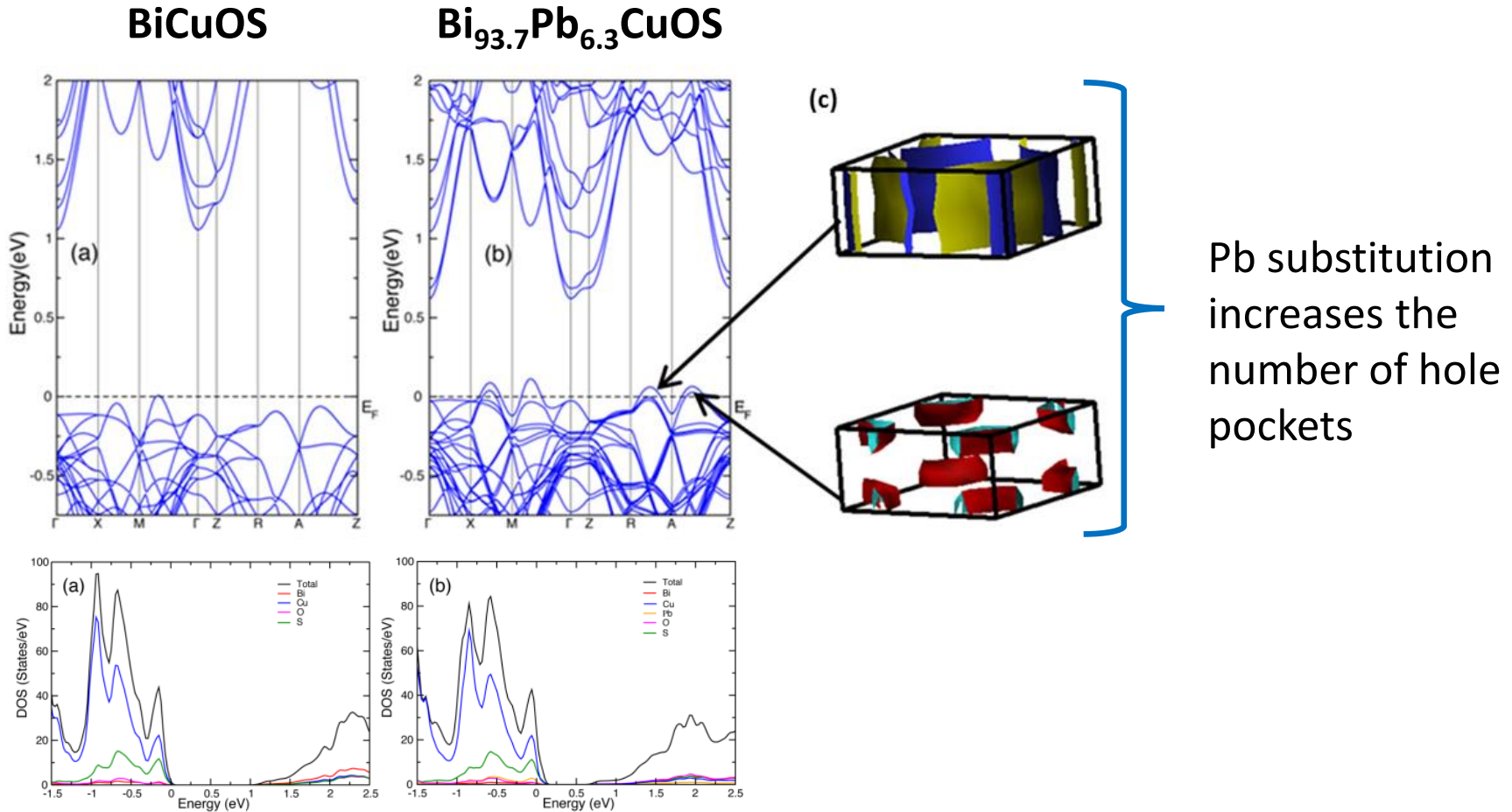
•  $\rho_{x=0.05} = 56 \text{ m}\Omega \text{ cm @ 700 K}$



