



# Modélisation et simulation in silico

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#### **IMPROVING THE FIGURE OF MERIT**





G. J. Snyder et al, *Adv. Mater.* **2012**, 24, 6125

Thermoelectric Nanomaterials Materials Design and Applications, Eds. T. Mori & K. Koumoto, Springer-Verlag Berlin Heidelberg 2013

#### **BAND ENGINEERING**

$$S = \frac{8\pi^2 k_b^2 T}{3qh^2} m_d^* \left(\frac{\pi}{3n}\right)^{2/3}$$
$$m_d^* = N_v^{2/3} m_b^*$$



#### **DECREASING LATTICE THERMAL CONDUCTIVITY**

#### **PHONON GLASS ELECTRON CRYSTAL**

Search for compounds with:

- Complex crystal structure
- Large unit cell
- Heavy atoms
- Low electronegativity differences between the

constituent atoms

#### NANOSTRUCTURING

M. S. Dresselhaus et al, *Adv. Mater.* 2007, 19, 1043
M. G. Kanatzidis et al, *Energy Environ. Sci.* 2014, 7, 251



G. Slack *CRC Handbook of Thermoelectrics* (Ed D. M. Rowe) CRC, Boca Raton, U.S. pp 407-440 (1995)

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#### **Theory Can Help but...**



- Why DFT?
- Key parameters?
- What are the limitations?
- What about phonons?
- Computing transport properties?



#### "Anyone can do calculations nowadays. Anyone can also operate a scalpel. That doesn't mean all our medical problems are solved."

Karl Irikura

### **Conducting a Computational Project**



These questions should be answered:

- What do you want to know?
- How accurate does the prediction need to be?
- How much time can be devoted to the problem?
- What approximations are being made?

The answers to these questions will determine the type of calculation, model and basis set to be used.

#### **Computational models**



- A model is a system of equations, or computations used to determine the energetics of a molecule
- Different models use different approximations (or levels of theory) to produce results of varying levels of accuracy.
- There is a trade off between accuracy and computational time.
- There are two main types of models; those that use Schrödinger's equation (or simplifications of it) and those that do not.



Approximations are required.

# Born-Oppenheimer



 $\frac{m_n}{m_e} >> 1$   $\Rightarrow Nuclei are much$ slower than electrons



electronic/nuclear decoupling



**The electronic Schrödinger's Equation** 





Approximations are still required. Different approaches:

- Quantum Chemistry (Hartree-Fock, Cl...)
- Quantum Monte Carlo
- Perturbation theory (propagators)
- Density Functional Theory (DFT)

 $\{\vec{R}_{\mu}\} \stackrel{\Psi_{n,\{\vec{R}_{\mu}\}}}{\longrightarrow} \{\vec{R}_{\mu}\} \stackrel{(\{\vec{r}_{i}\})}{\longrightarrow} = E_{sol} \stackrel{el}{\longrightarrow} \stackrel{(\{\vec{r}_{i}\})}{\longrightarrow} \stackrel{(\{$ 

#### **Motivations for DFT**



- The wave function itself is essentially uninterpretable.
- Reduce problem size
- Wave function based methods quickly become intractable for large systems, even with continued improvement in computing power, due to the coupled motion of the electrons.
- A desire to work with some physical observable rather than probability amplitude.

## Why is DFT so popular?



- Transferability. We can use the same codes/methods for very different materials
- Simplicity. The Kohn-Sham equations are conceptually very similar to the Schrödinger equation for a single electron in an external potential
- Reliability. Often we can predict materials properties with high accuracy, sometimes even before experiments
- Software sharing. The development of DFT has become a global enterprise, e.g. open source and collaborative software development
- Robust platform. Often the shortcomings of DFT can be cured by using more sophisticated approaches, which still use DFT as their starting point

# The basics of Density Functional Theory

All properties of the many-body system are determined by the ground state density  $n_0(r)$ 



W. Kohn, chemistry Nobel prize 1998

Each property is a functional of the ground state density  $n_0(r)$  which is written as  $f[n_0]$ 

A functional f  $[n_0]$  maps a function to a result:  $n_0(r) \rightarrow f$ 

#### **The Hohenberg-Kohn Theorems**



• Theorem I: For any system of electrons in an external potential  $V_{ext}(\mathbf{r})$ , that potential is determined uniquely, except for a constant, by the ground state density  $n(\mathbf{r})$ .

