

# **Poudres nanocristallines de $\text{Bi}_2\text{Se}_3$ synthétisées à partir de $\text{H}_2\text{Se}$ généré *in-situ* par voie électrochimique**

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# Outline

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**I. Introduction**

**II. Synthesis of  $\text{Bi}_2\text{Se}_3$  nanocrystalline powders**

**III. Structural and microstructural properties**

**IV. Thermopower of “nanostructured bulk”  $\text{Bi}_2\text{Se}_3$**

**V. Conclusions**

$\text{Bi}_2\text{Te}_3$  and  $\text{Bi}_x\text{Sb}_{2-x}\text{Te}_{3-y}\text{Se}_y$  alloys : Best bulk thermoelectric (TE) materials at room temperature since 1960 ( $ZT \sim 1$  at 300 K)

**BUT small progress with time**

1990's: Low dimensionality effects improve the thermoelectrical properties

Theoretical studies:  $ZT_{1D} > ZT_{2D} > ZT_{3D}$  Dresselhaus et al. Phys Rev B 1993

Experimental work:

1D:  $\text{Bi}_x\text{Sb}_{2-x}\text{Te}_{3-y}\text{Se}_y$  nanowire arrays, **BUT no evidence of ZT improvement yet**  
*Stacey et al. JACS 2001 Nano Lett. 2005, Fleurial et al. Adv. Mater. 2005*

2D:  $\text{Bi}_2\text{Te}_3/\text{Sb}_2\text{Te}_3$  superlattice structures:  $ZT \sim 2.4$  at 300 K, **BUT high cost technology**  
*Venkatasubramanian et al. Nature 2001*

## Nanostructured 3D materials (from nanocrystalline powders)

Low dimensionality effects

Conventional technology

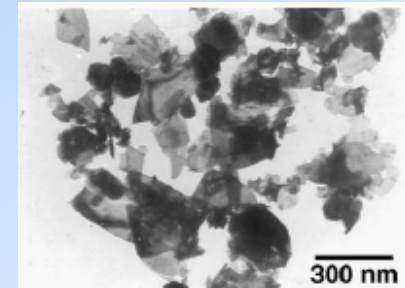
Large-scale production

Recent promising news:  $\text{Bi}_x\text{Sb}_{2-x}\text{Te}_{3-y}$   $ZT \sim 1.4$  at 373 K,  $\text{Bi}_2\text{Te}_3/\text{Sb}_2\text{Te}_3$   $ZT \sim 1.5$  at 440 K  
*Poudel et al. Nature 2008*      *Zhao et al. Appl. Phys. Lett. 2008*

## Synthesis of $\text{Bi}_2\text{X}_3$ (X= Te, Se) nanocrystals

### Solution techniques (low temp and cost):

- **Solvothermal** *Zhao et al. J. Alloys and Compounds 2005*
- **Sonochemistry** *Qiu et al. Inorg. Chem. Commun. 2004*
- **Micellar synthesis** *Foos et al. Nano Lett. 2001*
- etc.



**Produced amounts are generally low (~ mg)**

**Considerable amounts of nanocrystalline (NC) powders are required for nanostructured 3D TE materials !**

**Bubbling gas precursors into the solution:**

$\text{H}_2\text{X}$  (X = Se, Te) as precursors: easily up-scalable  
(~ grams of CdTe NC powders *Rogah et al. J. Phys. Chem. C 2007*)

**$\text{H}_2\text{X}$  are toxic and present poor stability : storage problems!**

**Our proposal: Electrochemical generation *in-situ*!**

## Experimental set-up

### 1. Electrochemical cell:

- Aqueous electrolyte ( $[\text{H}_2\text{SO}_4] = 0.5 \text{ M}$ )
- Cathode: graphite-Se ingot
- Anode: Platinum wire
- $I = -0.1 \text{ A}$

### 2. Chemical cells:

- Aqueous solution
- $[\text{Bi}(\text{Ac})_3] = 1 \times 10^{-2} \text{ M}$
- $[\text{HClO}_4] = 1 - 10 \text{ M}$
- $[\text{X-100 Triton}] = 5 \times 10^{-4} \text{ M}$  (surfactant)