- when  $x \nearrow$  :  $S \searrow$

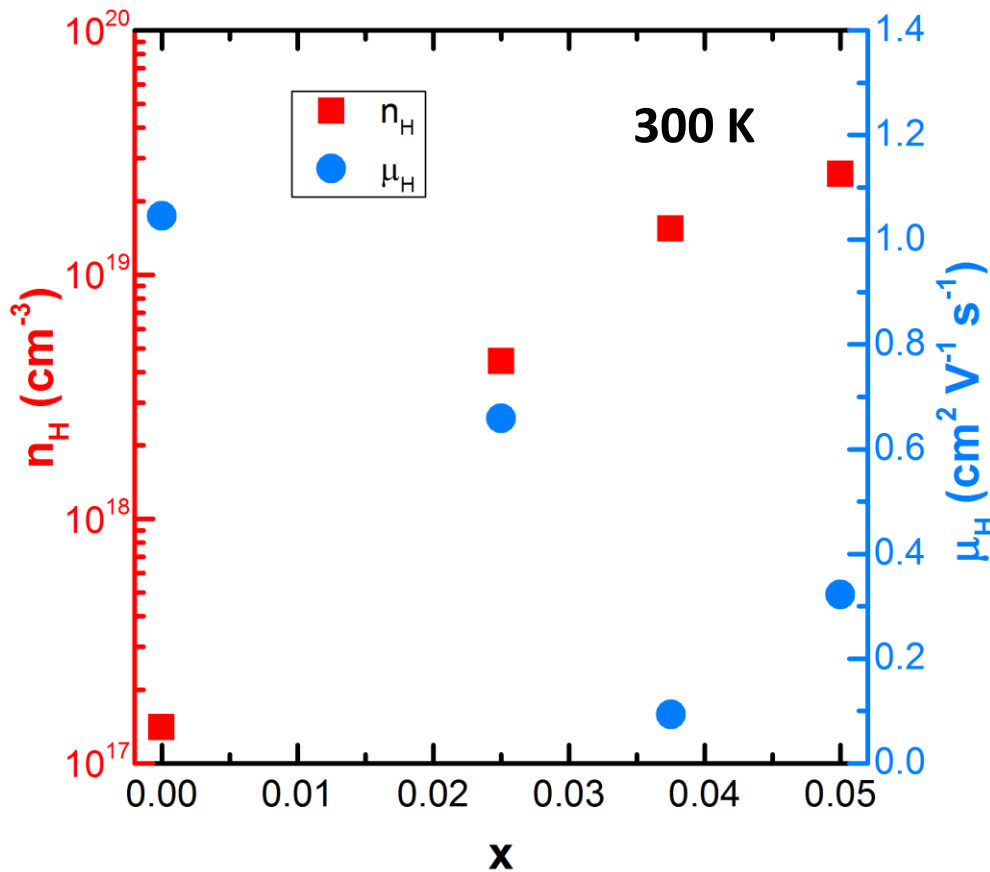
•  $S_{x=0.05} = 340 \text{ }\mu\text{V K}^{-1} \text{ @ 700 K}$

# DFT calculations



- band-gap decreases with Pb-doping
- top of VB : - hybridization between Cu 3*d* and S 3*p*  
- contribution of Pb 6*s* in Pb-doped samples

# TE properties of $\text{Bi}_{1-x}\text{Pb}_x\text{CuOS}$ ( $0 \leq x \leq 0.05$ )



when  $x \nearrow$  :

- $n \nearrow \rightarrow$  confirms  $\text{Pb}^{2+}$  substitution
- $n$  reaches  $2.6 \times 10^{19} \text{ cm}^{-3}$  ( $x = 0.05$ )

too low to reach optimal PF

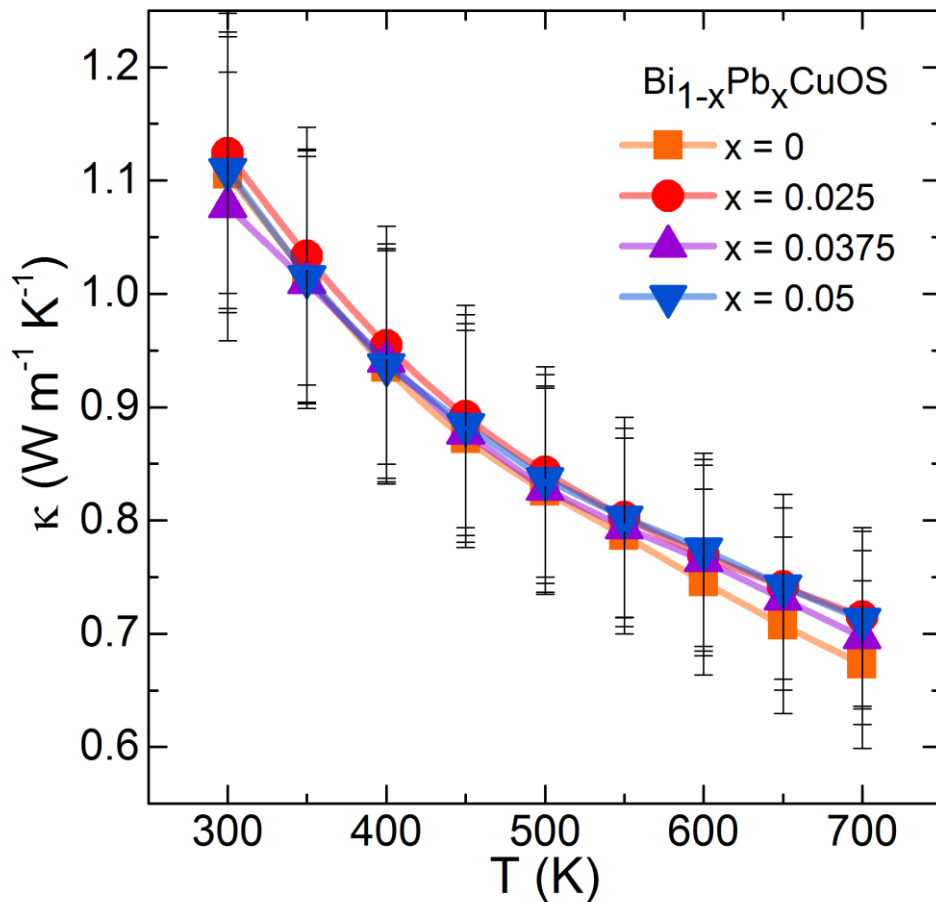
- $\mu \searrow$  until  $0.3 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$  ( $x = 0.05$ )