Corollary I: Since the hamiltonian is thus fully determined, except for a constant shift of the energy, the full many-body wavefunction and all other properties of the system are also completely determined!

#### **The Hohenberg-Kohn Theorems**



• Theorem II: A <u>universal functional</u> for the energy E[n] of the density  $n(\mathbf{r})$  can be defined for all electron systems. The exact ground state energy is the global minimum for a given  $V_{ext}(\mathbf{r})$ , and the density  $n(\mathbf{r})$  which minimizes this functional is the exact ground state density.

Corollary II: The functional E[n] alone is sufficient to determine the exact ground state energy and density. Excited states of the electrons must be determined by other means.

Minimizing E[n] for a given  $V_{ext}(r) \rightarrow n_0(r)$  and E variational principle

In principle, one can find all other properties and they are functionals of  $n_0(r)$ .

#### **The Kohn-Sham Ansatz**



- Kohn-Sham (1965) Replace original many-body problem with an independent electron problem that can be solved!
- The ground state density is required to be the same as the exact density  $n_0(\mathbf{r}) = \sum_{\sigma} \sum_{i=1}^{\sigma} |\psi_i^{\sigma}(\mathbf{r})|^2$ ,



• The new paradigm – find useful, approximate functionals

#### The Kohn-Sham orbitals and eigenvalues

Non-interacting electrons in the Kohn-Sham potential :

$$\begin{pmatrix} -\frac{1}{2}\nabla^{2} + V_{KS}(\mathbf{r}) \end{pmatrix} \psi_{i}(\mathbf{r}) = \varepsilon_{i}\psi_{i}(\mathbf{r})$$
Density  $n(\mathbf{r}) = \sum \psi_{i}^{*}(\mathbf{r})\psi_{i}(\mathbf{r})$ 

$$V_{KS}(\mathbf{r}) = V_{ext}(\mathbf{r}) + \int \frac{n(\mathbf{r}_{1})}{|\mathbf{r}_{1} - \mathbf{r}|} d\mathbf{r}_{1} + \underbrace{\frac{\delta E_{xc}[n]}{\delta n(\mathbf{r})}}_{Hartree}$$
Exchange-correlation potential
Hartree
self-consistently ! potential

To be solved s J

⇒ Using the variational principle for non-interacting electrons, one can show that the solution of the Kohn-Sham self-consistent system of equations is equivalent to the minimisation of

$$E_{\text{KS}}\left[\left\{\psi_{i}\right\}\right] = \sum_{i} \left\langle\psi_{i}\right| - \frac{1}{2}\nabla^{2}\left|\psi_{i}\right\rangle + \int V_{ext}(\mathbf{r})n(\mathbf{r})d\mathbf{r} + \frac{1}{2}\int\frac{n(\mathbf{r}_{1})n(\mathbf{r}_{2})}{\left|\mathbf{r}_{1}-\mathbf{r}_{2}\right|}d\mathbf{r}_{1}d\mathbf{r}_{2} + E_{\text{xc}}\left[n\right]$$



spin

Spin polarized

non spin polarized

analytic functions (e.g. Hankel) atomic orbitals. e.g. Slater (STO), Gaussians (GTO) numerical

#### **Overview of DFT concepts**



**XC** functionals





B2GP-PLYP

**B3LYP, PBE0, HSE** 

TPSS, tHCTH, Mo6L, SCAN

BP86, PBE, PW91, BLYP...

VWN

The case of Fe



Ecole Nationale Supérieure de Chimie de Rennes

#### **Influence of XC on band gaps**





Mg<sub>2</sub>Si

→ LDA & GGA functionals underestimate band gaps

J. Mater. Chem. C, 2015, 3, 530-537



 $\rightarrow$  mBJ potential

Phys. Rev. B 99, 035139 (2019)

#### **The Becke-Johnson potential**

THE JOURNAL OF CHEMICAL PHYSICS 124, 221101 (2006)

#### A simple effective potential for exchange

Axel D. Becke<sup>a)</sup> and Erin R. Johnson Department of Chemistry, Queen's University, Kingston, Ontario K7L 3N6, Canada

(Received 29 March 2006; accepted 22 May 2006; published online 9 June 2006)