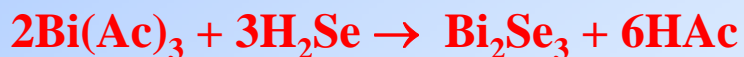
### 1. Electrochemical generation of $\text{H}_2\text{Se}$



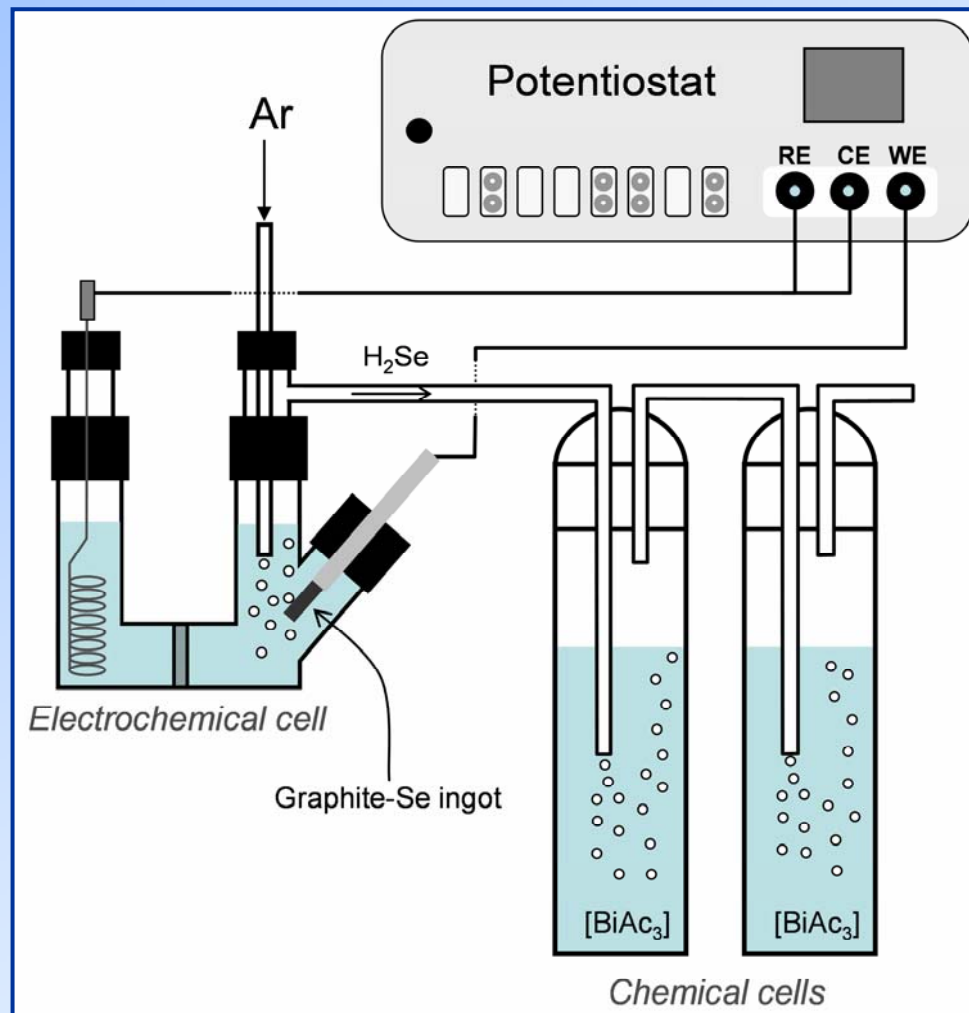
*Bastide et al. J. Electrochem. Soc. 2005*