low mobility

hole effective mass estimated from Seebeck coefficient and the carrier concentration :

x	0	0.025	0.0375	0.05
$m^*/m_e$	1.71	1.81	1.85	2.2

# TE properties of $\text{Bi}_{1-x}\text{Pb}_x\text{CuOS}$ ( $0 \leq x \leq 0.05$ )



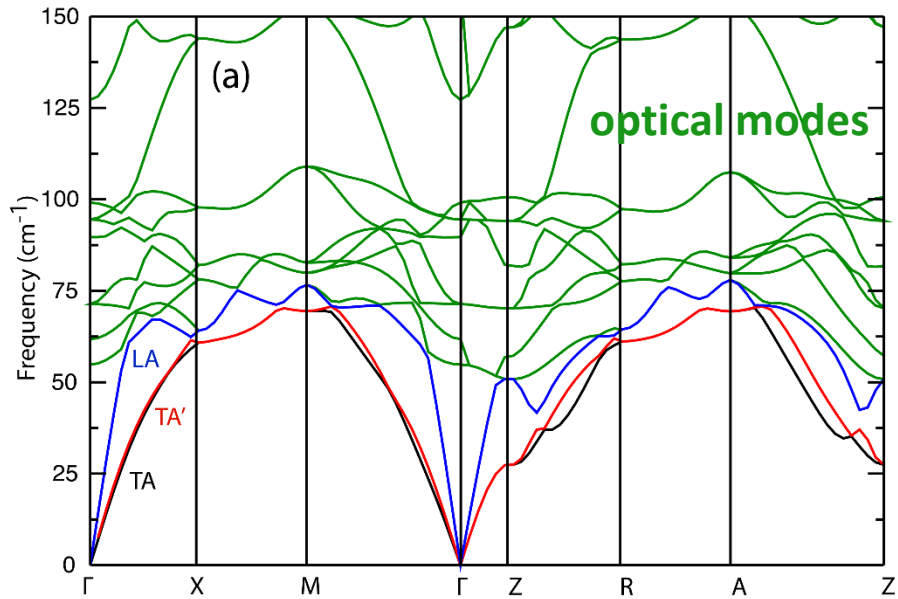
- when  $x \nearrow$  :  $\kappa$  remains constant  
Bi/Pb: - low mass fluctuation  
- moderate size mismatch

- low  $\kappa$  ( $\kappa_l \gg \kappa_e$ )  $\rightarrow$  layered structure, strong phonon scattering



$$\kappa_{700\text{ K}} \sim 0.70 \text{ W m}^{-1} \text{ K}^{-1}$$

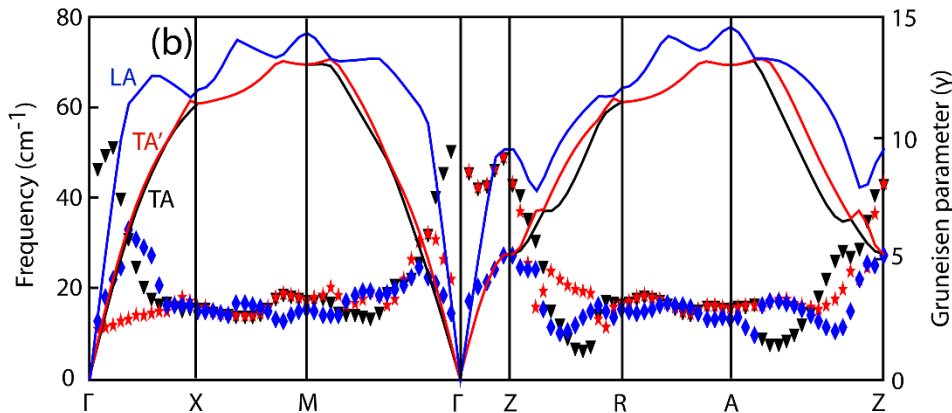
# Thermal conductivity of BiCuOS



## phonon dispersion of BiCuOS:

- 3 acoustic modes with low boundary frequency  
65  $\text{cm}^{-1}$  along  $\Gamma$ -X and  $\Gamma$ -M  
50  $\text{cm}^{-1}$  along  $\Gamma$ -Z (001 direction)