The optimized effective potential (OEP) for exchange was introduced some time ago by Sharp and Horton [Phys. Rev. **90**, 317 (1953)] and by Talman and Shadwick [Phys. Rev. A **14**, 36 (1976)]. The integral equation for the OEP is difficult to solve, however, and a variety of approximations have therefore been proposed. These are explicitly orbital dependent and require the same two-electron integrals as Hartree-Fock theory. We have found a remarkably simple approximate effective potential that closely resembles the Talman-Shadwick potential in atoms. It depends only on total densities and requires no two-electron integrals. © 2006 American Institute of Physics. [DOI: 10.1063/1.2213970]



#### **The Becke-Johnson potential**

week ending

5 JUNE 2009



#### PRL 102, 226401 (2009)

PHYSICAL REVIEW LETTERS

#### Accurate Band Gaps of Semiconductors and Insulators with a Semilocal Exchange-Correlation Potential

#### Fabien Tran and Peter Blaha

Institute of Materials Chemistry, Vienna University of Technology, Getreidemarkt 9/165-TC, A-1060 Vienna, Austria (Received 5 December 2008; published 3 June 2009)

A modified version of the exchange potential proposed by Becke and Johnson [J. Chem. Phys. **124**, 221101 (2006)] is tested on solids for the calculation of band gaps. The agreement with experiment is very good for all types of solids we considered (e.g., wide band gap insulators, sp semiconductors, and strongly correlated 3*d* transition-metal oxides) and is of the same order as the agreement obtained with the hybrid functionals or the *GW* methods. This semilocal exchange potential, which recovers the local-density approximation (LDA) for a constant electron density, mimics very well the behavior of orbital-dependent potentials and leads to calculations which are barely more expensive than LDA calculations. Therefore, it can be applied to very large systems in an efficient way.



# $\begin{array}{c} 55\\50\\45\\\end{array}$





npj Computational Materials (2020) 96

#### **The Becke-Johnson potential**

#### **Relativistic effects**





#### **Relativistic effects**

(a)

F

**K-POINTS** 

Г

3

2

Energy (eV)

-1

-2

-3⊾ Г

Ζ

GGA







Can be important for "heavy" atoms

AIP Advances **10**, 095018 (2020)

#### **Relativistic effects**





Ag<sub>4</sub>Mo<sub>9</sub>Se<sub>11</sub>



#### Vibrational eigenmodes and eigenvalues





Taylor expansion of potential energy surface for small displacements

$$U = U_0 + \sum_{I\alpha} u_{I\alpha} \frac{\partial U}{\partial R_{I\alpha}} + \frac{1}{2} \sum_{I\alpha, J\beta} \frac{\partial^2 U}{\partial R_{I\alpha} \partial R_{J\beta}} u_{I\alpha} u_{J\beta} + \mathcal{O}(u^3)$$

$$K_{I\alpha, J\beta}$$
matrix of force constants

#### **Calculations of vibrational frequencies**

#### 1) Frozen-phonon method



$$\frac{\partial^2 U}{\partial R_{I\alpha}^2} \simeq \frac{U(R_{I\alpha}^0 + u) - 2U_0 + U(R_{I\alpha}^0 - u)}{u^2}$$

or alternatively using the forces

$$rac{\partial^2 U}{\partial R_{Ilpha}^2} \simeq -rac{F_{Ilpha}(R_{Ilpha}^0+u) - F_{Ilpha}(R_{Ilpha}^0-u)}{2u}$$

2) Density-functional perturbation theory (DFPT) The method of choice for crystals see Baroni et al, Rev. Mod. Phys. 73, 515 (2001)

#### **Phonons dispersions relations**





#### DFT/LDA

• Inelastic neutron scattering

![](_page_34_Figure_0.jpeg)

#### **Negative frequencies**

![](_page_34_Picture_2.jpeg)

#### **Negative frequencies**

![](_page_35_Picture_1.jpeg)

![](_page_35_Picture_2.jpeg)

![](_page_35_Figure_3.jpeg)

![](_page_35_Figure_4.jpeg)

![](_page_36_Figure_0.jpeg)

#### **How Theory Can Help?**

![](_page_36_Picture_2.jpeg)

#### **Structural models**

![](_page_37_Picture_1.jpeg)

#### Ag<sub>4</sub>Mo<sub>9</sub>Se<sub>11</sub>

![](_page_37_Figure_3.jpeg)