### 2. Transport of $\text{H}_2\text{Se}$ (Ar: carrier gas)

### 3. $\text{Bi}_2\text{Se}_3$ formation (global reaction):



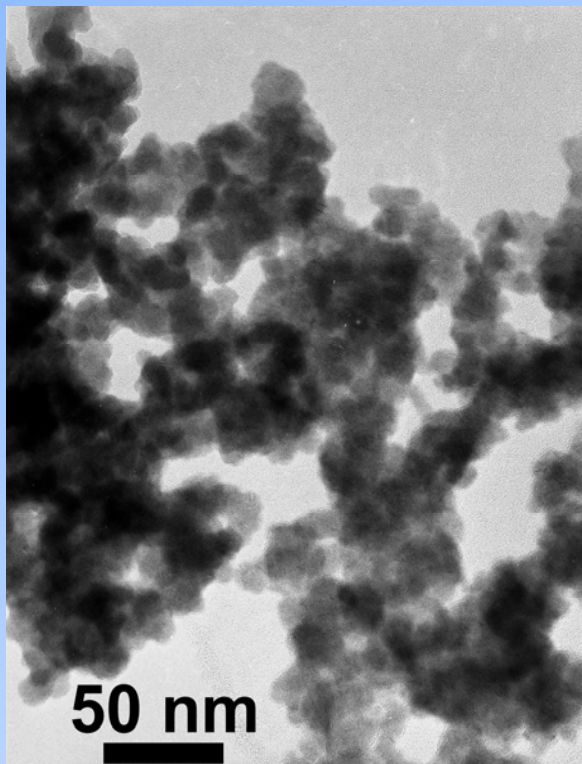
### 4. Filtration: powder collection



# Structural properties

## Transmission Electron Microscopy (TEM) (before filtration)

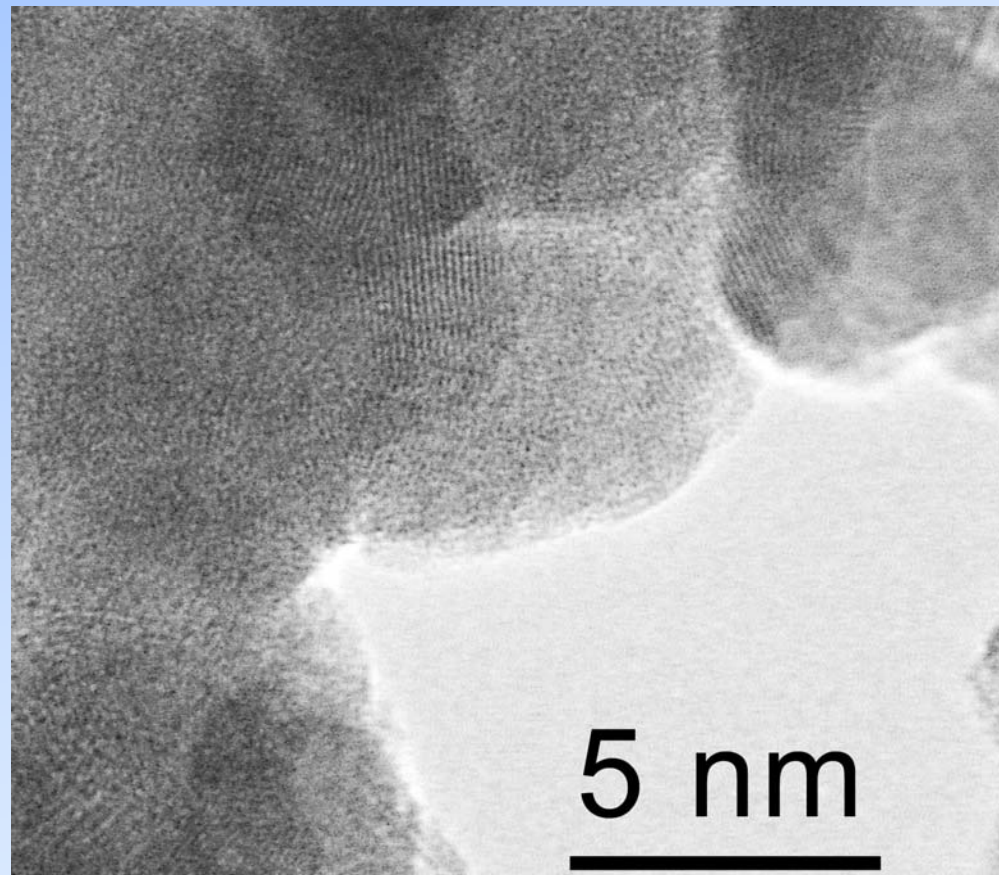