↳ weak inter-layers bonding



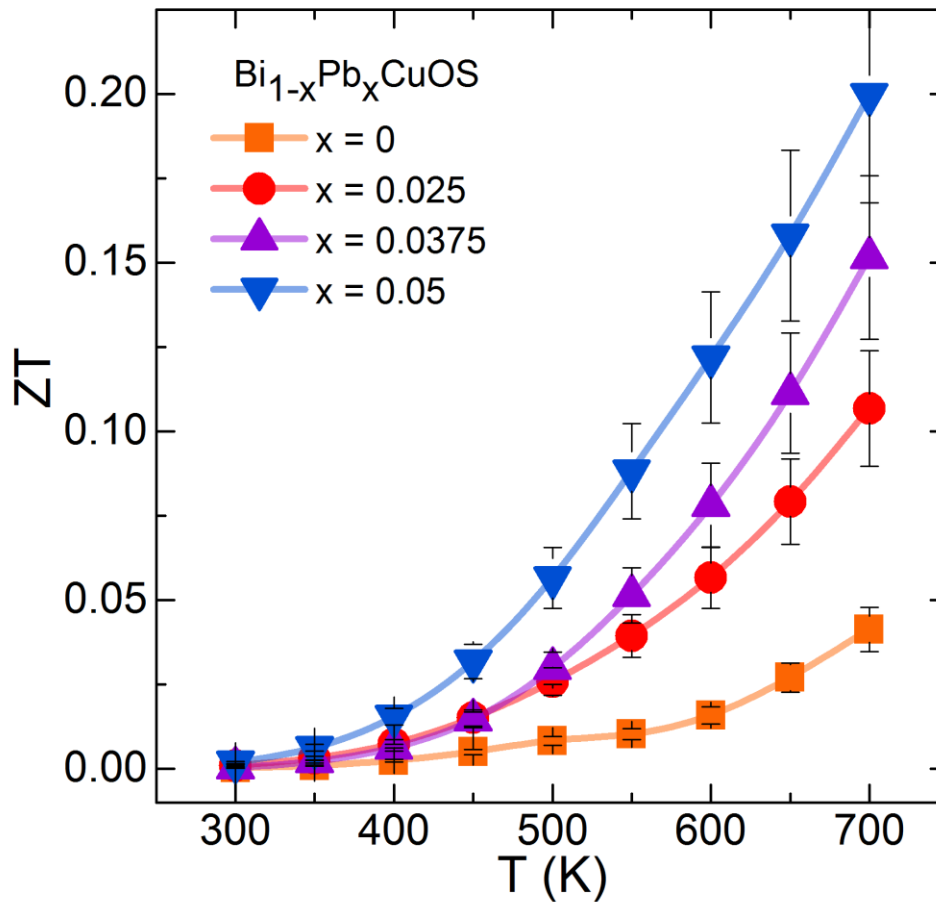
## Grüneisen parameters of BiCuOS :

- ~ 8 along  $\Gamma$ -Z

↳ strong anharmonicity between the layers



# TE properties of $\text{Bi}_{1-x}\text{Pb}_x\text{CuOS}$ ( $0 \leq x \leq 0.05$ )



- when  $x \nearrow$  :  $n \nearrow$  and  $\rho \searrow \rightarrow ZT \nearrow$

- $ZT = 0.2$  @ 700 K with 5 at% Pb  $\rightarrow$

Best value among the oxysulfides

# Conclusions

---

XRD + charge carrier concentration  $\longrightarrow$  confirms  $\text{Pb}^{2+}$  substitution



~ 1 % Bi secondary phase when  $x \geq 0.0375$

when  $x = 0.05$  : charge carrier concentration

x 185 at RT



resistivity

divided by 53 at RT



**remains too high to reach optimal PF**



low thermal conductivity



**ZT = 0.2 @ 700 K with 5 at% Pb**

# Acknowledgements

---



*Emmanuel Guilmeau*

*Rabih Al Rahal Al Orabi*

*Jacinthe Gamon*

*Philippe Barboux*

*Tristan Barbier*

*David Berthebaud*

*Antoine Maignan*

*Thierry Le Mercier (Solvay RIC Paris)*

*Lauriane d'Alençon (Solvay RIC Paris)*

*Solvay for the financial support of this study*