#### **Structural models**

![](_page_38_Picture_1.jpeg)

![](_page_38_Picture_2.jpeg)

![](_page_38_Picture_3.jpeg)

#### **Structural models**

![](_page_39_Picture_1.jpeg)

![](_page_39_Picture_2.jpeg)

![](_page_39_Picture_3.jpeg)

Ag<sub>4</sub>Mo<sub>9</sub>Se<sub>11</sub>

#### **Drude theory of metals**

![](_page_40_Picture_1.jpeg)

![](_page_40_Figure_2.jpeg)

➡ Valence electrons move freely: electron gas model

![](_page_41_Picture_0.jpeg)

Trajectory of conduction electron diffused by ions according to Drude

- porteurs de charge positive immobiles (// Born-Oppenheimer)
- approximation des électrons indépendants
- approximation des électrons libres
- les collisions sont des évènements **instantanés** ⇔ négligence des forces à longue portée
- probabilité qu'un électron subisse une collision dans l'intervalle de temps dt est égal à dt/τ avec τ appelé temps de collision, temps de vol moyen ou temps de relaxation.
- équilibre thermique des électrons atteint uniquement par le biais des collisions

#### **Drude theory of metals**

![](_page_42_Picture_1.jpeg)

#### Thermal conductivity of metals

#### ➡ Wiedemann-Franz law

 $\frac{\kappa}{\sigma \mathbf{T}} = \mathbf{cste}$ 

Lorenz number

- $\kappa: \ thermal \ conductivity$
- $\sigma$ : electrical conductivity
- T: temperature

Élément	273 K		373 K	
	$(\mathrm{W.cm}^{-1}.\mathrm{K}^{-1})$	$\kappa/\sigma T$ (W. $\Omega$ .K <sup>-2</sup> )	$\kappa$ (W.cm <sup>-1</sup> .K <sup>-1</sup> )	$\kappa/\sigma T$ (W. $\Omega$ .K <sup>-2</sup> )
Li	0,71	$2,22 \times 10^{-8}$	0,73	$2,43 \times 10^{-8}$
Na	1,38	2,12		
Κ	1,0	2,23		
$\mathbf{Rb}$	0,6	2,42		
Cu	3,85	2,20	3,82	2,29
$\mathbf{A}\mathbf{g}$	4,18	2,31	4,17	2,38
$\mathbf{A}\mathbf{u}$	$^{3,1}$	2,32	3,1	2,36
$\mathbf{Be}$	$^{2,3}$	2,36	1,7	$2,\!42$
Mg	1,5	$2,\!14$	1,5	2,25
Nb	0,52	2,90	0,54	2,78
Fe	0,80	2,61	0,73	2,88
Zn	1,13	2,28	1,1	2,30
$\mathbf{Cd}$	1,0	2,49	1,0	
$\mathbf{Al}$	2,38	2,14	2,30	2,19
In	0,88	2,58	0,80	$2,\!60$
Tl	0,5	2,75	$0,\!45$	2,75
$\mathbf{Sn}$	$0,\!64$	2,48	0,60	2,54
$\mathbf{Pb}$	0,38	2,64	0,35	2,53
Bi	0,09	3,53	0,08	3,35
$\mathbf{Sb}$	0,18	2,57	0,17	$2,\!69$

Source : G. W. C. Kaye et T. H. Laby, *Table of Physical and Chemical Constants*, Longmans Green, London, 1966.

### Modeling Transport Properties: The Semi-Classical Approach

![](_page_43_Picture_1.jpeg)

![](_page_43_Figure_2.jpeg)

#### Wave packet velocity

![](_page_44_Picture_1.jpeg)

![](_page_44_Figure_2.jpeg)

![](_page_44_Figure_3.jpeg)

### Modeling Transport Properties: The Semi-Classical Approach

![](_page_45_Picture_1.jpeg)

![](_page_45_Figure_2.jpeg)

#### **Electrical transport properties**

![](_page_46_Picture_1.jpeg)

![](_page_46_Figure_2.jpeg)

Inorg. Chem. 56, 4229 (2017)

![](_page_47_Figure_0.jpeg)

#### Main codes

![](_page_48_Picture_1.jpeg)

https://en.wikipedia.org/wiki/List of quantum chemistry and solid-state physics software