Bright field image



50 nm

Agglomerates > 10 nm

High resolution TEM



5 nm

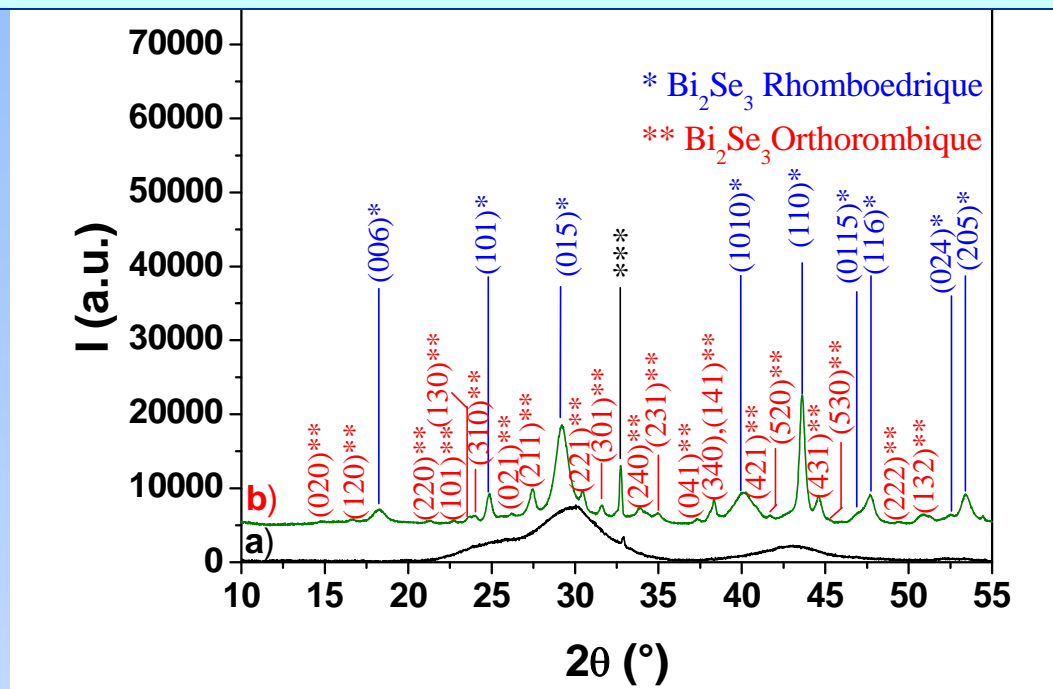
Crystalline domains ~ 5 nm



## X ray Diffraction

### Orthorhombic $\text{Bi}_2\text{Se}_3$ : metastable

Under high pressure *Atabaeva et al. 1973* or special ultrasonic treatment *Qiu et al. JACS 2004*



As synthesized powders: Extremely broad diffraction peaks,

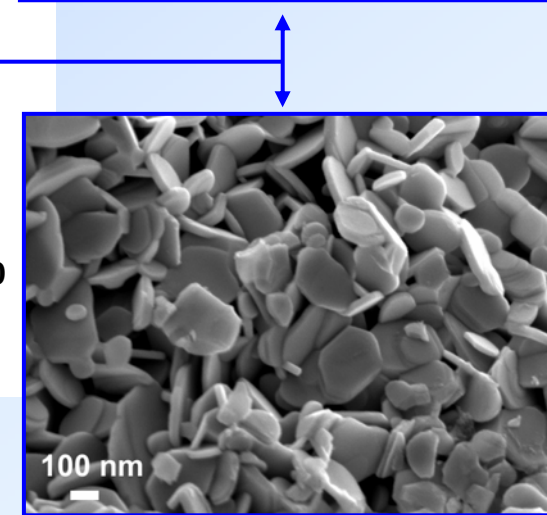
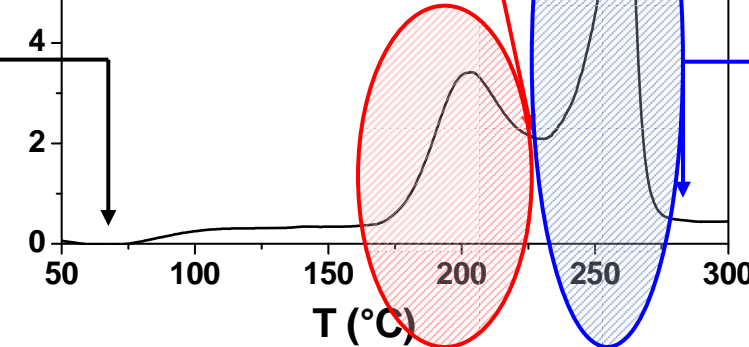
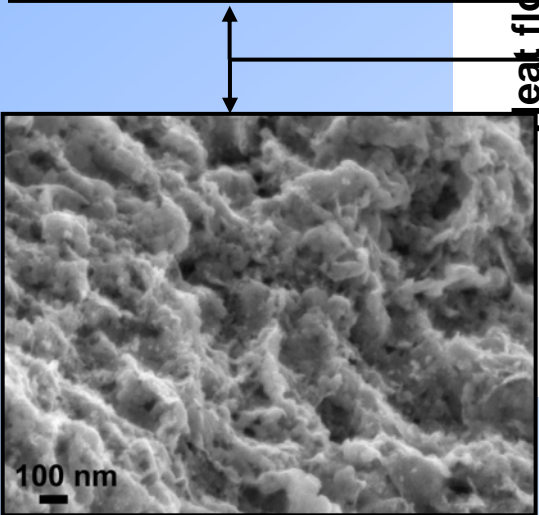
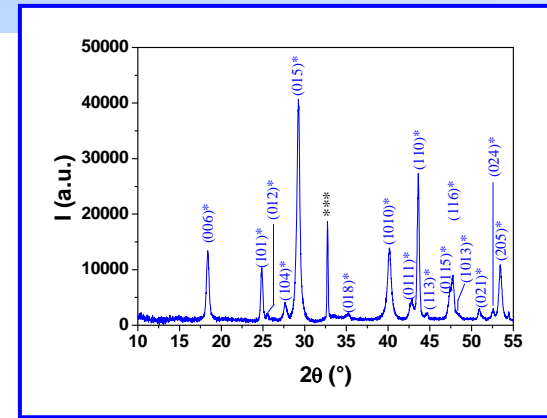
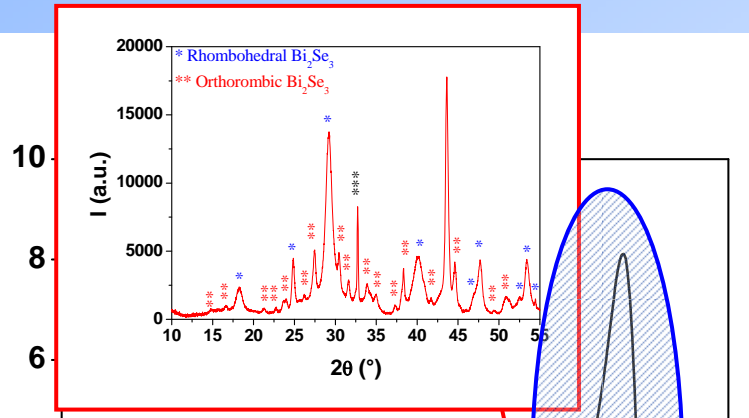
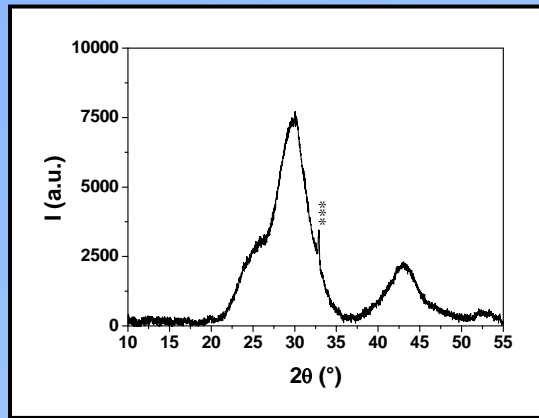
Ambiguous interpretation (small crystal size, several phases, ...?)

Annealing at low temperature ( $200^\circ\text{C}$ ): Two  $\text{Bi}_2\text{Se}_3$  phases: rhombohedral and orthorhombic  
No spurious phases

# Structural properties

Stability range of orthorhombic  $\text{Bi}_2\text{Se}_3$  in NC powders?

Differential Scanning Calorimetry (DSC)



Orthorhombic/rhombohedral  
 chemical transition, ...?

Orthorhombic  $\text{Bi}_2\text{Se}_3$  stable up to  $T > 200^\circ\text{C}$



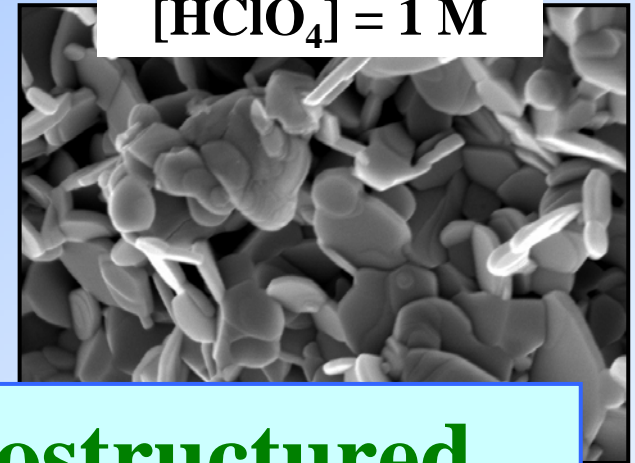
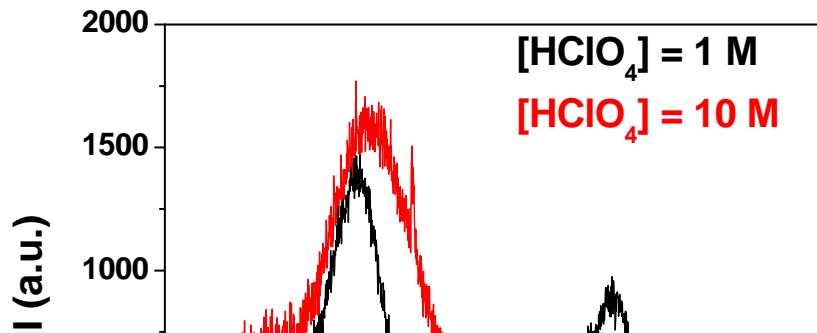
# Structural properties

## [HClO<sub>4</sub>] effects

As synthesized

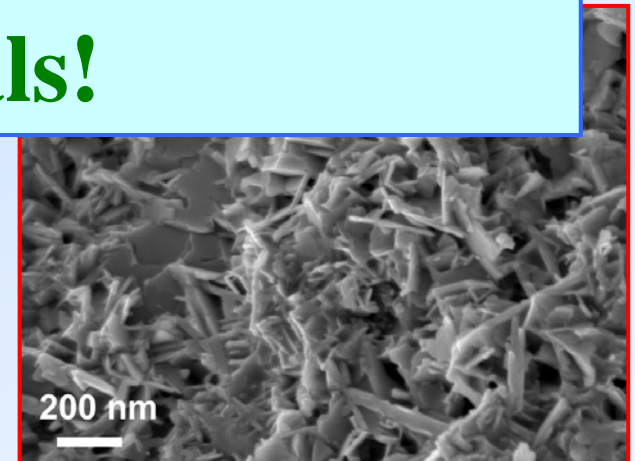
Annealed at 400°C

[HClO<sub>4</sub>] = 1 M



**Wide possibilities for nanostructured bulk TE materials!**

2θ (°)



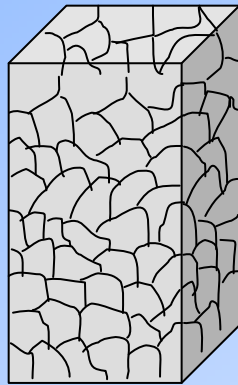
**[H<sup>+</sup>] seems to be a major parameter:**

[HClO<sub>4</sub>] ↗ {  
- crystal size ↘  
- Orth-Bi<sub>2</sub>Se<sub>3</sub> / Rhomb-Bi<sub>2</sub>Se<sub>3</sub> ↗

# “Nanostructured bulk” $\text{Bi}_2\text{Se}_3$

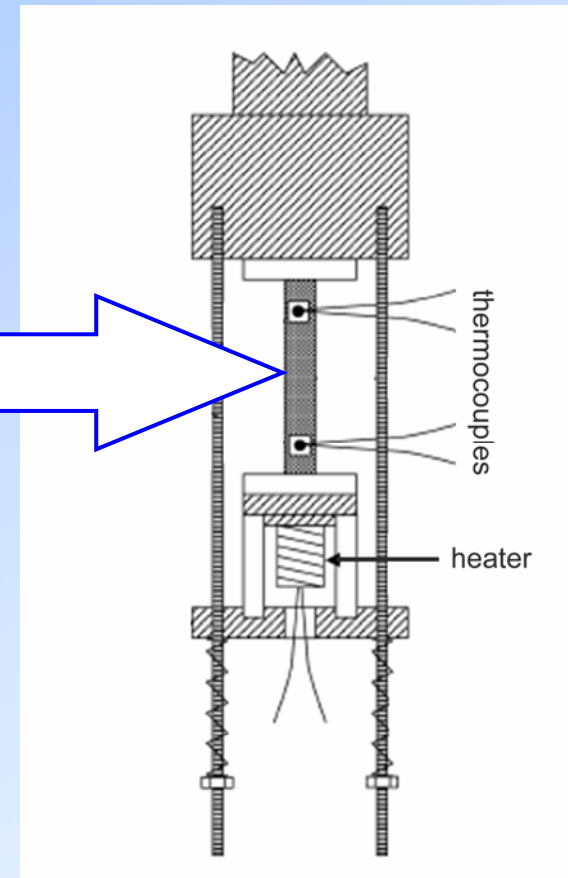
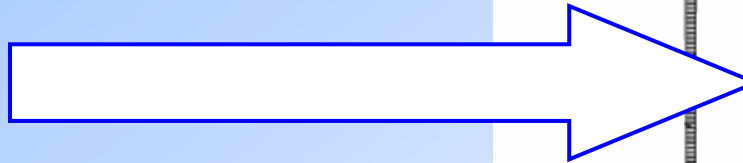
Densification of  $\text{Bi}_2\text{Se}_3$  NC powders  
(pressing at 300 °C under ~ 400 MPa)

Thermopower measurements



10mm×2mm×1mm  
nanostructured bulk samples

Rhombohedral  $\text{Bi}_2\text{Se}_3$



**Preliminary results:  $S = - 125 \mu\text{V}/\text{K}$  at room temperature**

## Conclusions and **future work**

- Innovative route (**low cost and low temperature**) to obtain considerable amounts ( $> 200$  mg/synthesis) of powders constituted of  $\text{Bi}_2\text{Se}_3$  nanocrystals ( $\sim 5$  nm)
- Synthesis of **orthorhombic  $\text{Bi}_2\text{Se}_3$  nanocrystals** (stable up to  $T > 200$  °C, higher than working temperature of  $\text{Bi}_2\text{Te}_3$  based TE devices ).
- “Nanostructured bulk”  $\text{Bi}_2\text{Se}_3$  with Seebeck coefficient  $\sim -125$   $\mu\text{V}/\text{K}$ .
- **Optimization of the densification process (Spark Plasma Sintering)**
- **Study of thermoelectric properties of orthorhombic  $\text{Bi}_2\text{Se}_3$**
- **“Nanostructured bulk” with rhombohedral/orthorhombic interfaces to reduce the thermal conductivity**

**Merci pour votre attention**

